

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 1 Chemical Reactions and Equations Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 1 Chemical Reactions and Equations Download 2017 2018 New Edition PDF>

Class 10 Science NCERT SOLUTIONS For Chapter 1 Chemical Reactions and Equations Download 2017 2018 New Edition PDF

Question 1:

Why should a magnesium ribbon be cleaned before burning in air?

Answer

Magnesium is very reactive metal. When stored it reacts with oxygen to form a layer magnesium oxide on its surface. This layer of magnesium oxide is quite stable and prevents further reaction of magnesium with oxygen. The magnesium ribbon is cleaned by sand paper to remove this layer so that the underlying metal can be exposed into air.

Question 2:

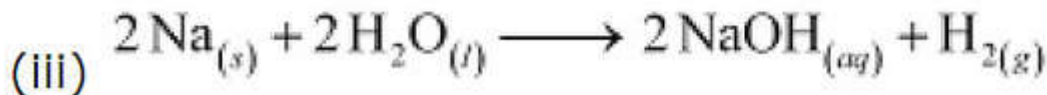
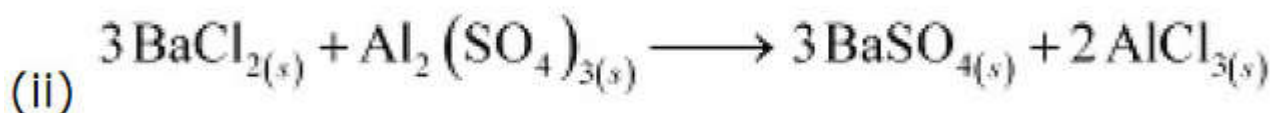
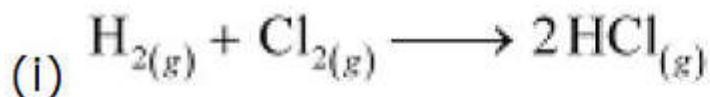
Write the balanced equation for the following chemical reactions.

(i) Hydrogen + Chlorine \rightarrow Hydrogen chloride

(ii) Barium chloride + Aluminium sulphate \rightarrow Barium sulphate + Aluminium chloride

(iii) Sodium + Water \rightarrow Sodium hydroxide + Hydrogen

Answer



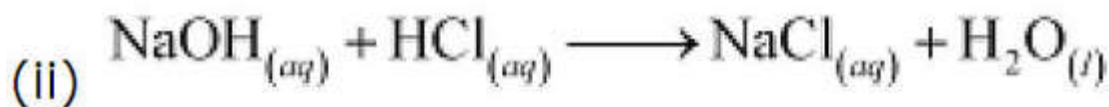
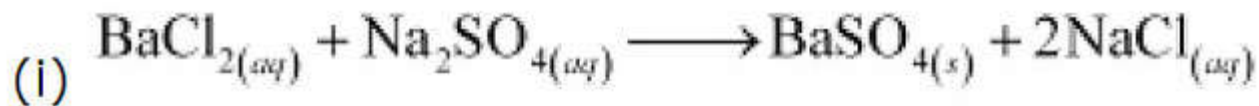
Question 3:

Write a balanced chemical equation with state symbols for the following reactions.

(i) Solutions of barium chloride and sodium sulphate in water react to give insoluble barium sulphate and the solution of sodium chloride.

(ii) Sodium hydroxide solution (in water) reacts with hydrochloric acid solution (in water) to produce sodium chloride solution and water.

Answer



Question 1:

A solution of a substance 'X' is used for white washing.

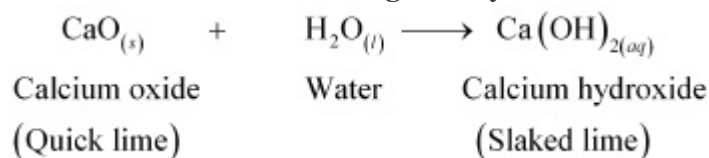
(i) Name the substance 'X' and write its formula.

(ii) Write the reaction of the substance 'X' named in (i) above with water.

Answer

(i) The substance 'X' is calcium oxide. Its chemical formula is CaO.

(ii) Calcium oxide reacts vigorously with water to form calcium hydroxide (slaked lime).



Question 2:

Why is the amount of gas collected in one of the test tubes in Activity 1.7 double of the amount collected in the other? Name this gas.

Answer

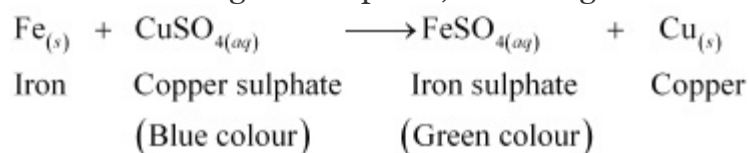
Water (H₂O) contains two parts hydrogen and one part oxygen. Therefore, the amount of hydrogen and oxygen produced during electrolysis of water is in a 2:1 ratio. During electrolysis, since hydrogen goes to one test tube and oxygen goes to another, the amount of gas collected in one of the test tubes is double of the amount collected in the other.

Question 1:

Why does the colour of copper sulphate solution change when an iron nail is dipped in it?

Answer

When an iron nail is placed in a copper sulphate solution, iron displaces copper from copper sulphate solution forming iron sulphate, which is green in colour.



Therefore, the blue colour of copper sulphate solution fades and green colour appears.

Question 2

Give an example of a double displacement reaction other than the one given in Activity

Answer

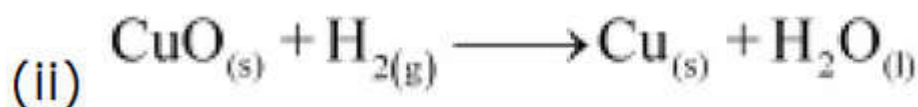
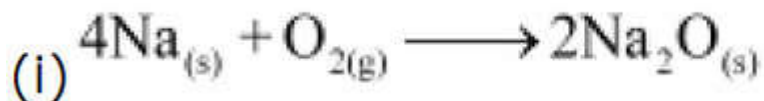
Sodium carbonate reacts with calcium chloride to form calcium carbonate and sodium chloride.



In this reaction, sodium carbonate and calcium chloride exchange ions to form two new compounds. Hence, it is a double displacement reaction.

Question 3:

Identify the substances that are oxidised and the substances that are reduced in the following reactions.



Answer

(i) Sodium (Na) is oxidised as it gains oxygen and oxygen gets reduced.

(ii) Copper oxide (CuO) is reduced to copper (Cu) while hydrogen (H₂) gets oxidised to water (H₂O).

Exercise solution

Question 1:

Which of the statements about the reaction below are incorrect?



(a) Lead is getting reduced.

(b) Carbon dioxide is getting oxidised.

(c) Carbon is getting oxidised.

(d) Lead oxide is getting reduced.

(i) (a) and (b)

(ii) (a) and (c)

(iii) (a), (b) and (c)

(iv) all

Answer

(i)(a) and (b)

Question 2:



The above reaction is an example of a

(a) combination reaction.

(b) double displacement reaction.

(c) decomposition reaction.

(d) displacement reaction.

Answer

(d) The given reaction is an example of a displacement reaction.

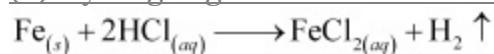
Question 3:

What happens when dilute hydrochloric acid is added to iron filings? Tick the correct answer.

- (a) Hydrogen gas and iron chloride are produced.
- (b) Chlorine gas and iron hydroxide are produced.
- (c) No reaction takes place.
- (d) Iron salt and water are produced.

Answer

(a) Hydrogen gas and iron chloride are produced. The reaction is as follows:

**Question 4:**

What is a balanced chemical equation? Why should chemical equations be balanced?

Answer

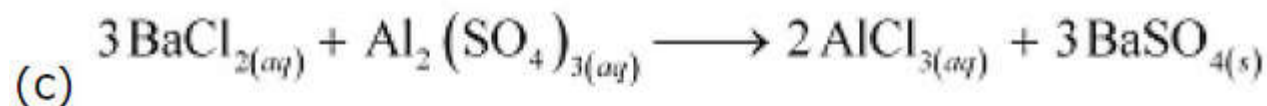
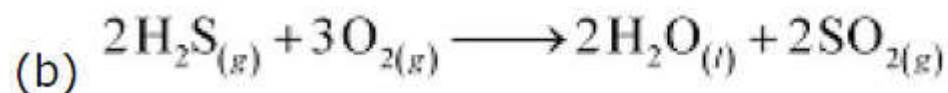
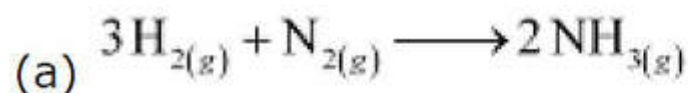
A reaction which has an equal number of atoms of all the elements on both sides of the chemical equation is called a balanced chemical equation.

The law of conservation of mass states that mass can neither be created nor destroyed. Hence, in a chemical reaction, the total mass of reactants should be equal to the total mass of the products. It means that the total number of atoms of each element should be equal on both sides of a chemical equation. Hence, it is for this reason that chemical equations should be balanced.

Question 5:

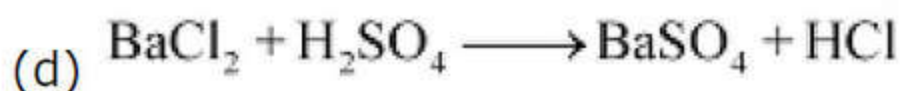
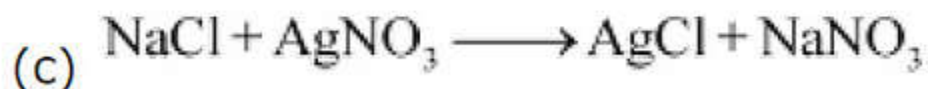
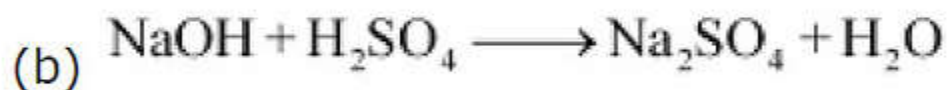
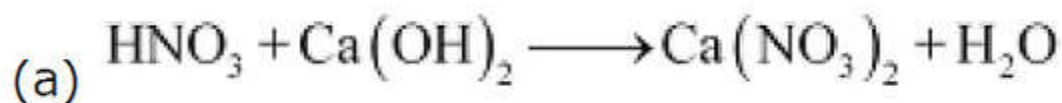
Translate the following statements into chemical equations and then balance them.

- (a) Hydrogen gas combines with nitrogen to form ammonia.
- (b) Hydrogen sulphide gas burns in air to give water and sulphur dioxide.
- (c) Barium chloride reacts with aluminium sulphate to give aluminium chloride and a precipitate of barium sulphate.
- (d) Potassium metal reacts with water to give potassium hydroxide and hydrogen gas.

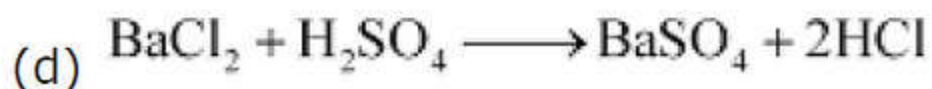
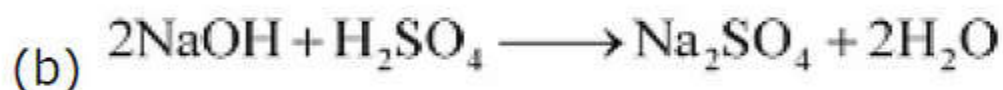
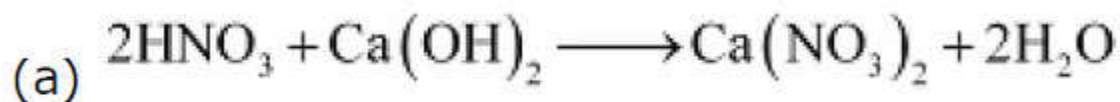
Answer

Question 6:

Balance the following chemical equations.



Answer



Question 7:

Write the balanced chemical equations for the following reactions.

(a) Calcium hydroxide + Carbon dioxide \rightarrow Calcium carbonate + Water

(b) Zinc + Silver nitrate \rightarrow Zinc nitrate + Silver

(c) Aluminium + Copper chloride \rightarrow Aluminium chloride + Copper

(d) Barium chloride + Potassium sulphate \rightarrow Barium sulphate + Potassium chloride

very

if you.....

Answer

- (a) $\text{Ca(OH)}_2 + \text{CO}_2 \longrightarrow \text{CaCO}_3 + \text{H}_2\text{O}$
- (b) $\text{Zn} + 2 \text{AgNO}_3 \longrightarrow \text{Zn(NO}_3)_2 + 2 \text{Ag}$
- (c) $2 \text{Al} + 3 \text{CuCl}_2 \longrightarrow 2 \text{AlCl}_3 + 3 \text{Cu}$
- (d) $\text{BaCl}_2 + \text{K}_2\text{SO}_4 \longrightarrow \text{BaSO}_4 + 2 \text{KCl}$

Question 8:

Write the balanced chemical equation for the following and identify the type of reaction in each case.

- (a) Potassium bromide (aq) + Barium iodide (aq) \rightarrow Potassium iodide (aq) + Barium bromide(s)
- (b) Zinc carbonate (s) \rightarrow Zinc oxide (s) + Carbon dioxide (g)
- (c) Hydrogen (g) + Chlorine (g) \rightarrow Hydrogen chloride (g)
- (d) Magnesium (s) + Hydrochloric acid (aq) \rightarrow Magnesium chloride (aq) + Hydrogen (g)

Answer

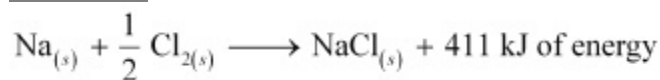
- (a) $2\text{KBr}_{(aq)} + \text{BaI}_{2(aq)} \longrightarrow 2\text{KI}_{(aq)} + \text{BaBr}_{2(s)}$; Double displacement reaction
- (b) $\text{ZnCO}_{3(s)} \longrightarrow \text{ZnO}_{(s)} + \text{CO}_{2(g)}$; Decomposition reaction
- (c) $\text{H}_{2(g)} + \text{Cl}_{2(g)} \longrightarrow 2\text{HCl}_{(g)}$; Combination reaction
- (d) $\text{Mg}_{(s)} + 2\text{HCl}_{(aq)} \longrightarrow \text{MgCl}_{2(aq)} + \text{H}_{2(g)}$; Displacement reaction

Question 9:

What does one mean by exothermic and endothermic reactions? Give examples.

Answer

Chemical reactions that release energy in the form of heat, light, or sound are called exothermic reactions.



Example: Mixture of sodium and chlorine to yield table salt. In other words, combination reactions are exothermic.

Reactions that absorb energy or require energy in order to proceed are called endothermic reactions.



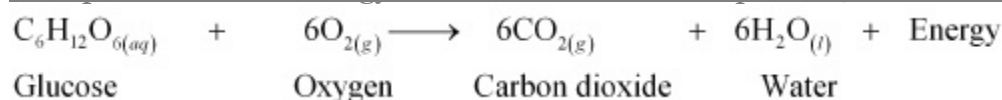
For example: In the process of photosynthesis, plants use the energy from the sun to convert carbon dioxide and water to glucose and oxygen.

Question 10:

Why is respiration considered an exothermic reaction? Explain.

Answer

Energy is required to support life. Energy in our body is obtained from the food we eat. During digestion, large molecules of food are broken down into simpler substances such as glucose. Glucose combines with oxygen in the cells and provides energy. The special name of this combustion reaction is respiration. Since energy is released in the whole process, it is an exothermic process.



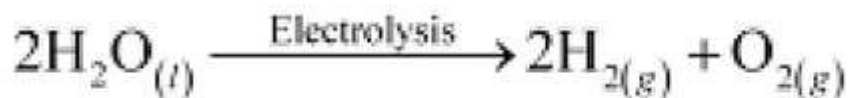
Question 11:

Why are decomposition reactions called the opposite of combination reactions? Write equations for these reactions.

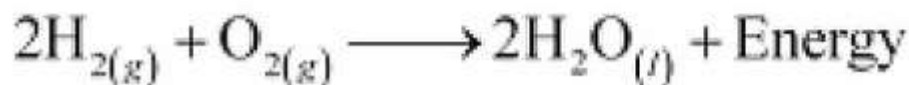
Answer

Decomposition reactions are those in which a compound breaks down to form two or more substances. These reactions require a source of energy to proceed. Thus, they are the exact opposite of combination reactions in which two or more substances combine to give a new substance with the release of energy.

Decomposition reaction: $\text{AB} + \text{Energy} \longrightarrow \text{A} + \text{B}$



Combination reaction: $\text{A} + \text{B} \longrightarrow \text{AB} + \text{Energy}$

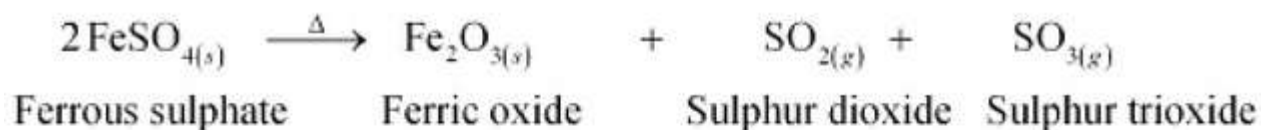


Question 12:

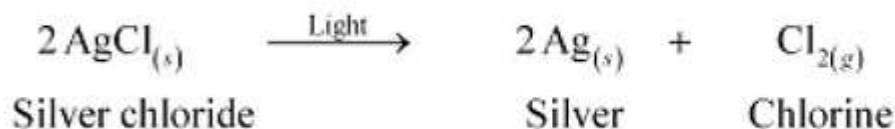
Write one equation each for decomposition reactions where energy is supplied in the form of heat, light or electricity.

Answer

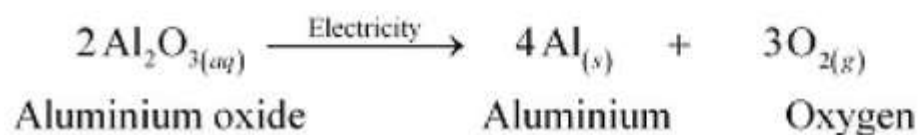
(a) Thermal decomposition:



(b) Decomposition by light:



(c) Decomposition by electricity:



Question 13:

What is the difference between displacement and double displacement reactions? Write equations for these reactions.

Answer

In a displacement reaction, a more reactive element replaces a less reactive element from a compound.

$A + BX \rightarrow AX + B$; where A is more reactive than B

In a double displacement reaction, two atoms or a group of atoms switch places to form new compounds.

$AB + CD \rightarrow AD + CB$

For example:

Displacement reaction:

$\text{CuSO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{Cu}$

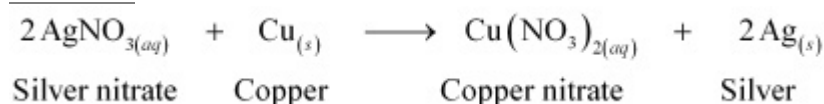
Double displacement reaction:

$\text{Na}_2\text{SO}_4 + \text{BaCl}_2 \rightarrow \text{BaSO}_4 + 2\text{NaCl}$

Question 14:

In the refining of silver, the recovery of silver from silver nitrate solution involved displacement by copper metal. Write down the reaction involved.

Answer



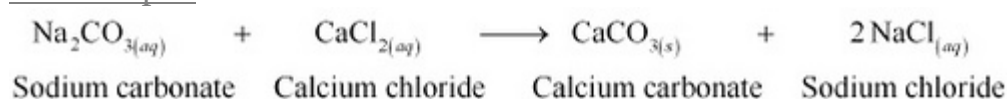
Question 15:

What do you mean by a precipitation reaction? Explain by giving examples.

Answer

A reaction in which an insoluble solid (called precipitate) is formed is called a precipitation reaction.

For example:



Another example of precipitation reaction is:



Question 16:

(a) Oxidation

(b) Reduction

Answer

(a) Oxidation is the gain of oxygen.

For example:



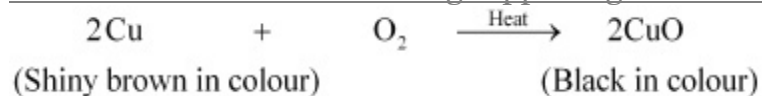
For example:



Question 17:

Answer

'X' is copper (Cu) and the black-coloured compound formed is copper oxide (CuO). The equation of the reaction involved on heating copper is given below.



Question 18:

Why do we apply paint on iron articles?

Answer

Iron articles are painted because it prevents them from rusting. When painted, the contact of iron articles from moisture and air is cut off. Hence, rusting is prevented their presence is essential for rusting to take place.

Question 19:

Oil and fat containing food items are flushed with nitrogen. Why?

Answer

Nitrogen is an inert gas and does not easily react with these substances. On the other hand, oxygen reacts with food substances and makes them rancid. Thus, bags used in packing food items are flushed with nitrogen gas to remove oxygen inside the pack. When oxygen is not present inside the pack, rancidity of oil and fat containing food items is avoided.

Question 20:

Explain the following terms with one example each.

(a) Corrosion

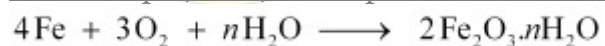
(b) Rancidity

Answer

(a) Corrosion:

Corrosion is defined as a process where materials, usually metals, deteriorate as a result of a chemical reaction with air, moisture, chemicals, etc.

For example, iron, in the presence of moisture, reacts with oxygen to form hydrated iron oxide.



Hydrated iron oxide

This hydrated iron oxide is rust.

(b) Rancidity:

The process of oxidation of fats and oils that can be easily noticed by the change in taste and smell is known as rancidity.

For example, the taste and smell of butter changes when kept for long.

Rancidity can be avoided by:

1. Storing food in air tight containers
2. Storing food in refrigerators
3. Adding antioxidants
4. Storing food in an environment of nitrogen

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 2 Acids Bases and Salts Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 2 Acids Bases and Salts Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 2 Acids Bases and Salts NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

You have been provided with three test tubes. One of them contains distilled water and the other two contain an acidic solution and a basic solution, respectively. If you are given only red litmus paper, how will you identify the contents of each test tube?

Answer

If the colour of red litmus paper gets changed to blue, then it is a base and if there is no colour change, then it is either acidic or neutral. Thus, basic solution can be easily identified.

Let us mark the three test tubes as A, B, and C. A drop of the solution in A is put on the red litmus paper. Same is repeated with solution B and C. If either of them changes colour to blue, then it is basic. Therefore, out of three, one is eliminated. Out of the remaining two, any one can be acidic or neutral. Now a drop of basic solution is mixed with a drop of each of the remaining two solutions separately and then the nature of the drops of the mixtures is checked. If the colour of red litmus turns blue, then the second solution is neutral and if there is no change in colour, then the second solution is acidic. This is because acidic and basic solutions neutralize each other. Hence, we can distinguish between the three types of solutions.

WE ARE WITH YOU.....

Question 1:

Why should curd and sour substances not be kept in brass and copper vessels?

Answer

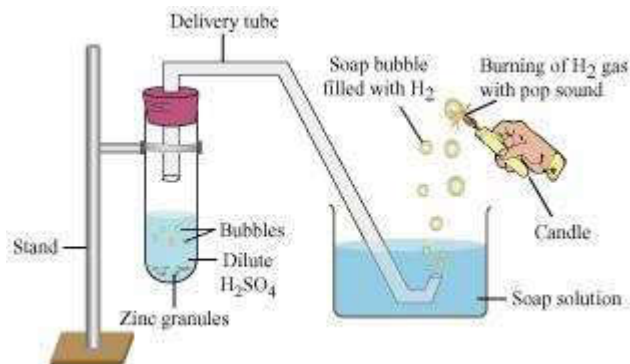
Curd and other sour substances contain acids. Therefore, when they are kept in brass and copper vessels, the metal reacts with the acid to liberate hydrogen gas and harmful products, thereby spoiling the food.

Question 2:

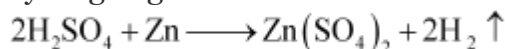
Which gas is usually liberated when an acid reacts with a metal? Illustrate with an example. How will you test for the presence of this gas?

Answer

Hydrogen gas is usually liberated when an acid reacts with a metal.



Take few pieces of zinc granules and add 5 ml of dilute H_2SO_4 . Shake it and pass the gas produced into a soap solution. The bubbles of the soap solution are formed. These soap bubbles contain hydrogen gas.

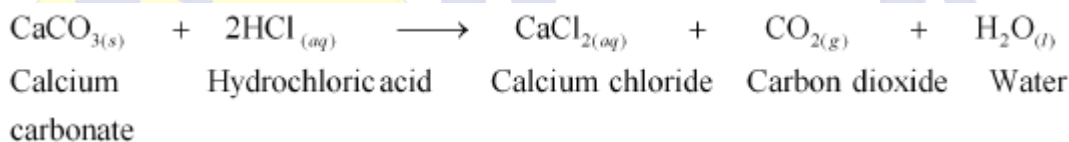


We can test the evolved hydrogen gas by its burning with a pop sound when a candle is brought near the soap bubbles.

Question 3:

Metal compound A reacts with dilute hydrochloric acid to produce effervescence. The gas evolved extinguishes a burning candle. Write a balanced chemical equation for the reaction if one of the compounds formed is calcium chloride.

Answer



Question 1:

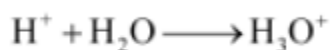
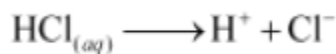
Why do HCl, HNO_3 , etc., show acidic characters in aqueous solutions while solutions of compounds like alcohol and glucose do not show acidic character?

Answer

The dissociation of HCl or HNO_3 to form hydrogen ions always occurs in the presence of water.

Hydrogen ions (H^+) combine with H_2O to form hydronium ions (H_3O^+).

The reaction is as follows:



Question 2:

Why does an aqueous solution of an acid conduct electricity?

Answer

Acids dissociate in aqueous solutions to form ions. These ions are responsible for conduction of electricity.

Question 3:

Why does dry HCl gas not change the colour of the dry litmus paper?

Answer

Colour of the litmus paper is changed by the hydrogen ions. Dry HCl gas does not contain H^+ ions. It is only in the aqueous solution that an acid dissociates to give ions. Since in this case, neither HCl is in the aqueous form nor the litmus paper is wet, therefore, the colour of the litmus paper does not change.

Question 4:

While diluting an acid, why is it recommended that the acid should be added to water and not water to the acid?

Answer

Since the process of dissolving an acid in water is exothermic, it is always recommended that acid should be added to water. If it is done the other way, then it is possible that because of the large amount of heat generated, the mixture splashes out and causes burns.

Question 5:

How is the concentration of hydronium ions (H_3O^+) affected when a solution of an acid is diluted?

Answer

When an acid is diluted, the concentration of hydronium ions (H_3O^+) per unit volume decreases. This means that the strength of the acid decreases.

Question 6:

How is the concentration of hydroxide ions (OH^-) affected when excess base is dissolved in a solution of sodium hydroxide?

Answer

The concentration of hydroxide ions (OH^-) would increase when excess base is dissolved in a solution of sodium hydroxide.

=

Question 1:

You have two solutions, A and B. The pH of solution A is 6 and pH of solution B is 8. Which solution has more hydrogen ion concentration? Which of this is acidic and which one is basic?

Answer

A pH value of less than 7 indicates an acidic solution, while greater than 7 indicates a basic solution. Therefore, the solution with pH = 6 is acidic and has more hydrogen ion concentration than the solution of pH = 8 which is basic.

Question 2:

What effect does the concentration of $H^+(aq)$ ions have on the nature of the solution?

Answer

Concentration of $H^+(aq)$ can have a varied effect on the nature of the solution. With an increase in H^+ ion concentration, the solution becomes more acidic, while a decrease of H^+ ion causes an increase in the basicity of the solution.

Question 3:

Do basic solutions also have $H^+(aq)$ ions. If yes, then why are these basic?

Answer

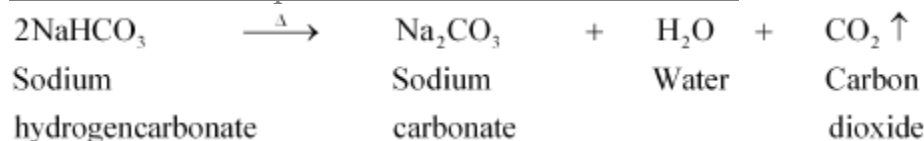
Yes, basic solution also has $H^+(aq)$ ions. However, their concentration is less as compared to the concentration of OH^- ions that makes the solution basic.

Question 4:

Under what soil condition do you think a farmer would treat the soil of his fields with quick lime (calcium oxide) or slaked lime (calcium hydroxide) or chalk (calcium carbonate)?

Answer

If the soil is acidic and improper for cultivation, then to increase the basicity of soil, the farmer would treat the soil with quick lime or slaked lime or chalk



=

Question 1:

What is the common name of the compound CaOCl_2 ?

Answer

The common name of the compound CaOCl_2 is bleaching powder.

Question 2:

Name the substance which on treatment with chlorine yields bleaching powder?

Answer

Calcium hydroxide $[\text{Ca}(\text{OH})_2]$, on treatment with chlorine, yields bleaching powder.

Question 3:

Name the sodium compound which is used for softening hard water.

Answer

Washing soda ($\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$) is used for softening hard water.

Question 4:

What will happen if a solution of sodium hydrocarbonate is heated? Give the equation of the reaction involved.

Answer

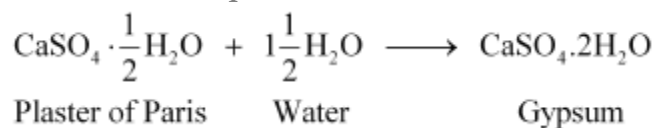
When a solution of sodium hydrocarbonate (sodium hydrogencarbonate) is heated, sodium carbonate and water are formed with the evolution of carbon dioxide gas.

Question 5:

Write an equation to show the reaction between Plaster of Paris and water.

Answer

The chemical equation for the reaction of Plaster of Paris and water can be represented as

**Exercise solution**

Question 1:

A solution turns red litmus blue, its pH is likely to be

- (a) 1 (b) 4 (c) 5 (d) 10

Answer

(d) Bases turn red litmus blue and acids turn blue litmus red. Basic solution has a pH value more than 7. Since the solution turns red litmus blue, its pH is likely to be 10.

Question 2:

A solution reacts with crushed egg-shells to give a gas that turns lime-water milky. The solution contains

- (a) NaCl (b) HCl (c) LiCl (d) KCl

Answer

(b) The solution contains HCl.

Question 3:

10 mL of a solution of NaOH is found to be completely neutralised by 8 mL of a given solution of HCl. If we take 20 mL of the same solution of NaOH, the amount of HCl solution (the same solution as before) required to neutralise it will be

- (a) 4 mL (b) 8mL (c) 12 mL (d) 16 mL

Answer

(d) 16 mL of HCl solution will be required.

Question 4:

Which one of the following types of medicines is used for treating indigestion?

- (a) Antibiotic
(b) Analgesic
(c) Antacid
(d) Antiseptic

Answer

(c) Antacid is used for treating indigestion.

Question 5:

Write word equations and then balanced equations for the reaction taking place when –

- (a) dilute sulphuric acid reacts with zinc granules.
(b) dilute hydrochloric acid reacts with magnesium ribbon.
(c) dilute sulphuric acid reacts with aluminium powder.
(d) dilute hydrochloric acid reacts with iron filings.

Answer

- (a) Sulphuric acid + Zinc \rightarrow Zinc sulphate + Hydrogen
(b) Hydrochloric acid + Magnesium \rightarrow Magnesium chloride + Hydrogen
(c) Sulphuric acid + Aluminium \rightarrow Aluminium sulphate + Hydrogen
(d) Hydrochloric acid + Iron \rightarrow Ferric chloride + Hydrogen

Question 6:

Compounds such as alcohols and glucose also contain hydrogen but are not categorized as acids. Describe an activity to prove it.

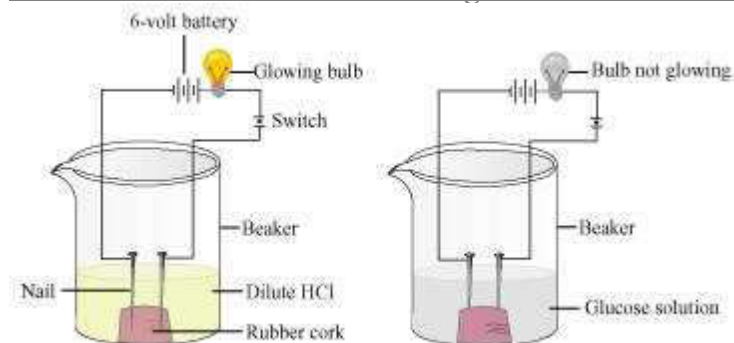
Answer

Two nails are fitted on a cork and are kept it in a 100 mL beaker. The nails are then connected to the two terminals of a 6-volt battery through a bulb and a switch. Some dilute HCl is poured in the beaker and the current is switched on. The same experiment is then performed with glucose solution and

alcohol solution.

Observations:

It will be observed that the bulb glows in the HCl solution and does not glow in the glucose solution.



Result:

HCl dissociates into H^+ and Cl^- ions. These ions conduct electricity in the solution resulting in the glowing of the bulb. On the other hand, the glucose solution does not dissociate into ions. Therefore, it does not conduct electricity.

Conclusion:

From this activity, it can be concluded that all acids contain hydrogen but not all compounds containing hydrogen are acids.

That is why, though alcohols and glucose contain hydrogen, they are not categorised as acids.

Question 7:

Why does distilled water not conduct electricity, whereas rain water does?

Answer

Distilled water is a pure form of water and is devoid of any ionic species. Therefore, it does not conduct electricity. Rain water, being an impure form of water, contains many ionic species such as acids and therefore it conducts electricity.

Question 8:

Why do acids not show acidic behaviour in the absence of water?

Answer

Acids do not show acidic behaviour in the absence of water because the dissociation of hydrogen ions from an acid occurs in the presence of water only. It is the hydrogen ions that are responsible for the acidic behaviour.

Question 9:

Five solutions A, B, C, D and E when tested with universal indicator showed pH as 4, 1, 11, 7 and 9, respectively. Which solution is

- (a) neutral?
- (b) strongly alkaline?
- (c) strongly acidic?
- (d) weakly acidic?
- (e) weakly alkaline?

Arrange the pH in increasing order of hydrogen-ion concentration.

Answer

- (a) Neutral \rightarrow Solution D with pH 7
- (b) Strongly alkaline \rightarrow Solution C with pH 11
- (c) Strongly acidic \rightarrow Solution B with pH 1
- (d) Weakly acidic \rightarrow Solution A with pH 4
- (e) Weakly alkaline \rightarrow Solution E with pH 9

The pH can be arranged in the increasing order of the concentration of hydrogen ions as: $11 < 9 < 7 < 4 < 1$

Question 10:

Equal lengths of magnesium ribbons are taken in test tubes A and B. Hydrochloric acid (HCl) is added to test tube A, while acetic acid (CH₃COOH) is added to test tube B. In which test tube will the fizzing occur more vigorously and why?

Answer

The fizzing will occur strongly in test tube A, in which hydrochloric acid (HCl) is added. This is because HCl is a stronger acid than CH₃COOH and therefore produces hydrogen gas at a faster speed due to which fizzing occurs.

Question 11:

Fresh milk has a pH of 6. How do you think the pH will change as it turns into curd? Explain your answer.

Answer

The pH of milk is 6. As it changes to curd, the pH will reduce because curd is acidic in nature. The acids present in it decrease the pH.

Question 12:

A milkman adds a very small amount of baking soda to fresh milk.

(a) Why does he shift the pH of the fresh milk from 6 to slightly alkaline?

(b) Why does this milk take a long time to set as curd?

Answer

(a) The milkman shifts the pH of the fresh milk from 6 to slightly alkaline because in alkaline condition, milk does not set as curd easily.

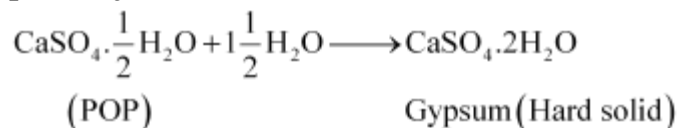
(b) Since this milk is slightly basic than usual milk, acids produced to set the curd are neutralized by the base. Therefore, it takes a longer time for the curd to set.

Question 13:

Plaster of Paris should be stored in a moisture-proof container. Explain why?

Answer

Plaster of Paris (POP) should be stored in a moisture-proof container because Plaster of Paris, a powdery mass, absorbs water (moisture) to form a hard solid known as gypsum.



Question 14:

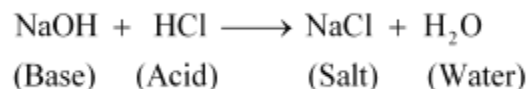
What is a neutralization reaction? Give two examples.

Answer

A reaction in which an acid and base react with each other to give a salt and water is termed as neutralization reaction. In this reaction, energy is evolved in the form of heat.

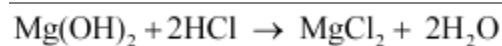
For example:

(i)



(ii) During indigestion (caused due to the production of excess of hydrochloric acid in the stomach), we administer an antacid (generally milk of magnesia, Mg(OH)₂ which is basic in nature). The

antacid neutralizes the excess of acids and thus gives relief from indigestion.



Question 15:

Give two important uses of washing soda and baking soda.

Answer

Two important uses of washing soda and baking soda are as follows:

(1) Washing soda:

(a) It is used in glass, soap, and paper industries.

(b) It is used to remove permanent hardness of water.

(2) Baking soda:

(a) It is used as baking powder. Baking powder is a mixture of baking soda and a mild acid known as tartaric acid. When it is heated or mixed in water, it releases CO₂ that makes bread or cake fluffy.

(b) It is used in soda-acid fire extinguishers.



NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 3 Metals and Non-metals Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 3 Metals and Non-metals Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 3 NCERT SOLUTIONS For Metals and Non-metals Download 2017 2018 New Edition PDF

Question 1:

Give an example of a metal which

- (i) is a liquid at room temperature.
- (ii) can be easily cut with a knife.
- (iii) is the best conductor of heat.
- (iv) is a poor conductor of heat.

Answer

- (i) Metal that exists in liquid state at room temperature → Mercury
- (ii) Metal that can be easily cut with a knife → Sodium
- (iii) Metal that is the best conductor of heat → Silver
- (iv) Metals that are poor conductors of heat → Mercury and lead

Question 2:

Explain the meanings of malleable and ductile.

Answer

Malleable: Substances that can be beaten into thin sheets are called malleable. For example, most of the metals are malleable.

Ductile: Substances that can be drawn into thin wires are called ductile. For example, most of the metals are ductile.

—

Question 1:

Define the following terms.

- (i) Mineral (ii) Ore (iii) Gangue

Answer

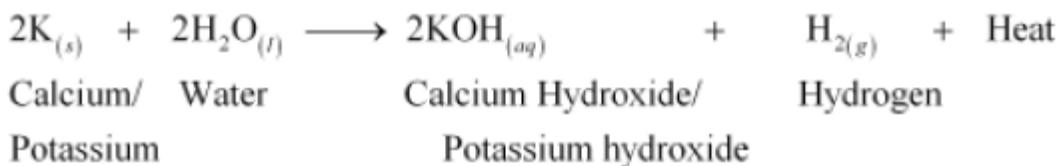
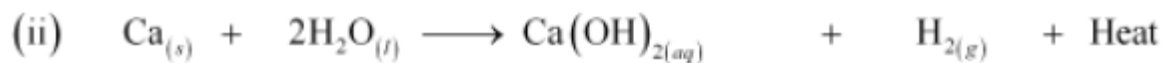
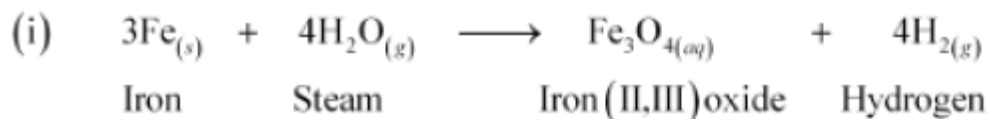
- (i) Mineral: Most of the elements occur in nature as in combined state as minerals. The chemical composition of minerals is fixed.
- (ii) Ore: Minerals from which metals can be extracted profitably are known as ores.
- (iii) Gangue: The impurities (sand, silt, soil, gravel, etc.) present in the ore are called gangue.

Question 2:

Write equations for the reactions of

- (i) iron with steam
- (ii) calcium and potassium with water

Answer



Question 3:

Samples of four metals A, B, C and D were taken and added to the following solution one by one. The results obtained have been tabulated as follows.

Metal	Iron (II) sulphate	Cooper (II) sulphate	Zinc sulphate	Silver nitrate
A.	No reaction	Displacement		
B.	Displacement		No reaction	
C.	No reaction	No reaction	No reaction	Displacement
D.	No reaction	No reaction	No reaction	No reaction

Use the Table above to answer the following questions about metals A, B, C and D.

(i) Which is the most reactive metal?

(ii) What would you observe if B is added to a solution of copper (II) sulphate?

(iii) Arrange the metals A, B, C and D in the order of decreasing reactivity.

Answer

Explanation

A + FeSO₄ → No reaction, i.e., A is less reactive than iron

A + CuSO₄ → Displacement, i.e., A is more reactive than copper

B + FeSO₄ → Displacement, i.e., B is more reactive than iron

B + ZnSO₄ → No reaction, i.e., B is less reactive than zinc

C + FeSO₄ → No reaction, i.e., C is less reactive than iron

C + CuSO₄ → No reaction, i.e., C is less reactive than copper

C + ZnSO₄ → No reaction, i.e., C is less reactive than zinc

C + AgNO₃ → Displacement, i.e., C is more reactive than silver

D + FeSO₄/CuSO₄/ZnSO₄/AgNO₃ → No reaction, i.e., D is less reactive than iron, copper, zinc, and silver

From the above equations, we obtain:

Zn	Most Reactive
B	
Fe	Reactivity Series
A	
Cu	
C	
Ag	
D	Least Reactive

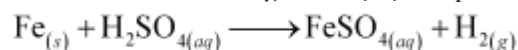
- (i) B is the most reactive metal.
(ii) If B is added to a solution of copper (II) sulphate, then it would displace copper.
 $B + CuSO_4 \rightarrow$ Displacement
(iii) The arrangement of the metals in the order of decreasing reactivity is:
 $B > A > C > D$

Question 4:

Which gas is produced when dilute hydrochloric acid is added to a reactive metal? Write the chemical reaction when iron reacts with dilute H_2SO_4 .

Answer

Hydrogen gas is evolved when dilute hydrochloric acid is added to a reactive metal. When iron reacts with dilute H_2SO_4 , iron (II) sulphate with the evolution of hydrogen gas is formed.



Question 5:

What would you observe when zinc is added to a solution of iron (II) sulphate? Write the chemical reaction that takes place.

Answer

Zinc is more reactive than iron. Therefore, if zinc is added to a solution of iron (II) sulphate, then it would displace iron from the solution.



=

Question 1:

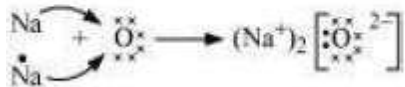
- (i) Write the electron-dot structures for sodium, oxygen and magnesium.
(ii) Show the formation of Na_2O and MgO by the transfer of electrons.
(iii) What are the ions present in these compounds?

Answer

- (i) The representation of elements with valence electrons as dots around the elements is referred to as electron-dot structure for elements.

- (a) Sodium (2, 8, 1) = Na^\bullet
 (b) Oxygen (2, 6) = $\text{:}\ddot{\text{O}}\text{:}$
 (c) Magnesium (2, 8, 2) = Mg^\bullet

(ii)



(iii) The ions present in Na_2O are Na^+ and O^{2-} ions and in MgO are Mg^{2+} and O^{2-} ions.

Question 2:

Name two metals which are found in nature in the free state.

Answer

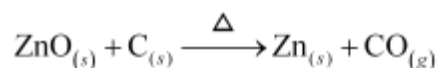
The metals at the bottom of the reactivity series are mostly found in free state. For example: gold, silver, and platinum.

Question 3:

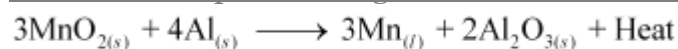
What chemical process is used for obtaining a metal from its oxide?

Answer

The chemical process used for obtaining a metal from its oxide is reduction. In this process, metal oxides are reduced by using suitable reducing agents such as carbon or by highly reactive metals to displace the metals from their oxides. For example, zinc oxide is reduced to metallic zinc by heating with carbon.



Manganese dioxide is reduced to manganese by treating it with aluminium powder. In this case, aluminium displaces manganese from its oxide.



Oxides of more reactive metals are reduced by electrolysis.

==

Question 1:

Metallic oxides of zinc, magnesium and copper were heated with the following metals.

Metal	Zinc	Magnesium	Copper
Zinc oxide	-	-	-
Magnesium oxide	-	-	-
Copper oxide	-	-	-

In which cases will you find displacement reactions taking place?

Answer

Metal	Zinc	Magnesium	Copper
Zinc oxide	No reaction	Displacement	No reaction
Magnesium oxide	No reaction	No reaction	No reaction
Copper oxide	Displacement	Displacement	No reaction

Question 2:

Which metals do not corrode easily?

Answer

More reactive a metal is, more likely it is to be corroded. Therefore, less reactive metals are less likely to get corroded. This is why gold plating provides high resistance to corrosion.

Question 3:

What are alloys?

Answer

Alloys are homogeneous mixtures of two or more elements. The elements could be two metals, or a metal and a non-metal. An alloy is formed by first melting the metal and then dissolving the other elements in it. For example, steel is an alloy of iron and carbon.

=

Question 1:

Which of the following pairs will give displacement reactions?

- (a) NaCl solution and copper metal
- (b) MgCl₂ solution and aluminium metal
- (c) FeSO₄ solution and silver metal
- (d) AgNO₃ solution and copper metal.

Answer

(d) AgNO_3 solution and copper metal

Question 2:

Which of the following methods is suitable for preventing an iron frying pan from rusting?

- (a) Applying grease
- (b) Applying paint
- (c) Applying a coating of zinc
- (d) all of the above.

Answer

(c) Applying a coating of zinc (We can also apply grease and paint to prevent iron from rusting. However, in case of iron frying pan, grease and paint cannot be applied because when the pan will be heated and washed again and again, the coating of grease and paint would get destroyed.)

Question 3:

An element reacts with oxygen to give a compound with a high melting point. This compound is also soluble in water. The element is likely to be

- (a) calcium
- (b) carbon
- (c) silicon
- (d) iron

Answer

(a) The element is likely to be calcium.

Question 4:

Food cans are coated with tin and not with zinc because

- (a) zinc is costlier than tin.
- (b) zinc has a higher melting point than tin.
- (c) zinc is more reactive than tin.
- (d) zinc is less reactive than tin.

Answer

(c) Food cans are coated with tin and not with zinc because zinc is more reactive than tin.

Question 5:

You are given a hammer, a battery, a bulb, wires and a switch.

- (a) How could you use them to distinguish between samples of metals and non-metals?
- (b) Assess the usefulness of these tests in distinguishing between metals and nonmetals.

Answer

(a) With the hammer, we can beat the sample and if it can be beaten into thin sheets (that is, it is malleable), then it is a metal otherwise a non-metal. Similarly, we can use the battery, bulb, wires, and a switch to set up a circuit with the sample. If the sample conducts electricity, then it is a metal otherwise a non-metal.

(b) The above tests are useful in distinguishing between metals and non-metals as these are based on the physical properties. No chemical reactions are involved in these tests.

Question 6:

What are amphoteric oxides? Give two examples of amphoteric oxides.

Answer

Those oxides that behave as both acidic and basic oxides are called amphoteric oxides. Examples: aluminium oxide (Al_2O_3), zinc oxide (ZnO)

Question 7:

Name two metals which will displace hydrogen from dilute acids, and two metals which will not.

Answer

Metals that are more reactive than hydrogen displace it from dilute acids. For example: sodium and potassium. Metals that are less reactive than hydrogen do not displace it. For example: copper and silver.

Question 8:

In the electrolytic refining of a metal M, what would you take as the anode, the cathode and the electrolyte?

Answer

In the electrolytic refining of a metal M:

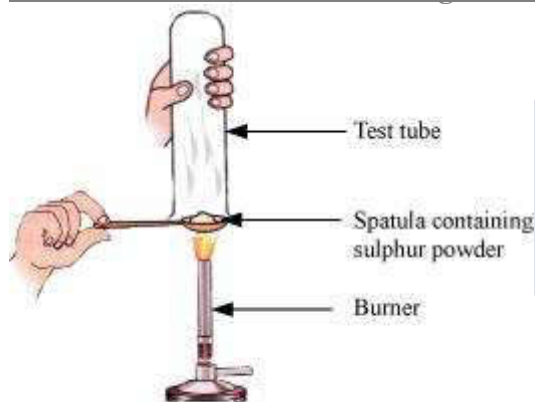
Anode → Impure metal M

Cathode → Thin strip of pure metal M

Electrolyte → Solution of salt of the metal M

Question 9:

Pratyush took sulphur powder on a spatula and heated it. He collected the gas evolved by inverting a test tube over it, as shown in figure below.



Collection of gas

(a) What will be the action of gas on

(i) dry litmus paper?

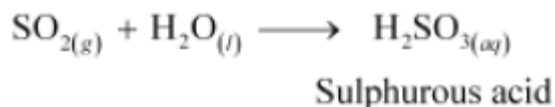
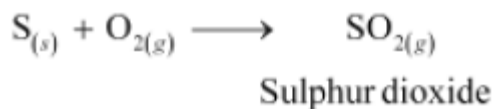
(ii) moist litmus paper?

(b) Write a balanced chemical equation for the reaction taking place.

Answer

(a) (i) There will be no action on dry litmus paper.

(ii) Since the gas is sulphur dioxide (SO₂), it turns moist blue litmus paper to red because sulphur dioxide reacts with moisture to form sulphurous acid.



(b)

Question 10:

State two ways to prevent the rusting of iron.

Answer

Two ways to prevent the rusting of iron are:

(i) Oiling, greasing, or painting: By applying oil, grease, or paint, the surface becomes water proof and the moisture and oxygen present in the air cannot come into direct contact with iron. Hence, rusting is prevented.

(ii) Galvanisation: An iron article is coated with a layer of zinc metal, which prevents the iron to come in contact with oxygen and moisture. Hence, rusting is prevented.

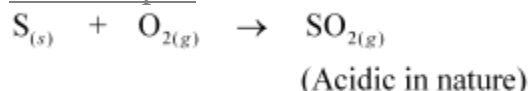
Question 11:

What type of oxides is formed when non-metals combine with oxygen?

Answer

Non-metals combine with oxygen to form acidic oxides.

For example:



Question 12:

Give reasons

(a) Platinum, gold and silver are used to make jewellery.

(b) Sodium, potassium and lithium are stored under oil.

(c) Aluminium is a highly reactive metal, yet it is used to make utensils for cooking.

(d) Carbonate and sulphide ores are usually converted into oxides during the process of extraction.

Answer

(a) Platinum, gold, and silver are used to make jewellery because they are very lustrous. Also, they are very less reactive and do not corrode easily.

(b) Sodium, potassium, and lithium are very reactive metals and react very vigorously with air as well as water. Therefore, they are kept immersed in kerosene oil in order to prevent their contact with air and moisture.

(c) Though aluminium is a highly reactive metal, it is resistant to corrosion. This is because aluminium reacts with oxygen present in air to form a thin layer of aluminium oxide. This oxide layer is very stable and prevents further reaction of aluminium with oxygen. Also, it is light in weight and a good conductor of heat. Hence, it is used to make cooking utensils.

(d) Carbonate and sulphide ores are usually converted into oxides during the process of extraction because metals can be easily extracted from their oxides rather than from their carbonates and sulphides.

Question 13:

You must have seen tarnished copper vessels being cleaned with lemon or tamarind juice. Explain why these sour substances are effective in cleaning the vessels.

Answer

Copper reacts with moist carbon dioxide in air to form copper carbonate and as a result, copper vessel loses its shiny brown surface forming a green layer of copper carbonate. The citric acid present in the lemon or tamarind neutralises the basis copper carbonate and dissolves the layer. That is why, tarnished copper vessels are cleaned with lemon or tamarind juice to give the surface of the copper vessel its characteristic lustre.

Question 14:

Differentiate between metal and non-metal on the basis of their chemical properties.

Answer

Metal	Non-metal
Metals are electropositive.	Non-metals are electronegative.
They react with oxygen to form basic oxides. $4\text{Na} + \text{O}_2 \longrightarrow 2\text{Na}_2\text{O}$ These have ionic bonds.	They react with oxygen to form acidic or neutral oxides. $\text{C} + \text{O}_2 \longrightarrow \text{CO}_2$ These have covalent bonds.
They react with water to form oxides and hydroxides. Some metals react with cold water, some with hot water, and some with steam. $2\text{Na} + 2\text{H}_2\text{O} \longrightarrow 2\text{NaOH} + \text{H}_2 \uparrow$	They do not react with water.
They react with dilute acids to form a salt and evolve hydrogen gas. However, Cu, Ag, Au, Pt, Hg do not react. $2\text{Na} + 2\text{HCl} \longrightarrow 2\text{NaCl} + \text{H}_2 \uparrow$	They do not react with dilute acids. These are not capable of replacing hydrogen.
They react with the salt solution of metals. Depending on their reactivity, displacement reaction can occur. $\text{CuSO}_4 + \text{Zn} \longrightarrow \text{ZnSO}_4 + \text{Cu}$	These react with the salt solution of non-metals.
They act as reducing agents (as they can easily lose electrons). $\text{Na} \longrightarrow \text{Na}^+ + \text{e}^-$	These act as oxidising agents (as they can gain electrons). $\text{Cl}_2 + 2\text{e}^- \longrightarrow 2\text{Cl}^-$

Question 15:

A man went door to door posing as a goldsmith. He promised to bring back the glitter of old and dull gold ornaments. An unsuspecting lady gave a set of gold bangles to him which he dipped in a particular solution. The bangles sparkled like new but their weight was reduced drastically. The lady was upset but after a futile argument the man beat a hasty retreat. Can you play the detective to find out the nature of the solution he had used?

Answer

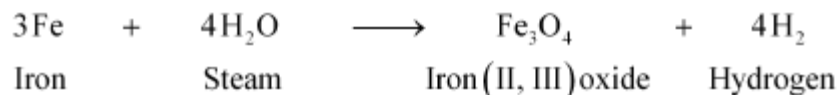
He must have dipped the gold metal in the solution of aqua regia – a 3:1 mixture of conc. HCl and conc. HNO₃. Aqua regia is a fuming, highly corrosive liquid. It dissolves gold in it. After dipping the gold ornaments in aqua regia, the outer layer of gold gets dissolved and the inner shiny layer appears. That is why the weight of gold ornament reduced.

Question 16:

Give reasons why copper is used to make hot water tanks and not steel (an alloy of iron).

Answer

Copper does not react with cold water, hot water, or steam. However, iron reacts with steam. If the hot water tanks are made of steel (an alloy of iron), then iron would react vigorously with the steam formed from hot water.



That is why copper is used to make hot water tanks, and not steel.

=

Question 1:

Why is sodium kept immersed in kerosene oil?

Answer

Sodium and potassium are very reactive metals and combine explosively with air as well as water. Hence, they catch fire if kept in open. Therefore, to prevent accidental fires and accidents, sodium is stored immersed in kerosene oil.

Question 2:

Why do ionic compounds have high melting points?

Answer

Ionic compounds have strong electrostatic forces of attraction between the ions. Therefore, it requires a lot of energy to overcome these forces. That is why ionic compounds have high melting points.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 4 Carbon and its Compounds Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 4 Carbon and its Compounds Download 2017 2018 New Edition PDF>

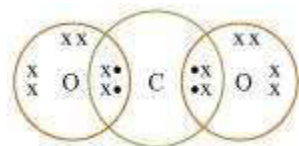
Class 10 Science Chapter 4 NCERT SOLUTIONS For Carbon and its Compounds Download 2017 2018 New Edition PDF

Question 1:

What would be the electron dot structure of carbon dioxide which has the formula CO_2 ?

Answer:

Electron dot structure of CO_2 is

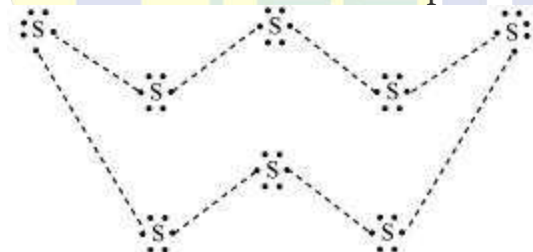


Question 2:

What would be the electron dot structure of a molecule of sulphur which is made up of eight atoms of sulphur? (Hint – the eight atoms of sulphur are joined together in the form of a ring.)

Answer:

Electron dot structure of a sulphur molecule



WE ARE WITH YOU.....

Question 1:

How many structural isomers can you draw for pentane?

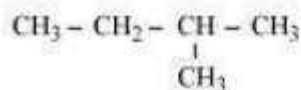
Answer:

Three structural isomers are possible for pentane.

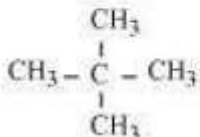
(i)



(ii)



(iii)



Question 2:

What are the two properties of carbon which lead to the huge number of carbon compounds we see around us?

Answer:

The two features of carbon that give rise to a large number of compounds are as follows:

(i) Catenation – It is the ability to form bonds with other atoms of carbon.

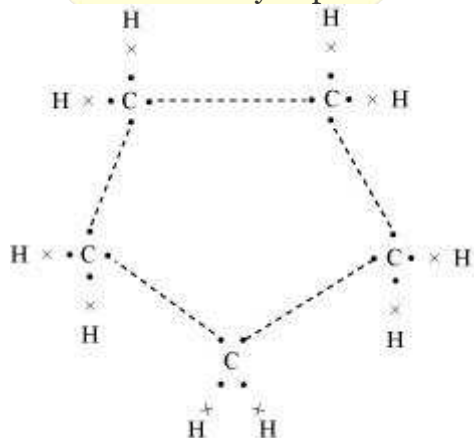
(ii) Tetravalency – With the valency of four, carbon is capable of bonding with four other atoms.

Question 3:

What will be the formula and electron dot structure of cyclopentane?

Answer:

The formula for cyclopentane is C_5H_{10} . Its electron dot structure is given below.



WE ARE WITH YOU.....

Question 4:

Draw the structures for the following compounds.

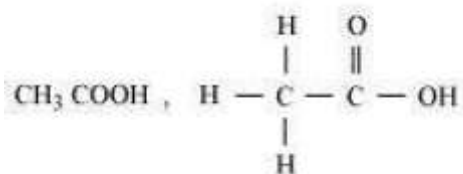
(i) Ethanoic acid (ii) Bromopentane*

(iii) Butanone (iv) Hexanal

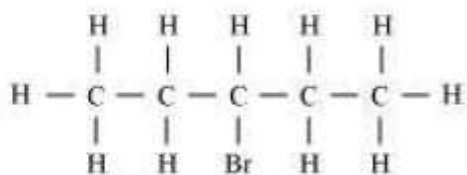
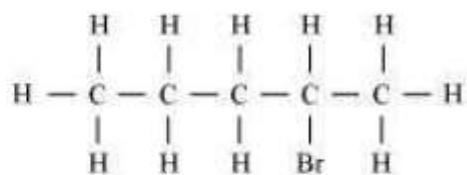
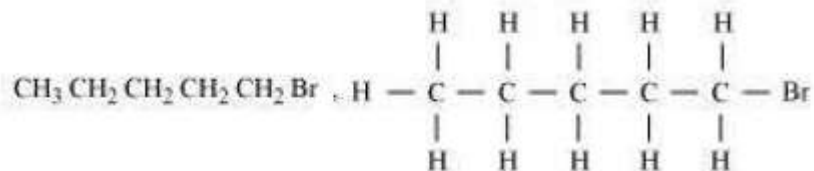
*Are structural isomers possible for bromopentane?

Answer:

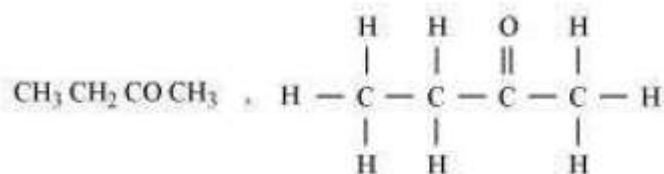
(i)



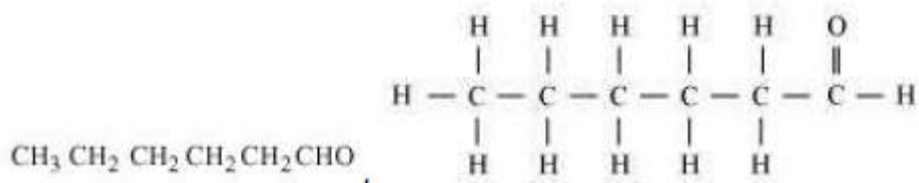
(ii) There are many structural isomers possible for bromopentane. Among them, the structures of three isomers are given.



(iii)



(iv)



Question 2:

Why do ionic compounds have high melting points?

Answer:

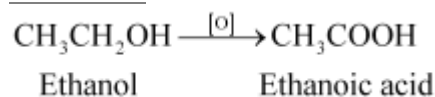
Ionic compounds have strong electrostatic forces of attraction between the ions. Therefore, it requires a lot of energy to overcome these forces. That is why ionic compounds have high melting points.

=

Question 1:

Why is the conversion of ethanol to ethanoic acid an oxidation reaction?

Answer:

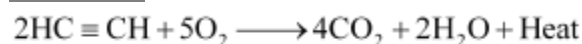


Since the conversion of ethanol to ethanoic acid involves the addition of oxygen to ethanol, it is an oxidation reaction.

Question 2:

A mixture of oxygen and ethyne is burnt for welding. Can you tell why a mixture of ethyne and air is not used?

Answer:



When ethyne is burnt in air, it gives a sooty flame. This is due to incomplete combustion caused by limited supply of air. However, if ethyne is burnt with oxygen, it gives a clean flame with temperature 3000°C because of complete combustion. This oxy-acetylene flame is used for welding. It is not possible to attain such a high temperature without mixing oxygen. This is the reason why a mixture of ethyne and air is not used.

=

Question 1:

How would you distinguish experimentally between an alcohol and a carboxylic acid?

Answer:

We can distinguish between an alcohol and a carboxylic acid on the basis of their reaction with carbonates and hydrogen carbonates. Acid reacts with carbonate and hydrogen carbonate to evolve CO₂ gas that turns lime water milky.

Metal Carbonate/ Metal Hydrogencarbonate + Carboxylic acid



Salt + Water + Carbon dioxide

Alcohols, on the other hand, do not react with carbonates and hydrogen carbonates.

Question 2:

What are oxidising agents?

Answer:

Some substances such as alkaline potassium permanganate and acidified potassium dichromate are capable of adding oxygen to others. These are known as oxidising agents.

=

Question 1:

Would you be able to check if water is hard by using a detergent?

Answer:

Detergents are ammonium or sulphonate salts of long chain carboxylic acids. Unlike soap, they do not react with calcium and magnesium ions present in hard water to form scum. They give a good amount of lather irrespective of whether the water is hard or soft. This means that detergents can be used in both soft and hard water. Therefore, it cannot be used to check whether the water is hard or not.

Question 2:

People use a variety of methods to wash clothes. Usually after adding the soap, they 'beat' the clothes on a stone, or beat it with a paddle, scrub with a brush or the mixture is agitated in a washing machine. Why is agitation necessary to get clean clothes?

Answer:

A soap molecule has two parts namely hydrophobic and hydrophilic. With the help of these, it attaches to the grease or dirt particle and forms a cluster called micelle. These micelles remain suspended as a colloid. To remove these micelles (entrapping the dirt), it is necessary to agitate clothes.

=

Question 1:

Ethane, with the molecular formula C_2H_6 has

- (a) 6 covalent bonds.
- (b) 7 covalent bonds.
- (c) 8 covalent bonds.
- (d) 9 covalent bonds.

Answer:

(b) Ethane has 7 covalent bonds.

Question 2:

Butanone is a four-carbon compound with the functional group

- (a) carboxylic acid.
- (b) aldehyde.
- (c) ketone.
- (d) alcohol.

Answer:

(c) The functional group of butanone is ketone.

Question 3:

While cooking, if the bottom of the vessel is getting blackened on the outside, it means that

- (a) the food is not cooked completely.
- (b) the fuel is not burning completely.
- (c) the fuel is wet.
- (d) the fuel is burning completely.

Answer:

(b) While cooking, if the bottom of the vessel is getting blackened on the outside, then it means that the fuel is not burning completely.

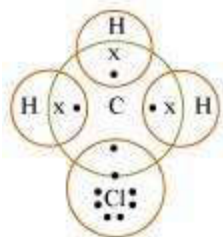
Question 4:

Explain the nature of the covalent bond using the bond formation in CH_3Cl .

Answer:

Carbon can neither lose four of its electrons nor gain four electrons as both the processes require

extra amount of energy and would make the system unstable. Therefore, it completes its octet by sharing its four electrons with other carbon atoms or with atoms of other elements. The bonds that are formed by sharing electrons are known as covalent bonds. In covalent bonding, both the atoms share the valence electrons, i.e., the shared electrons belong to the valence shells of both the atoms.



Here, carbon requires 4 electrons to complete its octet, while each hydrogen atom requires one electron to complete its duplet. Also, chlorine requires an electron to complete the octet. Therefore, all of these share the electrons and as a result, carbon forms 3 bonds with hydrogen and one with chlorine.

Question 5:

Draw the electron dot structures for

(a) ethanoic acid.

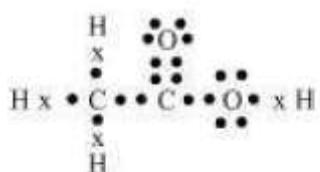
(b) H_2S .

(c) propanone.

(d) F_2 .

Answer:

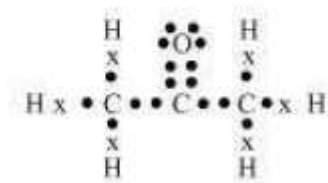
(a) Ethanoic acid



(b) H_2S



(c) Propanone



(d) F_2



Question 6:

What is a homologous series? Explain with an example.

Answer:

A homologous series is a series of carbon compounds that have different numbers of carbon atoms

but contain the same functional group. For example, methane, ethane, propane, butane, etc. are all part of the alkane homologous series. The general formula of this series is C_nH_{2n+2} .

Methane CH_4

Ethane CH_3CH_3

Propane $CH_3CH_2CH_3$

Butane $CH_3CH_2CH_2CH_3$

It can be noticed that there is a difference of $-CH_2$ unit between each successive compound.

Question 7:

How can ethanol and ethanoic acid be differentiated on the basis of their physical and chemical properties?

Answer:

– Ethanol is a liquid at room temperature with a pleasant odour while ethanoic acid has vinegar-like smell. The melting point of ethanoic acid is $17^\circ C$. This is below room temperature and hence, it freezes during winters.

– Ethanoic acid reacts with metal carbonates and metal hydrogencarbonates to form salt, water, and carbon dioxide gas while ethanol does not react with them.

Metal Carbonates/Metal Hydrogencarbonates + Carboxylic acid



Salt + Water + Carbon dioxide

For example,

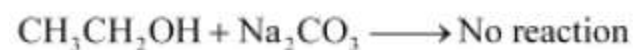


Metal Carbonates/Metal Hydrogencarbonates + Alcohols



No reaction

For example,



Question 8:

In the electrolytic refining of a metal M, what would you take as the anode, the cathode and the electrolyte?

Answer:

In the electrolytic refining of a metal M:

Anode \rightarrow Impure metal M

Cathode \rightarrow Thin strip of pure metal M

Electrolyte \rightarrow Solution of salt of the metal M

Question 9:

Why are carbon and its compounds used as fuels for most applications?

Answer:

Most of the carbon compounds give a lot of heat and light when burnt in air. Saturated hydrocarbons burn with a clean flame and no smoke is produced. The carbon compounds, used as a fuel, have high calorific values. Therefore, carbon and its compounds are used as fuels for most applications.

very
TH YOU.....

Question 10:

Explain the formation of scum when hard water is treated with soap.

Answer:

Soap does not work properly when the water is hard. A soap is a sodium or potassium salt of long chain fatty acids. Hard water contains salts of calcium and magnesium. When soap is added to hard water, calcium and magnesium ions present in water displace sodium or potassium ions from the soap molecules forming an insoluble substance called scum. A lot of soap is wasted in the process.

Question 11:

What change will you observe if you test soap with litmus paper (red and blue)?

Answer:

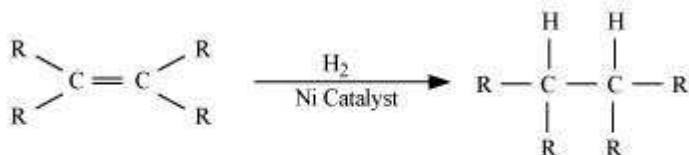
Since soap is basic in nature, it will turn red litmus blue. However, the colour of blue litmus will remain blue.

Question 12:

What is hydrogenation? What is its industrial application?

Answer:

Hydrogenation is the process of addition of hydrogen. Unsaturated hydrocarbons are added with hydrogen in the presence of palladium and nickel catalysts to give saturated hydrocarbons.



This reaction is applied in the hydrogenation of vegetables oils, which contain long chains of unsaturated carbons.

Question 13:

Which of the following hydrocarbons undergo addition reactions: C₂H₆, C₃H₈, C₃H₆, C₂H₂ and CH₄.

Answer:

Unsaturated hydrocarbons undergo addition reactions. Being unsaturated hydrocarbons, C₃H₆ and C₂H₂ undergo addition reactions.

Question 14:

Give a test that can be used to differentiate chemically between butter and cooking oil.

Answer:

Butter contains saturated fats. Therefore, it cannot be hydrogenated. On the other hand, oil has unsaturated fats. That is why it can be hydrogenated to saturated fats (solids).

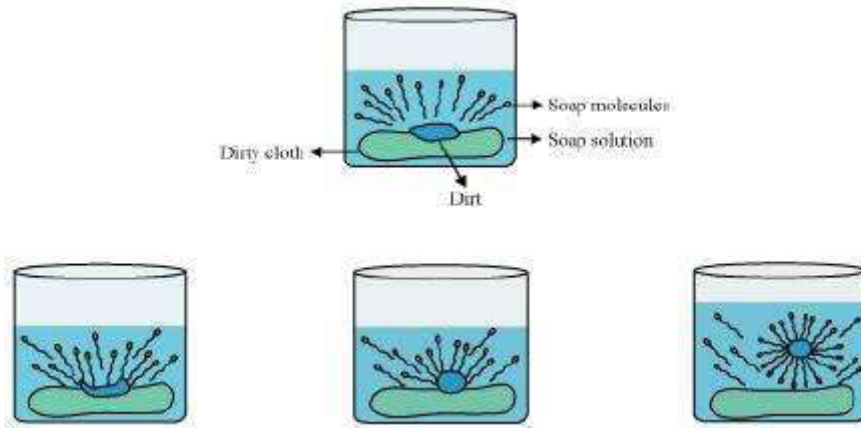
Question 15:

Explain the mechanism of the cleaning action of soaps.

Answer:

Cleansing action of soaps: The dirt present on clothes is organic in nature and insoluble in water. Therefore, it cannot be removed by only washing with water. When soap is dissolved in water, its hydrophobic ends attach themselves to the dirt and remove it from the cloth. Then, the molecules of soap arrange themselves in micelle formation and trap the dirt at the centre of the cluster. These

micelles remain suspended in the water. Hence, the dust particles are easily rinsed away by water.



Free Home Delivery

WE ARE WITH YOU.....

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 5 Periodic Classification of Elements Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 5 Periodic Classification of Elements Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 5 NCERT SOLUTIONS For Periodic Classification of Elements Download 2017 2018 New Edition PDF

Question 1:

Did Dobereiner's triads also exist in the columns of Newlands' Octaves? Compare and find out.

Answer:

Only one triad of Dobereiner's triads exists in the columns of Newlands' octaves. The triad formed by the elements Li, Na, and K of Dobereiner's triads also occurred in the columns of Newlands' octaves. Dobereiner's triads

Li Ca Cl

Na Sr Br

K Ba I

Newlands' octaves

H	Li	Be	B	C	N	O
F	Na	Mg	Al	Si	P	S
Cl	K	Ca	Cr	Ti	Mn	Fe
Co and Ni	Cu	Zn	Y	In	As	Se
Br	Rb	Sr	Ce and La	Zr	—	—

Question 2:

What were the limitations of Dobereiner's classification?

Answer:

Limitation of Dobereiner's classification: All known elements could not be classified into groups of triads on the basis of their properties.

Question 3:

What were the limitations of Newlands' Law of Octaves?

Answer:

Limitations of Newlands' law of octaves:

(i) It was not applicable throughout the arrangements. It was applicable up to calcium only. The

properties of the elements listed after calcium showed no resemblance to the properties of the elements above them.

(ii) Those elements that were discovered after Newlands' octaves did not follow the law of octaves.

(iii) The position of cobalt and nickel in the group of the elements (F, Cl) of different properties could not be explained.

(iv) Placing of iron far away from cobalt and nickel, which have similar properties as iron, could also not be explained.

–

Question 1:

Use Mendeleev's Periodic Table to predict the formulae for the oxides of the following elements: K, C, Al, Si, Ba.

Answer:

K is in group 1. Therefore, the oxide will be K_2O .

C is in group 4. Therefore, the oxide will be CO_2 .

Al is in group 3. Therefore, the oxide will be Al_2O_3 .

Si is in group 4. Therefore, the oxide will be SiO_2 .

Ba is in group 2. Therefore, the oxide will be BaO .

Question 2:

Besides gallium, which other elements have since been discovered that were left by Mendeleev in his Periodic Table? (any two)

Answer:

Scandium and germanium

Question 3:

What were the criteria used by Mendeleev in creating his Periodic Table?

Answer:

Mendeleev's periodic table was based on the observation that the properties of elements are a periodic function of their atomic masses. This means that if elements are arranged in the increasing order of their atomic masses, then their properties get repeated after regular intervals

Question 4:

Why do you think the noble gases are placed in a separate group?

Answer:

Noble gases are inert elements. Their properties are different from the all other elements. Therefore, the noble gases are placed in a separate group.

–

Question 1:

How could the Modern Periodic Table remove various anomalies of Mendeleev's Periodic Table?

Answer:

Mendeleev was unable to give fixed position to hydrogen and isotopes in the periodic table. In Mendeleev's periodic table, the increasing manner of atomic mass of the elements is not always regular from one to its next. It was believed that a more fundamental property than atomic mass could explain periodic properties in a better manner.

It was Henry Moseley who demonstrated that atomic number of an element could explain periodic properties in a better way than atomic mass of an element and arranged the elements in increasing

order of their atomic numbers. Then it was found that the various anomalies of Mendeleev's periodic table were removed by the modern periodic table.

Question 2:

Name two elements you would expect to show chemical reactions similar to magnesium. What is the basis for your choice?

Answer:

Calcium (Ca) and strontium (Sr) are expected to show chemical reactions similar to magnesium (Mg). This is because the number of valence electrons (2) is same in all these three elements. And since chemical properties are due to valence electrons, they show same chemical reactions.

Question 3:

Name

(a) three elements that have a single electron in their outermost shells.

(b) two elements that have two electrons in their outermost shells.

(c) three elements with filled outermost shells.

Answer:

(a) Lithium (Li), sodium (Na), and potassium (K) have a single electron in their outermost shells.

(b) Magnesium (Mg) and calcium (Ca) have two electrons in their outermost shells.

(c) Neon (Ne), argon (Ar), and xenon (Xe) have filled outermost shells.

Question 4:

(a) Lithium, sodium, potassium are all metals that react with water to liberate hydrogen gas. Is there any similarity in the atoms of these elements?

(b) Helium is an unreactive gas and neon is a gas of extremely low reactivity. What, if anything, do their atoms have in common?

Answer:

(a) Yes. The atoms of all the three elements lithium, sodium, and potassium have one electron in their outermost shells.

(b) Both helium (He) and neon (Ne) have filled outermost shells. Helium has a duplet in its K shell, while neon has an octet in its L shell.

Question 5:

In the Modern Periodic Table, which are the metals among the first ten elements?

Answer:

Among the first ten elements, lithium (Li) and beryllium (Be) are metals.

Question 6:

By considering their position in the Periodic Table, which one of the following elements would you expect to have maximum metallic characteristic?

Answer:

Since Be lies to the extreme left hand side of the periodic table, Be is the most metallic among the given elements.

=

Question 1:

Which of the following statements is not a correct statement about the trends when going from left to right across the periods of periodic Table.

(a) The elements become less metallic in nature.

- (b) The number of valence electrons increases.
- (c) The atoms lose their electrons more easily.
- (d) The oxides become more acidic.

Answer:

- (c) The atoms lose their electrons more easily.

(On moving from left to right across the periods of the periodic table, the non-metallic character increases. Hence, the tendency to lose electrons decreases.)

Question 2:

Element X forms a chloride with the formula XCl_2 , which is a solid with a high melting point. X would most likely be in the same group of the Periodic Table as

- (a) Na (b) Mg (c) Al (d) Si

Answer:

- (b) X would most likely be in the same group of the Periodic Table as magnesium (Mg).

Question 3:

Which element has

- (a) two shells, both of which are completely filled with electrons?
- (b) the electronic configuration 2, 8, 2?
- (c) a total of three shells, with four electrons in its valence shell?
- (d) a total of two shells, with three electrons in its valence shell?
- (e) twice as many electrons in its second shell as in its first shell?

Answer:

- (a) Neon has two shells, both of which are completely filled with electrons (2 electrons in K shell and 8 electrons in L shell).
- (b) Magnesium has the electronic configuration 2, 8, 2.
- (c) Silicon has a total of three shells, with four electrons in its valence shell (2 electrons in K shell, 8 electrons in L shell and 4 electrons in M shell).
- (d) Boron has a total of two shells, with three electrons in its valence shell (2 electrons in K shell and 3 electrons in L shell).
- (e) Carbon has twice as many electrons in its second shell as in its first shell (2 electrons in K shell and 4 electrons in L shell).

Question 4:

- (a) What property do all elements in the same column of the Periodic Table as boron have in common?
- (b) What property do all elements in the same column of the Periodic Table as fluorine have in common?

Answer:

- (a) All the elements in the same column as boron have the same number of valence electrons (3). Hence, they all have valency equal to 3.
- (b) All the elements in the same column as fluorine have the same number of valence electrons (7). Hence, they all have valency equal to 1.

Question 5:

An atom has electronic configuration 2, 8, 7.

- (a) What is the atomic number of this element?
- (b) To which of the following elements would it be chemically similar? (Atomic numbers are given in parentheses.)
N(7) F(9) P(15) Ar(18)

Answer:

(a) The atomic number of this element is 17.

(b) It would be chemically similar to F(9) with configuration as 2, 7.

Question 6:

The position of three elements A, B and C in the Periodic Table are shown below –

Group 16 Group 17

– –

– A

– –

B C

(a) State whether A is a metal or non-metal.

(b) State whether C is more reactive or less reactive than A.

(c) Will C be larger or smaller in size than B?

(d) Which type of ion, cation or anion, will be formed by element A?

Answer:

(a) A is a non-metal.

(b) C is less reactive than A, as reactivity decreases down the group in halogens.

(c) C will be smaller in size than B as moving across a period, the nuclear charge increases and therefore, electrons come closer to the nucleus.

(d) A will form an anion as it accepts an electron to complete its octet.

Question 7:

Nitrogen (atomic number 7) and phosphorus (atomic number 15) belong to group 15 of the Periodic Table. Write the electronic configuration of these two elements. Which of these will be more electronegative? Why?

Answer:

Element..... K... L... M

Nitrogen 2... 5

Phosphorus. 2... 8... 5

Nitrogen is more electronegative than phosphorus. On moving down a group, the number of shell increases. Therefore, the valence electrons move away from the nucleus and the effective nuclear charge decreases. This causes the decrease in the tendency to attract electron and hence electronegativity decreases.

Question 8:

How does the electronic configuration of an atom relate to its position in the Modern Periodic Table?

Answer:

In the modern periodic table, atoms with similar electronic configurations are placed in the same column. In a group, the number of valence electrons remains the same. Elements across a period show an increase in the number of valence electrons.

Question 9:

In the Modern Periodic Table, calcium (atomic number 20) is surrounded by elements with atomic numbers 12, 19, 21, and 38. Which of these have physical and chemical properties resembling calcium?

Answer:

The element with atomic number 12 has same chemical properties as that of calcium. This is because both of them have same number of valence electrons (2).

Question 10:

Compare and contrast the arrangement of elements in Mendeleev's periodic Table and the Modern Periodic Table.

Answer:

	Mendeleev's periodic table		Modern periodic table
1.	Elements are arranged in the increasing order of their atomic masses.	1.	Elements are arranged in the increasing order of their atomic numbers.
2.	There are a total of 7 groups (columns) and 6 periods (rows).	2.	There are a total of 18 groups (columns) and 7 periods (rows).
3.	Elements having similar properties were placed directly under one another.	3.	Elements having the same valence shell are present in the same period while elements having the same number of valence electrons are present in the same group.
4.	The position of hydrogen could not be explained.	4.	Hydrogen is placed above alkali metals.
5.	No distinguishing positions for metals and non-metals.	5.	Metals are present at the left hand side of the periodic table whereas non-metals are present at the right hand side.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 6 Life Processes>

Download 2017 2018 New Edition PDF

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 6 Life Processes>
Download 2017 2018 New Edition PDF

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 6 Life Processes> Download 2017 2018 New Edition PDF

Question 1:

Why is diffusion insufficient to meet the oxygen requirements of multi-cellular organisms like humans?

Answer

Multicellular organisms such as humans possess complex body designs. They have specialised cells and tissues for performing various necessary functions of the body such as intake of food and oxygen. Unlike unicellular organisms, multicellular cells are not in direct contact with the outside environment. Therefore, diffusion cannot meet their oxygen requirements.

Question 2:

What criteria do we use to decide whether something is alive?

Answer

Any visible movement such as walking, breathing, or growing is generally used to decide whether something is alive or not. However, a living organism can also have movements, which are not visible to the naked eye. Therefore, the presence of life processes is a fundamental criterion that can be used to decide whether something is alive or not.

Question 3:

What are outside raw materials used for by an organism?

Answer

An organism uses outside raw materials mostly in the form of food and oxygen. The raw materials required by an organism can be quite varied depending on the complexity of the organism and its environment.

Question 4:

What processes would you consider essential for maintaining life?

Answer

Life processes such as nutrition, respiration, transportation, excretion, etc. are essential for maintaining life.

—

Question 1:

What are the differences between autotrophic nutrition and heterotrophic nutrition?

Answer

	Autotrophic nutrition		Heterotrophic nutrition
(i)	Food is synthesised from simple inorganic raw materials such as CO ₂ and water.	(i)	Food is obtained directly or indirectly from autotrophs. This food is broken down with the help of enzymes.
(ii)	Presence of green pigment (chlorophyll) is necessary.	(ii)	No pigment is required in this type of nutrition.
(iii)	Food is generally prepared during day time.	(iii)	Food can be prepared at all times.
(iv)	All green plants and some bacteria have this type of nutrition.	(iv)	All animals and fungi have this type of nutrition.

Question 2:

Where do plants get each of the raw materials required for photosynthesis?

Answer

The following raw materials are required for photosynthesis:

- The raw material CO₂ enters from the atmosphere through stomata.
- Water is absorbed from the soil by the plant roots.
- Sunlight, an important component to manufacture food, is absorbed by the chlorophyll and other green parts of the plants.

Question 3:

What is the role of the acid in our stomach?

Answer

The hydrochloric acid present in our stomach dissolves bits of food and creates an acidic medium. In this acidic medium, enzyme pepsinogen is converted to pepsin, which is a protein-digesting enzyme.

Question 4:

What is the function of digestive enzymes?

Answer

Digestive enzymes such as amylase, lipase, pepsin, trypsin, etc. help in the breaking down of complex food particles into simple ones. These simple particles can be easily absorbed by the blood and thus transported to all the cells of the body.

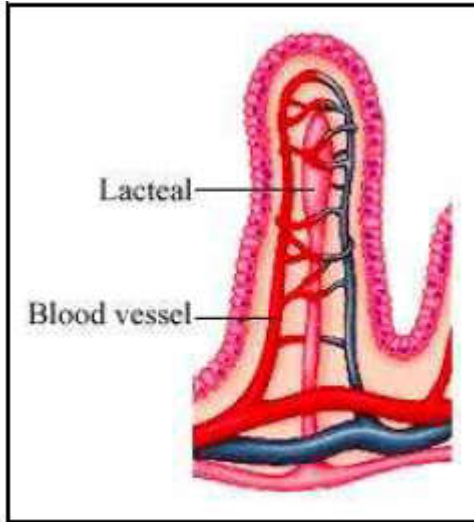
Question 5:

How is the small intestine designed to absorb digested food?

Answer

The small intestine has millions of tiny finger-like projections called villi. These villi increase the surface area for more efficient food absorption. Within these villi, many blood vessels are present that absorb the digested food and carry it to the blood stream. From the blood stream,

the absorbed food is delivered to each and every cell of the body.



Enlarged view of a villus

=

Question 1:

What advantage over an aquatic organism does a terrestrial organism have with regard to obtaining oxygen for respiration?

Answer

Terrestrial organisms take up oxygen from the atmosphere whereas aquatic animals need to utilize oxygen present in the water. Air contains more O₂ as compared to water. Since the content of O₂ in air is high, the terrestrial animals do not have to breathe faster to get more oxygen. Therefore, unlike aquatic animals, terrestrial animals do not have to show various adaptations for better gaseous exchange.

Question 2:

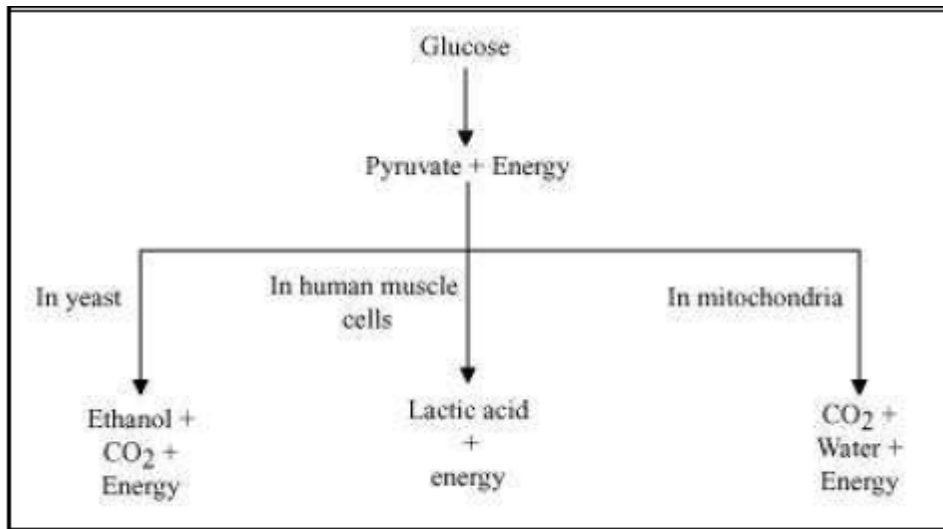
What are the different ways in which glucose is oxidized to provide energy in various organisms?

Answer

Glucose is first broken down in the cell cytoplasm into a three carbon molecule called pyruvate. Pyruvate is further broken down by different ways to provide energy.

The breakdown of glucose by different pathways can be illustrated as follows.

WE ARE WITH YOU.....



In yeast and human muscle cells, the breakdown of pyruvate occurs in the absence of oxygen whereas in mitochondria, the breakdown of pyruvate occurs in the presence of oxygen.

Question 3:

How is oxygen and carbon dioxide transported in human beings?

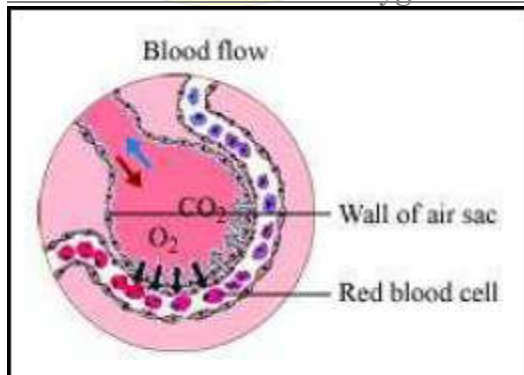
Answer

Haemoglobin transports oxygen molecule to all the body cells for cellular respiration.

The haemoglobin pigment present in the blood gets attached to four O₂ molecules that are obtained from breathing. It thus forms oxyhaemoglobin and the blood becomes oxygenated.

This oxygenated blood is then distributed to all the body cells by the heart. After giving away O₂ to the body cells, blood takes away CO₂ which is the end product of cellular respiration. Now the blood becomes de-oxygenated.

Since haemoglobin pigment has less affinity for CO₂, CO₂ is mainly transported in the dissolved form. This de-oxygenated blood gives CO₂ to lung alveoli and takes O₂ in return.



Transportation of O₂ and CO₂ in blood

Question 4:

How are the lungs designed in human beings to maximize the area for exchange of gases?

Answer

The exchange of gases takes place between the blood of the capillaries that surround the alveoli and the gases present in the alveoli. Thus, alveoli are the site for exchange of gases. The lungs get filled up with air during the process of inhalation as ribs are lifted up and diaphragm is flattened. The air that is rushed inside the lungs fills the numerous alveoli present in the lungs.

Each lung contains 300-350 million alveoli. These numerous alveoli increase the surface area for gaseous exchange making the process of respiration more efficient.

=

Question 1:

What are the components of the transport system in human beings? What are the functions of these components?

Answer

The main components of the transport system in human beings are the heart, blood, and blood vessels.

- Heart pumps oxygenated blood throughout the body. It receives deoxygenated blood from the various body parts and sends this impure blood to the lungs for oxygenation.
- Being a fluid connective tissue, blood helps in the transport of oxygen, nutrients, CO₂, and nitrogenous wastes.
- The blood vessels (arteries, veins, and capillaries) carry blood either away from the heart to various organs or from various organs back to the heart.

Question 2:

Why is it necessary to separate oxygenated and deoxygenated blood in mammals and birds?

Answer

Warm-blooded animals such as birds and mammals maintain a constant body temperature by cooling themselves when they are in a hotter environment and by warming their bodies when they are in a cooler environment. Hence, these animals require more oxygen (O₂) for more cellular respiration so that they can produce more energy to maintain their body temperature. Thus, it is necessary for them to separate oxygenated and de-oxygenated blood, so that their circulatory system is more efficient and can maintain their constant body temperature.

Question 3:

What are the components of the transport system in highly organised plants?

Answer

In highly organised plants, there are two different types of conducting tissues – xylem and phloem. Xylem conducts water and minerals obtained from the soil (via roots) to the rest of the plant. Phloem transports food materials from the leaves to different parts of the plant body.

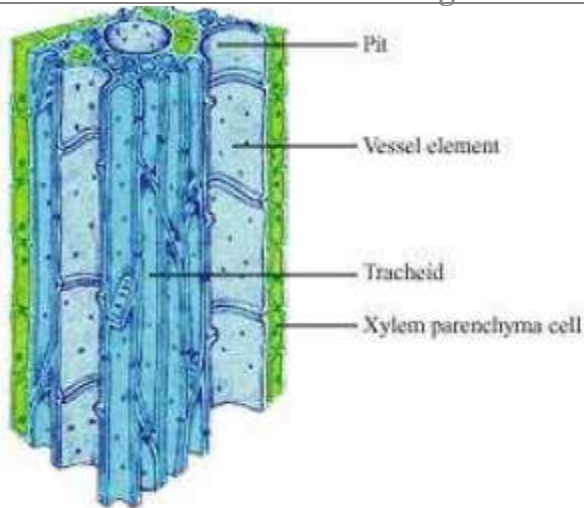
Question 4:

How are water and minerals transported in plants?

Answer

The components of xylem tissue (tracheids and vessels) of roots, stems, and leaves are interconnected to form a continuous system of water-conducting channels that reaches all parts of the plant. Transpiration creates a suction pressure, as a result of which water is forced into the xylem cells of the roots. Then there is a steady movement of water from the root xylem to all the plant parts through the

interconnected water-conducting channels.



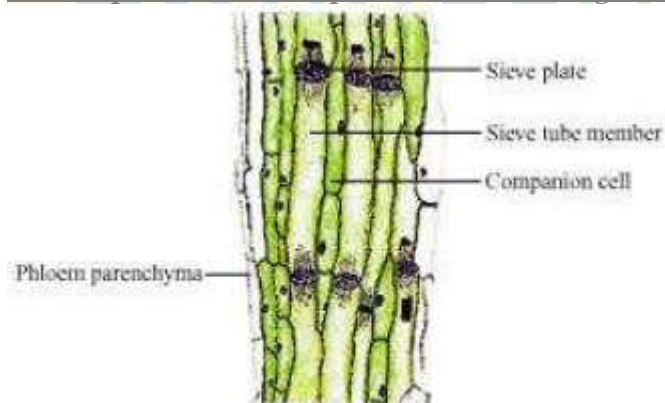
Question 5:

How is food transported in plants?

Answer

Phloem transports food materials from the leaves to different parts of the plant body.

The transportation of food in phloem is achieved by utilizing energy from ATP. As a result of this, the osmotic pressure in the tissue increases causing water to move into it. This pressure moves the material in the phloem to the tissues which have less pressure. This is helpful in moving materials according to the needs of the plant. For example, the food material, such as sucrose, is transported into the phloem tissue using ATP energy.



Components of phloem tissue

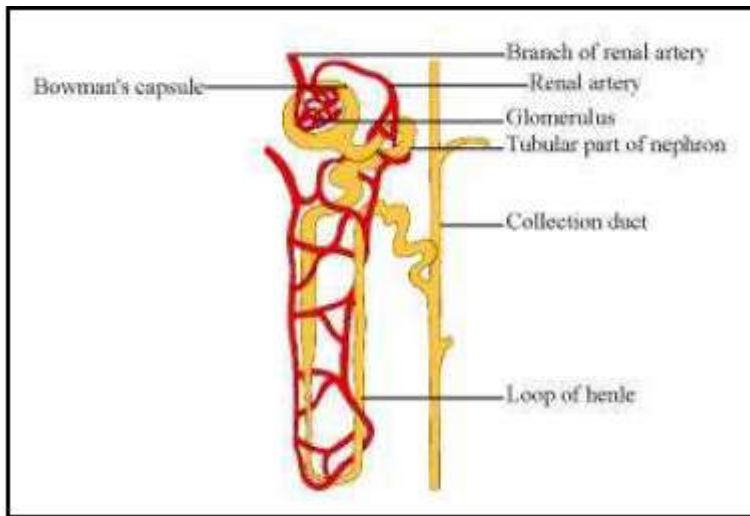
=

Question 1:

Describe the structure and functioning of nephrons.

Answer

Nephrons are the basic filtering units of kidneys. Each kidney possesses large number of nephrons, approximately 1-1.5 million. The main components of the nephron are glomerulus, Bowman's capsule, and a long renal tubule.



Structure of a nephron

Functioning of a nephron:

- The blood enters the kidney through the renal artery, which branches into many capillaries associated with glomerulus.
- The water and solute are transferred to the nephron at Bowman's capsule.
- In the proximal tubule, some substances such as amino acids, glucose, and salts are selectively reabsorbed and unwanted molecules are added in the urine.
- The filtrate then moves down into the loop of Henle, where more water is absorbed.
- From here, the filtrate moves upwards into the distal tubule and finally to the collecting duct. Collecting duct collects urine from many nephrons.
- The urine formed in each kidney enters a long tube called ureter. From ureter, it gets transported to the urinary bladder and then into the urethra.

Question 2:

What are the methods used by plants to get rid of excretory products?

Answer

Plants can get rid of excess of water by transpiration. Waste materials may be stored in the cell vacuoles or as gum and resin, especially in old xylem. It is also stored in the leaves that later fall off.

Question 3:

How is the amount of urine produced regulated?

Answer

The amount of urine produced depends on the amount of excess water and dissolved wastes present in the body. Some other factors such as habitat of an organism and hormone such as Antidiuretic hormone (ADH) also regulates the amount of urine produced.

=

Question 1:

The kidneys in human beings are a part of the system for

- nutrition.
- respiration.
- excretion.
- transportation.

Answer

(c) In human beings, the kidneys are a part of the system for excretion.

Question 2:

The xylem in plants are responsible for

- (a) transport of water.
- (b) transport of food.
- (c) transport of amino acids.
- (d) transport of oxygen.

Answer

(a) In a plant, the xylem is responsible for transport of water.

Question 3:

The autotrophic mode of nutrition requires

- (a) carbon dioxide and water.
- (b) chlorophyll.
- (c) sunlight.
- (d) all of the above.

Answer

(d) The autotrophic mode of nutrition requires carbon dioxide, water, chlorophyll and sunlight.

Question 4:

The breakdown of pyruvate to give carbon dioxide, water and energy takes place in

- (a) cytoplasm.
- (b) mitochondria.
- (c) chloroplast.
- (d) nucleus.

Answer

(b) The breakdown of pyruvate to give carbon dioxide, water and energy takes place in mitochondria.

Question 5:

How are fats digested in our bodies? Where does this process take place?

Answer

Fats are present in the form of large globules in the small intestine. The small intestine gets the secretions in the form of bile juice and pancreatic juice respectively from the liver and the pancreas. The bile salts (from the liver) break down the large fat globules into smaller globules so that the pancreatic enzymes can easily act on them. This is referred to as emulsification of fats. It takes place in the small intestine.

Question 6:

What is the role of saliva in the digestion of food?

Answer

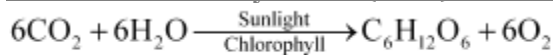
Saliva is secreted by the salivary glands, located under the tongue. It moistens the food for easy swallowing. It contains a digestive enzyme called salivary amylase, which breaks down starch into sugar.

Question 7:

What are the necessary conditions for autotrophic nutrition and what are its by-products?

Answer

Autotrophic nutrition takes place through the process of photosynthesis. Carbon dioxide, water, chlorophyll pigment, and sunlight are the necessary conditions required for autotrophic nutrition. Carbohydrates (food) and O₂ are the by-products of photosynthesis.



Question 8:

What are the differences between aerobic and anaerobic respiration? Name some organisms that use the anaerobic mode of respiration.

Answer

	Aerobic respiration		Anaerobic respiration
1.	It occurs in the presence of O ₂ .	1.	It occurs in the absence of O ₂ .
2.	It involves the exchange of gases between the organism and the outside environment.	2.	Exchange of gases is absent.
3.	It occurs in cytoplasm and mitochondria.	3.	It occurs only in cytoplasm.
4.	It always releases CO ₂ and H ₂ O.	4.	End products vary.
5.	It yields 36 ATPs.	5.	It yields only 2 ATPs.

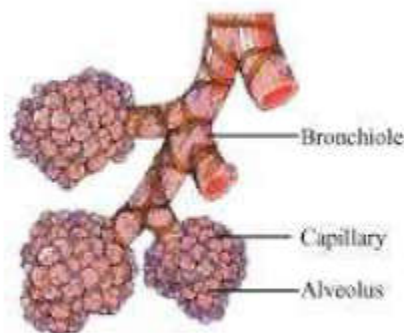
Anaerobic respiration occurs in the roots of some waterlogged plants, some parasitic worms, animal muscles, and some micro-organisms such as yeasts.

Question 9:

How are the alveoli designed to maximise the exchange of gases?

Answer

The alveoli are the small balloon-like structures present in the lungs. The walls of the alveoli consist of extensive network of blood vessels. Each lung contains 300–350 million alveoli, making it a total of approximately 700 million in both the lungs. The alveolar surface when spread out covers about 80 m² area. This large surface area makes the gaseous exchange more efficient.



Alveoli and capillaries

Question 10:

What would be the consequences of a deficiency of haemoglobin in our bodies?

Answer

Haemoglobin is the respiratory pigment that transports oxygen to the body cells for cellular respiration. Therefore, deficiency of haemoglobin in blood can affect the oxygen supplying capacity of blood. This can lead to deficiency of oxygen in the body cells. It can also lead to a disease called anaemia.

Question 11:

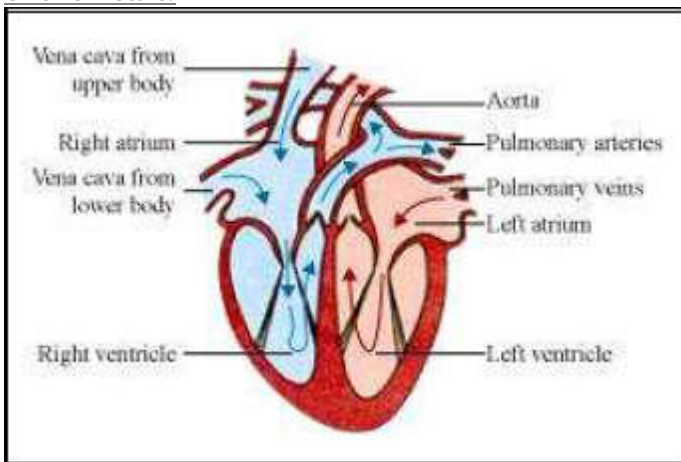
Describe double circulation in human beings. Why is it necessary?

Answer

The human heart is divided into four chambers – the right atrium, the right ventricle, the left atrium, and the left ventricle.

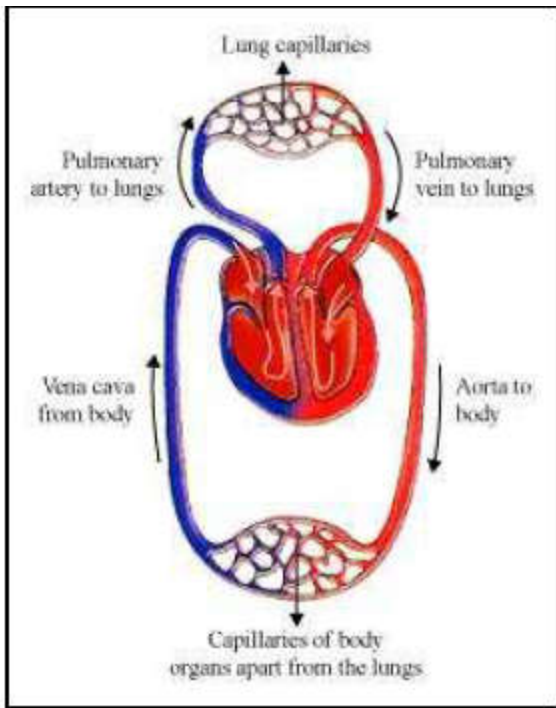
Flow of blood in the heart:

– The heart has superior and inferior vena cava, which carries de-oxygenated blood from the upper and lower regions of the body respectively and supplies this de-oxygenated blood to the right atrium of the heart.



Flow of blood in the human heart

- The right atrium then contracts and passes the de-oxygenated blood to the right ventricle, through an auriculo-ventricular aperture.
- Then the right ventricle contracts and passes the de-oxygenated blood into the two pulmonary arteries, which pumps it to the lungs where the blood becomes oxygenated.
- From the lungs, the pulmonary veins transport the oxygenated blood to the left atrium of the heart.
- Then the left atrium contracts and through the auriculo-ventricular aperture, the oxygenated blood enters the left ventricle.
- The blood passes to aorta from the left ventricle. The aorta gives rise to many arteries that distribute the oxygenated blood to all the regions of the body.



Schematic diagram of blood circulation in humans. Therefore, the blood goes twice through the heart. This is known as double circulation.

Importance of double circulation:

The separation of oxygenated and de-oxygenated blood allows a more efficient supply of oxygen to the body cells. This efficient system of oxygen supply is very useful in warm-blooded animals such as human beings.

As we know, warm-blooded animals have to maintain a constant body temperature by cooling themselves when they are in a hotter environment and by warming their bodies when they are in a cooler environment. Hence, they require more O₂ for more respiration so that they can produce more energy to maintain their body temperature. Thus, the circulatory system of humans is more efficient because of the double circulatory heart.

Question 12:

What are the differences between the transport of materials in xylem and phloem?

Answer

	Transport of materials in xylem		Transport of materials in phloem
(i)	Xylem tissue helps in the transport of water and minerals.	(i)	Phloem tissue helps in the transport of food.
(ii)	Water is transported upwards from roots to all other plant parts.	(ii)	Food is transported in both upward and downward directions.
(iii)	Transport in xylem occurs with the help of simple physical forces such as transpiration pull.	(iii)	Transport of food in phloem requires energy in the form of ATP.

Question 13:

Compare the functioning of alveoli in the lungs and nephrons in the kidneys with respect to their structure and functioning.

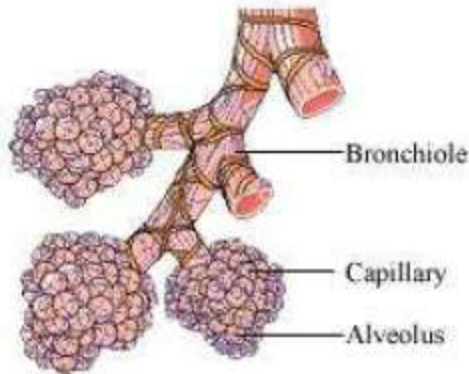


Free Home Delivery
WE ARE WITH YOU.....

Alveoli

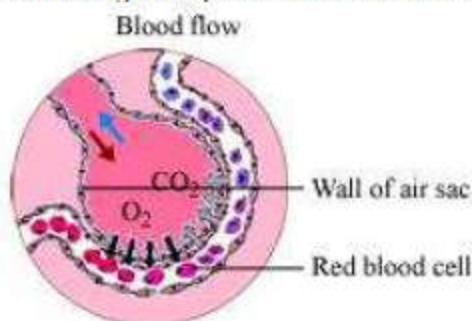
Structure

- (i) Alveoli are tiny balloon-like structures present inside the lungs.
- (ii) The walls of the alveoli are one cell thick and it contains an extensive network of blood capillaries.



Function

- (i) The exchange of O_2 and CO_2 takes place between the blood of the capillaries that surround the alveoli and the gases present in the alveoli.

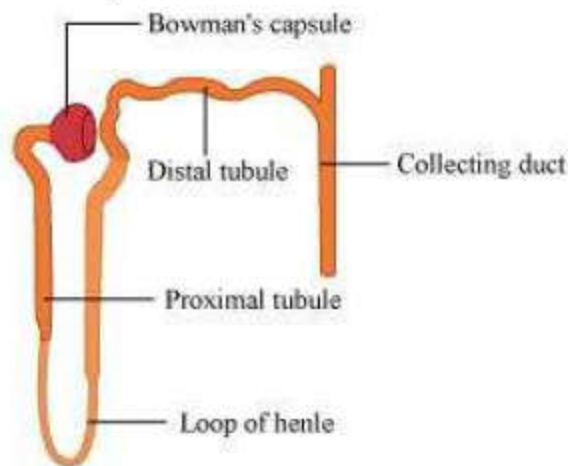


- (ii) Alveoli are the site of gaseous exchange.

Nephron

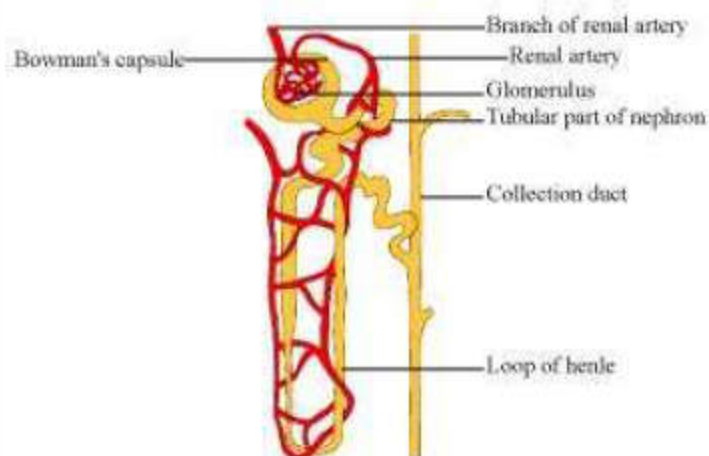
Structure

- (i) Nephrons are tubular structures present inside the kidneys.
- (ii) Nephrons are made of glomerulus, bowman's capsule, and a long renal tube. It also contains a cluster of thin-walled capillaries.



Function

- (i) The blood enters the kidneys through the renal artery which branches into many capillaries in the glomerulus. The water and solute are transferred to the nephron at Bowman's capsule. Then the filtrate moves through the proximal tubule and then down into the loop of henle. From henle's loop, filtrate passes into the distal tubule and then to the collecting duct. The collecting duct collects the urine from many nephrons and passes it to the ureter. During the flow of filtrate, some substances such as glucose, amino acids, and water are selectively re-absorbed.



- (ii) Nephrons are the basic filtration unit.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 7 Control and Coordination Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 7 Control and Coordination Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 7 Control and Coordination NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

What is the difference between a reflex action and walking?

Answer:

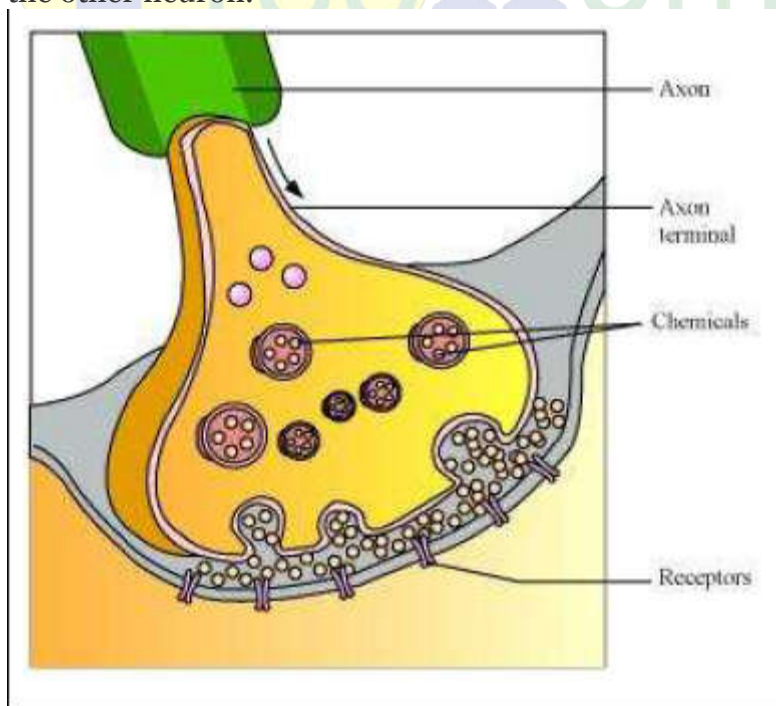
A reflex action is a rapid, automatic response to a stimulus. It does not involve any thinking. For example, we close our eyes immediately when the bright light is focused. Walking, on the other hand, is a voluntary action. It is under our conscious control.

Question 2:

What happens at the synapse between two neurons?

Answer:

A very small gap that occurs between the last portion of axon of one neuron and the dendron of the other neuron is known as a synapse. It acts as a one way valve to transmit impulses in one direction only. This uni-direction transfer of impulses occurs as the chemicals are produced in only one side of the neuron i.e., the axon's side. From axon, the impulses travel across the synapse to the dendron of the other neuron.



WE ARE WITH YOU.....

Question 3:

Which part of the brain maintains posture and equilibrium of the body?

Answer:

Cerebellum, a part of hindbrain is responsible for maintaining posture and equilibrium of the body.

Question 4:

How do we detect the smell of an agarbatti (incense stick)?

Answer:

The thinking part of our brain is the forebrain. It has separate areas that are specialized for hearing, smelling, sight, taste, touch, etc. The forebrain also has regions that collect information or impulses from the various receptors. When the smell of an incense stick reaches us, our forebrain detects it. Then, the forebrain interprets it by putting it together with the information received from other receptors and also with the information already stored in the brain.

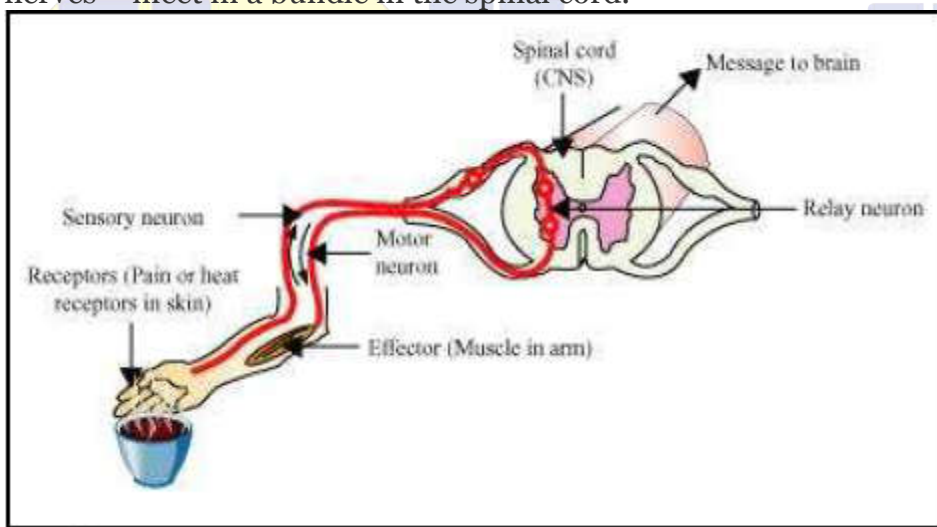
Question 5:

What is the role of the brain in reflex action?

Answer:

Reflex actions are sudden responses, which do not involve any thinking. For example, when we touch a hot object, we withdraw our hand immediately without thinking as thinking may take time which would be enough to get us burnt.

The sensory nerves that detect the heat are connected to the nerves that move the muscles of the hand. Such a connection of detecting the signal from the nerves (input) and responding to it quickly (output) is called a reflex arc. The reflex arcs –connections present between the input and output nerves – meet in a bundle in the spinal cord.

**Reflex arc**

Reflex arcs are formed in the spinal cord and the information (input) reaches the brain. The brain is only aware of the signal and the response that has taken place. However, the brain has no role to play in the creation of the response.

Question 1:

What are plant hormones?

Answer:

Plant hormones or phytohormones are naturally-occurring organic substances. These are synthesized in one part of the plant body (in minute quantities) and are translocated to other parts when required. The five major types of phytohormones are auxins, gibberellins, cytokinins, abscisic acid, and ethylene.

Question 2:

How is the movement of leaves of the sensitive plant different from the movement of a shoot towards light?

Answer:

The movement of leaves of the sensitive plant, Mimosa pudica or “touch me not”, occurs in response to touch or contact stimuli. This movement is independent of growth. The movement of shoot towards light is known as phototropism. This type of movement is directional and is growth dependent.

Question 3:

Give an example of a plant hormone that promotes growth.

Answer:

Auxin is an example of growth-promoting plant hormone.

Question 4:

How do auxins promote the growth of a tendril around a support?

Answer:

Auxin is synthesized at the shoot tip. It helps the cell grow longer. When a tendril comes in contact with a support, auxin stimulates faster growth of the cells on the opposite side, so that the tendril forms a coil around the support. This makes the tendrils appear as a watch spring.

Question 5:

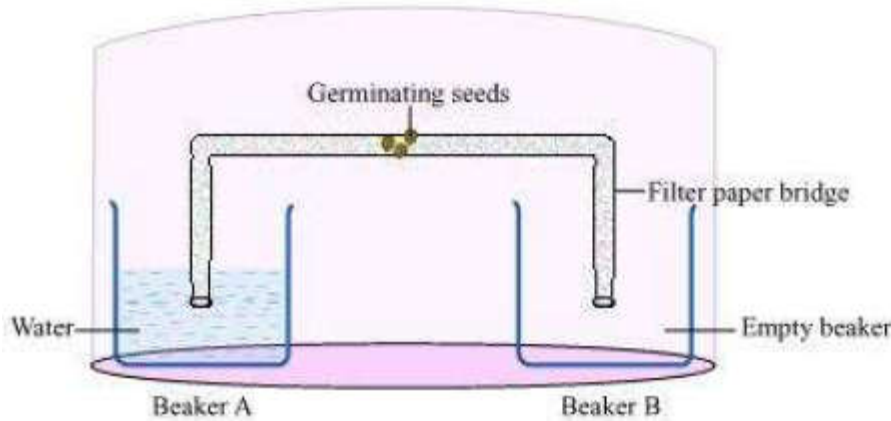
Design an experiment to demonstrate hydrotropism.

Answer:

Take two small beakers and label them as A and B. Fill beaker A with water. Now make a cylindrical-shaped roll from a filter paper and keep it as a bridge between beaker A and beaker B, as shown in the figure. Attach few germinating seeds in the middle of the filter paper bridge. Now, cover the entire

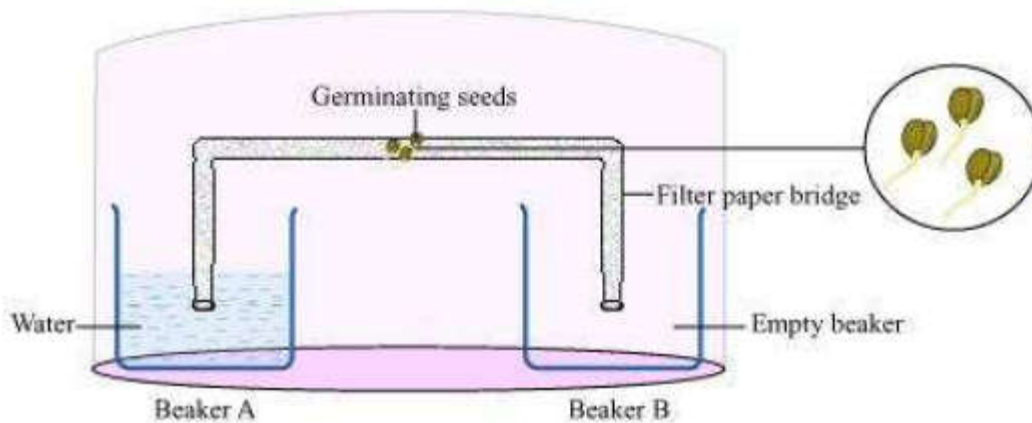
WE ARE WITH YOU.....

set-up with a transparent plastic container so that the moisture is retained.



Observation:

The roots of the germinating seeds will grow towards beaker A.



This experiment demonstrates the phenomenon of hydrotropism.

delivery
WITH YOU.....

==

Question 1:

How does chemical coordination take place in animals?

Answer:

Chemical coordination takes place in animals with the help of hormones. Hormone is the chemical messenger that regulates the physiological processes in living organisms. It is secreted by glands. The regulation of physiological processes, and control and coordination by hormones comes under the endocrine system. The nervous system along with the endocrine system in our body controls and coordinates the physiological processes.

Question 2:

Why is the use of iodised salt advisable?

Answer:

Iodine stimulates the thyroid gland to produce thyroxine hormone. It regulates carbohydrate, fat, and protein metabolism in our body. Deficiency of this hormone results in the enlargement of the thyroid

gland. This can lead to goitre, a disease characterized by swollen neck. Therefore, iodised salt is advised for normal functioning of the thyroid gland.

Question 3:

How does our body respond when adrenaline is secreted into the blood?

Answer:

Adrenalin is a hormone secreted by the adrenal glands in case of any danger or emergency or any kinds of stress. It is secreted directly into the blood and is transported to different parts of the body. When secreted in large amounts, it speeds up the heartbeat and hence supplies more oxygen to the muscles. The breathing rate also increases due to contractions of diaphragm and rib muscles. It also increases the blood pressure. All these responses enable the body to deal with any stress or emergency.

Question 4:

Why are some patients of diabetes treated by giving injections of insulin?

Answer:

Diabetes is a disease in which the level of sugar in the blood is too high. Insulin, a hormone secreted by the pancreas, helps in regulating the blood sugar levels. This is the reason why diabetic patients are treated by giving injections of insulin.

=

Question 1:

Which of the following is a plant hormone?

- (a) Insulin
- (b) Thyroxin
- (c) Oestrogen
- (d) Cytokinin

Answer:

(d) Cytokinin is a plant hormone.

Question 2:

The gap between two neurons is called a

- (a) dendrite.
- (b) synapse.
- (c) axon.
- (d) impulse.

Answer:

(b) The gap between two neurons is called a synapse.

Question 3:

The brain is responsible for

- (a) thinking.
- (b) regulating the heart beat.
- (c) balancing the body.
- (d) all of the above.

Answer:

(d) The brain is responsible for thinking, regulating the heart beat and balancing the body.



Question 4:

What is the function of receptors in our body? Think of situations where receptors do not work properly. What problems are likely to arise?

Answer:

Receptors are sensory structures (organs/tissues or cells) present all over the body. The receptors are either grouped in case of eye or ear, or scattered in case of skin.

Functions of receptors:

(i) They sense the external stimuli such as heat or pain.

(ii) They also trigger an impulse in the sensory neuron which sends message to the spinal cord.

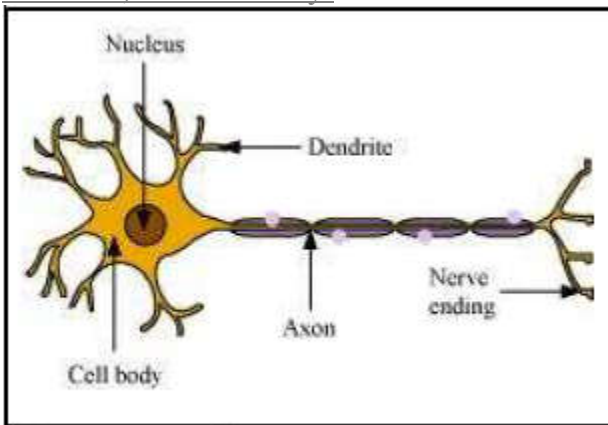
When the receptors are damaged, the external stimuli transferring signals to the brain are not felt. For example, in the case of damaged receptors, if we accidentally touch any hot object, then our hands might get burnt as damaged receptors cannot perceive the external stimuli of heat and pain.

Question 5:

Draw the structure of a neuron and explain its function.

Answer:

Neurons are the functional units of the nervous system. The three main parts of a neuron are axon, dendrite, and cell body.

**Structure of a neuron**

Functions of the three parts of a neuron:

Axon: It conducts messages away from the cell body.

Dendrite: It receives information from axon of another cell and conducts the messages towards the cell body.

Cell body: It contains nucleus, mitochondria, and other organelles. It is mainly concerned with the maintenance and growth.

Question 6:

What are the changes seen in girls at the time of puberty?

Answer:

Secondary sexual characteristics in girls:

- Increase in breast size and darkening of skin of the nipples present at the tips of the breasts.
- Appearance of hair in the genital area.
- Appearance of hair in other areas of skin like underarms, face, hands, and legs.
- Increase in the size of uterus and ovary.
- Beginning of menstrual cycle.
- More secretion of oil from the skin, which results in the appearance of pimples.

meDelivery

WE ARE WITH YOU.....

Question 7:

Which signals will get disrupted in case of a spinal cord injury?

Answer:

The reflex arc connections between the input and output nerves meet in a bundle in the spinal cord. In fact, nerves from all over the body meet in a bundle in the spinal cord on their way to the brain. In case of any injury to the spinal cord, the signals coming from the nerves as well as the signals coming to the receptors will be disrupted.

Question 8:

How does chemical coordination occur in plants?

Answer:

In animals, control and coordination occur with the help of nervous system. However, plants do not have a nervous system.

Plants respond to stimuli by showing movements. The growth, development, and responses to the environment in plants is controlled and coordinated by a special class of chemical substances known as hormones. These hormones are produced in one part of the plant body and are translocated to other needy parts. For example, a hormone produced in roots is translocated to other parts when required. The five major types of phytohormone are auxins, gibberellins, cytokinins, abscisic acid, and ethylene. These phytohormones are either growth promoters (such as auxins, gibberellins, cytokinins, and ethylene) or growth inhibitors such as abscisic acid.

Question 9:

What is the need for a system of control and coordination in an organism?

Answer:

The maintenance of the body functions in response to changes in the body by working together of various integrated body systems is known as coordination. All the movements that occur in response to stimuli are carefully coordinated and controlled. In animals, the control and coordination movements are provided by nervous and muscular systems. The nervous system sends messages to and away from the brain. The spinal cord plays an important role in the relay of messages. In the absence of this system of control and coordination, our body will not be able to function properly. For example, when we accidentally touch a hot utensil, we immediately withdraw our hand. In the absence of nerve transmission, we will not withdraw our hand and may get burnt.

Question 10:

How are involuntary actions and reflex actions different from each other?

Answer:

Involuntary actions cannot be consciously controlled. For example, we cannot consciously control the movement of food in the alimentary canal. These actions are however directly under the control of the brain. On the other hand, the reflex actions such as closing of eyes immediately when bright light is focused show sudden response and do not involve any thinking. This means that unlike involuntary actions, the reflex actions are not under the control of brain.

Question 11:

Compare and contrast nervous and hormonal mechanisms for control and coordination in animals.

Answer:

	Nervous system mechanism		Hormonal system mechanism
1.	The information is conveyed in the form of electric impulse.	1.	The information is conveyed in the form of chemical messengers.
2.	The axons and dendrites transmit the information through a coordinated effort.	2.	The information is transmitted or transported through blood.
3.	The flow of information is rapid and the response is quick.	3.	The information travels slowly and the response is slow.
4.	Its effects are short lived.	4.	It has prolonged effects.

Question 12:

What is the difference between the manner in which movement takes place in a sensitive plant and the movement in our legs?

Answer:

	Movement in sensitive plants		Movement in our legs
1.	The movement that takes place in a sensitive plant such as <i>Mimosa pudica</i> occurs in response to touch (stimulus).	1.	Movement in our legs is an example of voluntary actions.
2.	For this movement, the information is transmitted from cell to cell by electro-chemical signals as plants do not have any specialised tissue for conduction of impulses.	2.	The signal or messages for these actions are passed to the brain and hence are consciously controlled.
3.	For this movement to occur, the plant cells change shape by changing the amount of water in them.	3.	In animal muscle cells, some proteins are found which allow the movement to occur.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 8 How do Organisms Reproduce Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 8 How do Organisms Reproduce Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 8 How do Organisms Reproduce NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

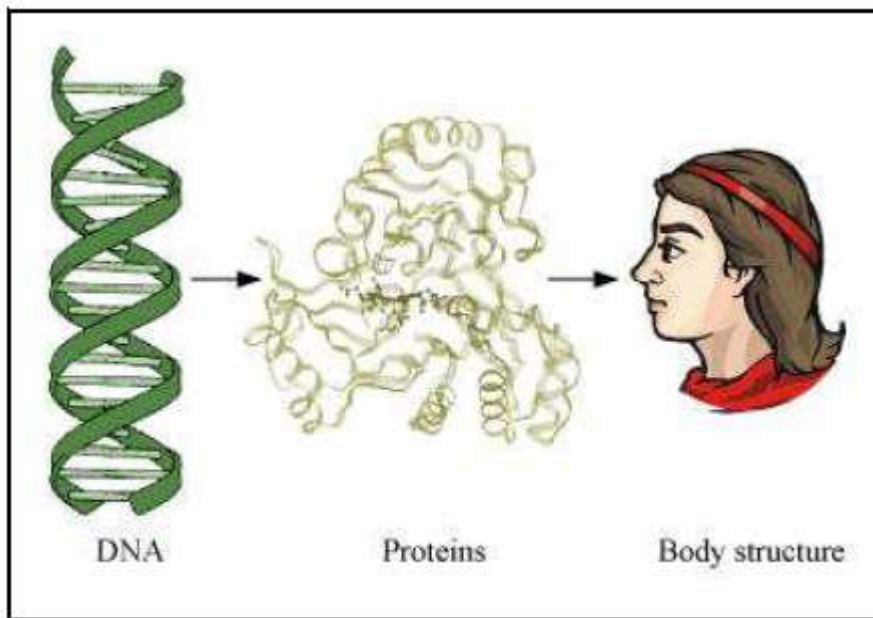
Question 1:

What is the importance of DNA copying in reproduction?

Answer:

DNA (Deoxyribonucleic acid) is the genetic material found in the chromosomes, which are present in the nucleus of a cell. The DNA is the information site for making proteins and each specific type of protein leads to a specific type of body design.

Thus, it is the DNA molecule that determines the body design of an individual. Therefore, it can be concluded that it is the DNA that gets transferred from parents to offsprings and makes them look similar.



DNA determines body structure

Question 2:

Why is variation beneficial to the species but not necessarily for the individual?

Answer:

Variations are beneficial to the species than individual because sometimes for a species, the environmental conditions change so drastically that their survival becomes difficult. For example, if the temperature of water increases suddenly, then most of the bacteria living in that water would die. Only few variants that are resistant to heat would be able to survive. However, if these variants were not there, then the entire species of bacteria would have been destroyed. Thus, these variants help in

the survival of the species. However, all variations are not necessarily beneficial for the individual organisms.

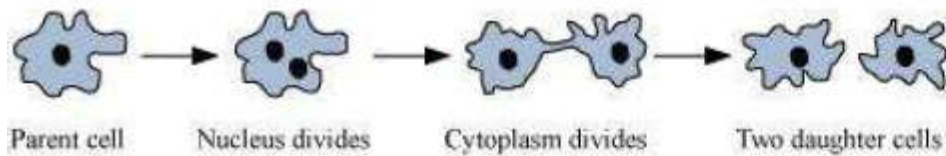
—

Question 1:

How does binary fission differ from multiple fission?

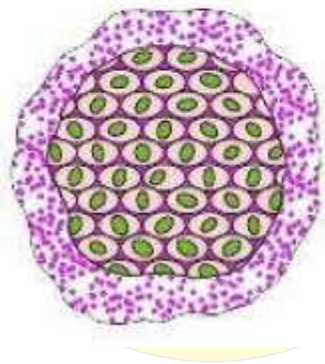
Answer:

In binary fission, a single cell divides into two equal halves. Amoeba and Bacteria divide by binary fission.



Binary fission in *Amoeba*

In multiple fission, a single cell divides into many daughter cells simultaneously. *Amoeba* and *Plasmodium* divide by multiple fission.



WE ARE WITH YOU.....

Question 2:

How will an organism be benefited if it reproduces through spores?

Answer:

There are many advantages, if an organism reproduces through spores.

Advantages of spore formation:

- Large numbers of spores are produced in one sporangium.
- Spores are distributed easily by air to far-off places to avoid competition at one place.
- Spores are covered by thick walls to prevent dehydration under unfavourable conditions.

Question 3:

Can you think of reasons why more complex organisms cannot give rise to new individuals through regeneration?

Answer:

Simple organisms such as Hydra and Planaria are capable of producing new individuals through the process of regeneration. The process of regeneration involves the formation of new organisms from its body parts. Simple organisms can utilize this method of reproduction as their entire body is made of similar kind of cells in which any part of their body can be formed by growth and development. However, complex organisms have organ-system level of organization. All the organ systems of their body work together as an interconnected unit. They can regenerate their lost

body parts such as skin, muscles, blood, etc. However, they cannot give rise to new individuals through regeneration.

Question 4:

Which of the following is a plant hormone?

- (a) Insulin
- (b) Thyroxin
- (c) Oestrogen
- (d) Cytokinin

Answer:

(d) Cytokinin is a plant hormone.

Question 5:

Why is DNA copying an essential part of the process of reproduction?

Answer:

DNA (Deoxyribonucleic acid) copying is an essential part of reproduction as it passes genetic information from parents to offspring. It determines the body design of an individual. The reproducing cells produce a copy of their DNA through some chemical reactions and result in two copies of DNA. The copying of DNA always takes place along with the creation of additional cellular structure. This process is then followed by division of a cell to form two cells.

=

Question 1:

How is the process of pollination different from fertilization?

Answer:

Pollination is the process of transfer of pollens from anther to stigma. It occurs with the help of certain pollinators such as air, water, birds, or some insects.

Fertilization, on the other hand, is the fusion of the male and female gametes. It occurs inside the ovule and leads to the formation of zygote.

Question 2:

What is the role of the seminal vesicles and the prostate gland?

Answer:

The secretions from seminal vesicles and prostate glands lubricate the sperms and provide a fluid medium for easy transport of sperms. Their secretion also provides nutrient in the form of fructose, calcium, and some enzymes.

Question 3:

What are the changes seen in girls at the time of puberty?

Answer:

Secondary sexual characteristics in girls:

- Increase in breast size and darkening of skin of the nipples present at the tips of the breasts.
- Appearance of hair in the genital area.
- Appearance of hair in other areas of skin like underarms, face, hands, and legs.
- Increase in the size of uterus and ovary.
- Beginning of menstrual cycle.
- More secretion of oil from the skin, which results in the appearance of pimples.

Question 4:

How does the embryo get nourishment inside the mother's body?

Answer:

The embryo develops inside the mother's body for about nine months. Inside the uterus, the outer tissue surrounding the embryo develops finger-like projections called villi. These villi are surrounded by uterine tissue and maternal blood. They provide a large surface area for exchange of oxygen and nutrients. Also, there is a special tissue called placenta, which is embedded in the uterine wall. The embryo receives the oxygen and nutrients from the mother's blood via the placenta. The waste materials produced by the embryo are also removed through the placenta.

Question 5:

If a woman is using a copper-T, will it help in protecting her from sexually transmitted diseases?

Answer:

No. Using a copper-T will not provide a protection from sexually transmitted diseases, as it does not prevent the entry of semen. It only prevents the implantation of the embryo in the uterus.

=

Question 1:

Asexual reproduction takes place through budding in

- (a) amoeba.
- (b) yeast.
- (c) plasmodium.
- (d) leishmania.

Answer:

(b) Asexual reproduction takes place through budding in yeast.

Question 2:

Which of the following is not a part of the female reproductive system in human beings?

- (a) Ovary
- (b) Uterus
- (c) Vas deferens
- (d) Fallopian tube

Answer:

(c) Vas deferens is not a part of the female reproductive system in human beings.

Question 3:

The anther contains

- (a) sepals.
- (b) ovules.
- (c) carpel.
- (d) pollen grains.

Answer:

(d) The anther contains pollen grains.

Question 4:

What are the advantages of sexual reproduction over asexual reproduction?

Answer:

Advantages of sexual reproduction:

- In sexual reproduction, more variations are produced. Thus, it ensures survival of species in a population.
- The new formed individual has characteristics of both the parents.
- Variations are more viable in sexual mode than in asexual one. This is because in asexual reproduction, DNA has to function inside the inherited cellular apparatus.

Question 5:

What are the functions performed by the testis in human beings?

Answer:

The testes are the male reproductive organs that are located outside the abdominal cavity within a pouch called scrotum.

Functions of testes:

- Produce sperms
- Produce a hormone called testosterone, which brings about secondary sexual characters in boys.

Question 6:

Why does menstruation occur?

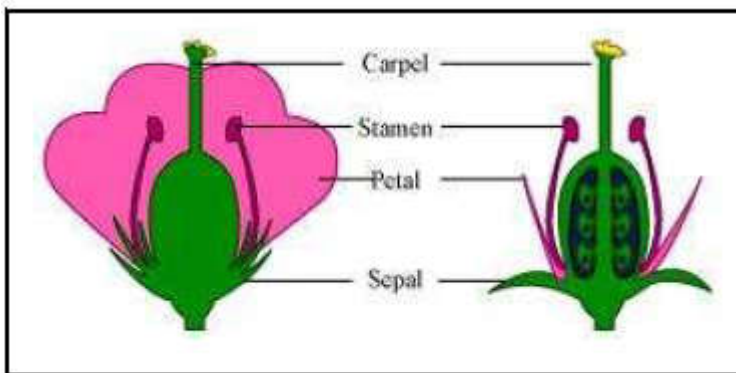
Answer:

Menstruation is a process in which blood and mucous flows out every month through the vagina. This process occurs every month because one egg is released from the ovary every month and at the same time, the uterus (womb) prepares itself to receive the fertilized egg. Thus, the inner lining of the uterus gets thickened and is supplied with blood to nourish the embryo. If the egg does not get fertilised, then the lining of the uterus breaks down slowly and gets released in the form of blood and mucous from the vagina.

Question 7:

Draw a labelled diagram of the longitudinal section of a flower.

Answer:



Question 8:

What are the different methods of contraception?

Answer:

The contraceptive methods can be broadly divided into the following types:

- Natural method→ It involves avoiding the chances of meeting of sperms and ovum. In this method, the sexual act is avoided from day 10th to 17th of the menstrual cycle because during this period, ovulation is expected and therefore, the chances of fertilization are very high.

- Barrier method → In this method, the fertilization of ovum and sperm is prevented with the help of barriers. Barriers are available for both males and females. Condoms are barriers made of thin rubber that are used to cover penis in males and vagina in females.
- Oral contraceptives → In this method, tablets or drugs are taken orally. These contain small doses of hormones that prevent the release of eggs and thus fertilization cannot occur.
- Implants and surgical methods → Contraceptive devices such as the loop or Copper-T are placed in uterus to prevent pregnancy. Some surgical methods can also be used to block the gamete transfer. It includes the blocking of vas deferens to prevent the transfer of sperms known as vasectomy. Similarly, fallopian tubes of the female can be blocked so that the egg will not reach the uterus known as tubectomy.

Question 9:

How are the modes for reproduction different in unicellular and multicellular organisms?

Answer:

In unicellular organisms, reproduction occurs by the division of the entire cell. The modes of reproduction in unicellular organisms can be fission, budding, etc. whereas in multicellular organisms, specialised reproductive organs are present. Therefore, they can reproduce by complex reproductive methods such as vegetative propagation, spore formation, etc. In more complex multicellular organisms such as human beings and plants, the mode of reproduction is sexual reproduction.

Question 10:

How does reproduction help in providing stability to populations of species?

Answer:

Living organisms reproduce for the continuation of a particular species. It helps in providing stability to the population of species by producing a new individual that resembles the parents. This is the reason why cats give birth to only cats or dogs give birth to only dogs. Therefore, reproduction provides stability to populations of dogs or cats or any other species.

Question 11:

What could be the reasons for adopting contraceptive methods?

Answer:

Contraceptive methods are mainly adopted because of the following reasons:

- To prevent unwanted pregnancies.
- To control population rise or birth rate.
- To prevent the transfer of sexually transmitted diseases.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 9 Heredity and Evolution Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 9 Heredity and Evolution Download 2017 2018 New Edition PDF>

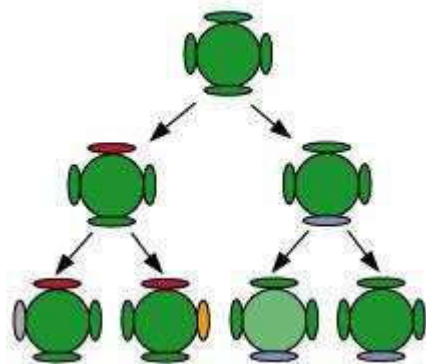
Class 10 Science Chapter 9 Heredity and Evolution NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

If a trait A exists in 10% of a population of an asexually reproducing species and a trait B exists in 60% of the same population, which trait is likely to have arisen earlier?

Answer:

In asexual reproduction, the reproducing cells produce a copy of their DNA through some chemical reactions. However, this copying of DNA is not accurate and therefore, the newly formed DNA has some variations.



It can be easily observed in the above figure that in asexual reproduction, very few variations are allowed. Therefore, if a trait is present in only 10% of the population, it is more likely that the trait has arisen recently. Hence, it can be concluded that trait B that exists in 60% of the same population has arisen earlier than trait A.

Question 2:

How does the creation of variations in a species promote survival?

Answer:

Sometimes for a species, the environmental conditions change so drastically that their survival becomes difficult. For example, if the temperature of water increases suddenly, most of the bacteria living in that water would die. Only few variants resistant to heat would be able to survive. If these variants were not there, then the entire species of bacteria would have been destroyed. Thus, these variants help in the survival of the species.

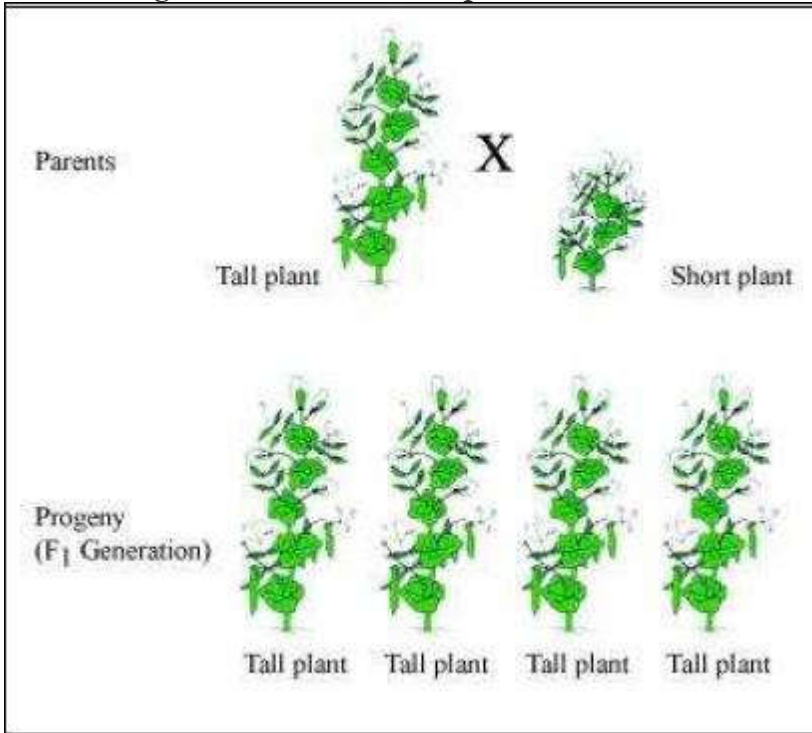
However, not all variations are useful. Therefore, these are not necessarily beneficial for the individual organisms.

Question 1:

How do Mendel's experiments show that traits may be dominant or recessive?

Answer:

Mendel selected true breeding tall (TT) and dwarf (tt) pea plants. Then, he crossed these two plants. The seeds formed after fertilization were grown and these plants that were formed represent the first filial or F₁ generation. All the F₁ plants obtained were tall.



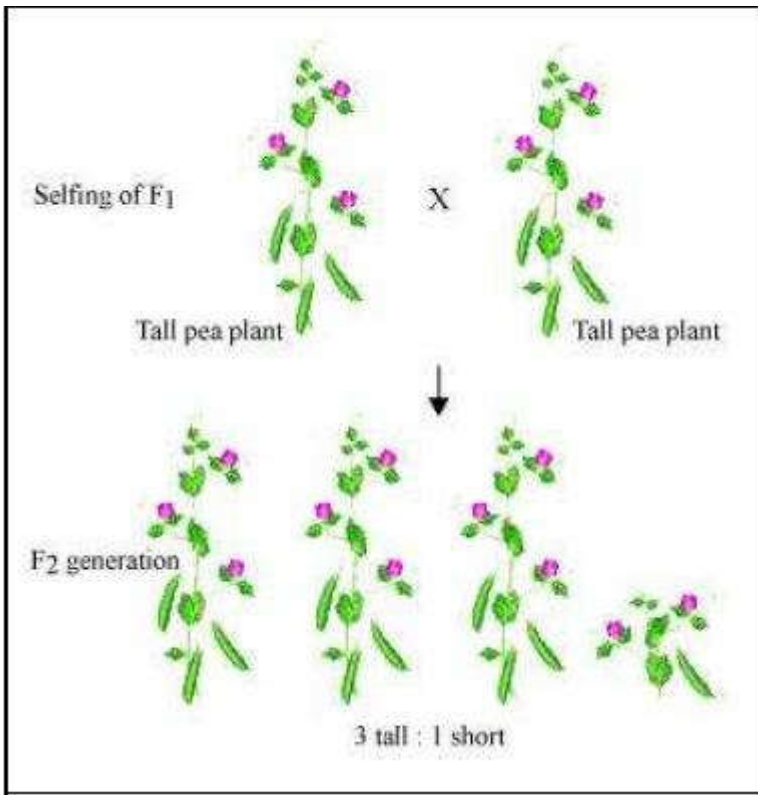
Cross-pollination of tall and short plant

Cross-pollination of tall and short plant

Then, Mendel self-pollinated the F₁ plants and observed that all plants obtained in the F₂ generation were not tall. Instead, one-fourth of the F₂ plants were short.

eDelivery

WE ARE WITH YOU.....



Self-pollination of F₁ plants

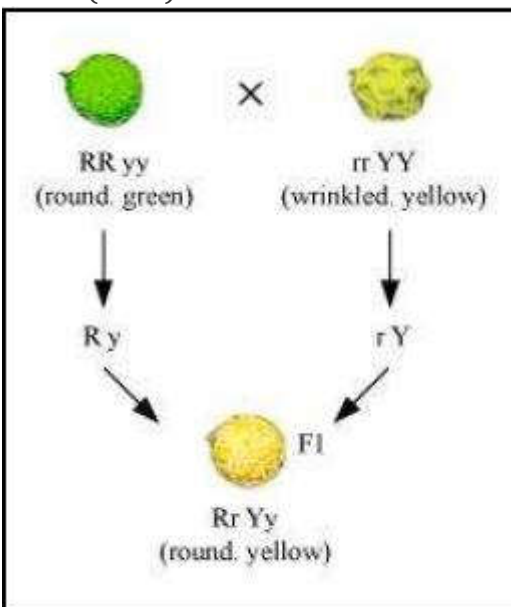
From this experiment, Mendel concluded that the F₁ tall plants were not true breeding. They were carrying traits of both short height and tall height. They appeared tall only because the tall trait is dominant over the dwarf trait.

Question 2:

How do Mendel's experiments show that traits are inherited independently?

Answer:


Mendel crossed pea plants having round green seeds (RRyy) with pea plants having wrinkled yellow seeds (rrYY).



An example of dihybrid crosses

Since the F₁ plants are formed after crossing pea plants having green round seeds and pea plants having yellow wrinkled seeds, F₁ generation will have both these characters in them. However, as we know that yellow seed colour and round seeds are dominant characters, therefore, the F₁ plants will have yellow round seeds.

Then this F₁ progeny was self-pollinated and the F₂ progeny was found to have yellow round seeds, green round seeds, yellow wrinkled seeds, and green wrinkled seeds in the ratio of 9:3:3:1.



Round and yellow (RrYy)

	Pollen grain	RY	Ry	ry	rY
Egg cell	RY	RRYY	RRYy	RrYY	RrYY
	Ry	RRYy	RRyy	Rryy	RrYy
	ry	RrYY	Rryy	rryy	rrYy
	rY	RrYY	RrYy	rrYy	rrYY

Independent inheritance of two different traits

In the above cross, more than two factors are involved, and these are independently inherited.

Question 3:

A man with blood group A marries a woman with blood group O and their daughter has blood group O. Is this information enough to tell you which of the traits – blood group A or O – is dominant? Why or why not?

Answer:

No. This information is not sufficient to determine which of the traits – blood group A or O – is dominant. This is because we do not know about the blood group of all the progeny.

Blood group A can be genotypically AA or AO. Hence, the information is incomplete to draw any such conclusion.

Question 4:

How is the sex of the child determined in human beings?

Answer:

In human beings, the females have two X chromosomes and the males have one X and one Y chromosome. Therefore, the females are XX and the males are XY.

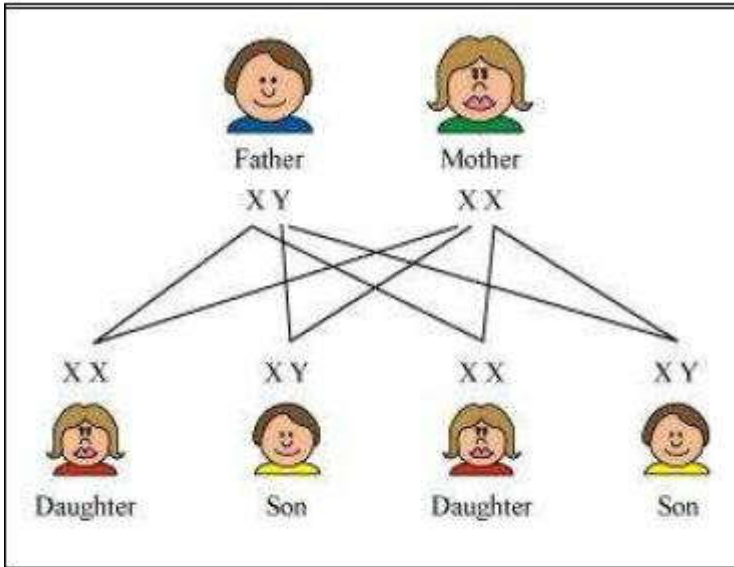
The gametes, as we know, receive half of the chromosomes. The male gametes have 22 autosomes and either X or Y sex chromosome.

Type of male gametes: 22+X OR 22+ Y.

However, since the females have XX sex chromosomes, their gametes can only have X sex

chromosome.

Type of female gamete: 22+X



Sex determination in humans

Thus, the mother provides only X chromosomes. The sex of the baby is determined by the type of male gamete (X or Y) that fuses with the X chromosome of the female.

=

Question 1:

What are the different ways in which individuals with a particular trait may increase in a population?

Answer:

Individuals with a particular trait may increase in a population as a result of the following:

- (i) Natural selection: When that trait offers some survival advantage.
- (ii) Genetic drift: When some genes governing that trait become common in a population.
- (iii) When that trait gets acquired during the individual's lifetime.

Question 2:

Why are traits acquired during the life-time of an individual not inherited?

Answer:

This happens because an acquired trait involves change in non-reproductive tissues (somatic cells) which cannot be passed on to germ cells or the progeny. Therefore, these traits cannot be inherited.

Question 3:

Why are the small numbers of surviving tigers a cause of worry from the point of view of genetics?

Answer:

Small numbers of tigers means that fewer variations in terms of genes are available. This means that when these tigers reproduce, there are less chances of producing progeny with some useful variations. Hence, it is a cause of worry from the point of view of genetics.

=

Question 1:

What factors could lead to the rise of a new species?

Answer:

Natural selection, genetic drift and acquisition of traits during the life time of an individual can give rise to new species.

Question 2:

Will geographical isolation be a major factor in the speciation of a self-pollinating plant species? Why or why not?

Answer:

Geographical isolation can prevent the transfer of pollens among different plants. However, since the plants are self-pollinating, which means that the pollens are transferred from the anther of one flower to the stigma of the same flower or of another flower of the same plant, geographical isolation cannot prevent speciation in this case.

Question 3:

Will geographical isolation be a major factor in the speciation of an organism that reproduces asexually? Why or why not?

Answer:

Geographical isolation prevents gene flow between populations of a species whereas asexual reproduction generally involves only one individual. In an asexually reproducing organism, variations can occur only when the copying of DNA is not accurate. Therefore, geographical isolation cannot prevent the formation of new species in an asexually reproducing organism.

=

Question 1:

Give an example of characteristics being used to determine how close two species are in evolutionary terms.

Answer:

The presence of feathers in dinosaurs and birds indicates that they are evolutionarily related. Dinosaurs had feathers not for flying but instead these feathers provided insulation to these warm-blooded animals. However, the feathers in birds are used for flight. This proves that reptiles and birds are closely related and that the evolution of wings started in reptiles.

Question 3:

Find out, from Table, the medium having highest optical density. Also find the medium with lowest

optical density.

Material medium	Refractive index	Material medium	Refractive index
Air	1.0003	Canada Balsam	1.53
Ice	1.31	-	-
Water	1.33	Rock salt	1.54
Alcohol	1.36	-	-
Kerosene	1.44	Carbon disulphide	1.63
Fused quartz	1.46	Dense flint glass	1.65
Turpentine oil	1.47	Ruby	1.71
Benzene	1.50	Sapphire	1.77
Crown glass	1.52	Diamond	2.42

Question 3:

What are fossils? What do they tell us about the process of evolution?

Answer:

Fossils are the remains of organisms that once existed on earth. They represent the ancestors of plants and animals that are alive today. They provide evidences of evolution by revealing the characteristics of the past organism and the changes that have occurred in these organisms to give rise to the present organisms.

=

Question 1:

Why are human beings who look so different from each other in terms of size, colour and looks said to belong to the same species?

Answer:

A species is a group of organisms that are capable of interbreeding to produce a fertile offspring. Skin colour, looks, and size are all variety of features present in human beings. These features are generally environmentally controlled. Various human races are formed based on these features. However, there

is no biological basis to this concept of races. Therefore, all human beings are a single species as humans of different colour, size, and looks are capable of reproduction and can produce a fertile offspring.

Question 2:

In evolutionary terms, can we say which among bacteria, spiders, fish and chimpanzees have a 'better' body design? Why or why not?

Answer:

Evolution cannot always be equated with progress or better body designs. Evolution simply creates more complex body designs. However, this does not mean that the simple body designs are inefficient. In fact, bacteria having a simple body design are still the most cosmopolitan organisms found on earth. They can survive hot springs, deep sea, and even freezing environment. Therefore, bacteria, spiders, fish, and chimpanzees are all different branches of evolution.

==

Question 1:

A Mendelian experiment consisted of breeding tall pea plants bearing violet flowers with short pea plants bearing white flowers. The progeny all bore violet flowers, but almost half of them were short. This suggests that the genetic make-up of the tall parent can be depicted as

(a) TTWW

(b) TTww

(c) TtWW

(d) TtWw

Answer:

(c) The genetic make-up of the tall parent can be depicted as TtWW Since all the progeny bore violet flowers, it means that the tall plant having violet flowers has WW genotype for violet flower colour. Since the progeny is both tall and short, the parent plant was not a pure tall plant. Its genotype must be Tt. Therefore, the cross involved in the given question is

TtWw × ttww

↓

TtWw – ttww

Therefore, half the progeny is tall, but all of them have violet flowers.

Question 2:

An example of homologous organs is

(a) our arm and a dog's fore-leg.

(b) our teeth and an elephant's tusks.

(c) potato and runners of grass.

(d) all of the above.

Answer:

(b) An example of homologous organs is our teeth and an elephant's tusks

Question 3:

In evolutionary terms, we have more in common with

(a) a Chinese school-boy.

(b) a chimpanzee.

(c) a spider.

(d) a bacterium.

Answer:

(a) In evolutionary terms, we have more in common with a Chinese school boy.

Question 4:

A study found that children with light-coloured eyes are likely to have parents with lightcoloured eyes. On this basis, can we say anything about whether the light eye colour trait is dominant or recessive? Why or why not?

Answer:

Let us assume that children with light-coloured eyes can either have LL or Ll or ll genotype. If the children have LL genotype, then their parents will also be of LL genotype.

LL × LL

↓

LL

If the children with light-coloured eyes have ll genotype, then their parents will also have ll genotype.

ll × ll

↓

ll

Therefore, it cannot be concluded whether light eye colour is dominant or recessive.

Question 5:

How are the areas of study – evolution and classification – interlinked?

Answer:

Classification involves grouping of organism into a formal system based on similarities in internal and external structure or evolutionary history.

Two species are more closely related if they have more characteristics in common. And if two species are more closely related, then it means they have a more recent ancestor.

For example, in a family, a brother and sister are closely related and they have a recent common ancestor i.e., their parents. A brother and his cousin are also related but less than the sister and her brother. This is because the brother and his cousin have a common ancestor i.e., their grandparents in the second generation whereas the parents were from the first generation.

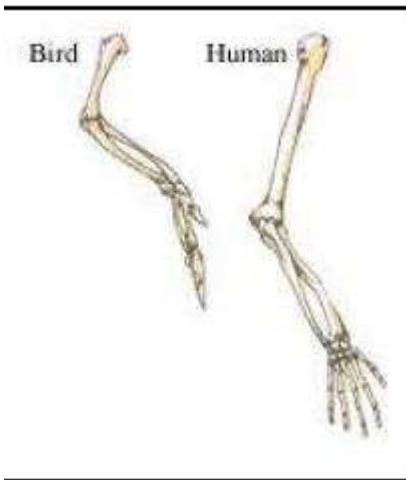
With subsequent generations, the variations make organisms more different than their ancestors. This discussion clearly proves that we classify organisms according to their resemblance which is similar to creating an evolutionary tree.

Question 6:

Explain the terms analogous and homologous organs with examples.

Answer:

Homologous organs are similar in origin (or are embryologically similar) but perform different functions. For example, the forelimbs of humans and the wings of birds look different externally but their skeletal structure is similar. It means that their origin is similar (as wings in birds are modifications of forearm) but functions are different – the wings help in flight whereas human forearm helps in various activities.



Homologous organs

Analogous organs, on the other hand, have different origin but perform similar functions. For example, the wings of a bird and a bat are similar in function but this similarity does not mean that these animals are more closely related. If we carefully look at these structures, then we will find that the wings of a bat are just the folds of skin that are stretched between its fingers whereas the wings of birds are present all along the arm. Therefore, these organs are analogous organs.

Question 7:

Outline a project which aims to find the dominant coat colour in dogs.

Answer:

Dogs have a variety of genes that govern coat colour. There are at least eleven identified gene series (A, B, C, D, E, F, G, M, P, S, T) that influence coat colour in dog. A dog inherits one gene from each of its parents. The dominant gene gets expressed in the phenotype. For example, in the B series, a dog can be genetically black or brown. Let us assume that one parent is homozygous black (BB), while the

WE ARE WITH YOU.....

other parent is homozygous brown (bb)

bb	BB	
	B	B
	b	Bb
	b	Bb

In this case, all the offsprings will be heterozygous (Bb).

Since black (B) is dominant, all the offsprings will be black. However, they will have both B and b alleles.

If such heterozygous pups are crossed, they will produce 25% homozygous black (BB), 50% heterozygous black (Bb), and 25% homozygous brown (bb) offsprings.

	B	b
B	BB	Bb
b	Bb	bb

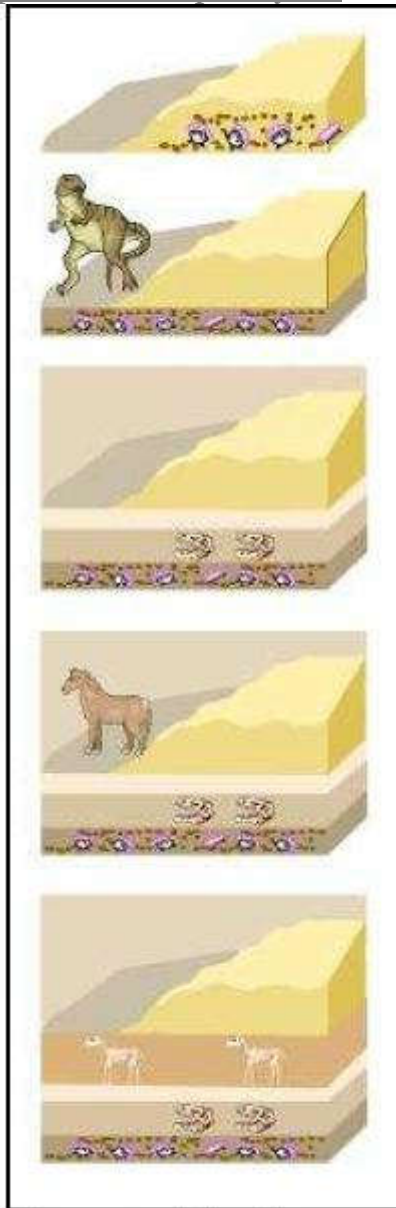
Question 8:

Explain the importance of fossils in deciding evolutionary relationships.

Answer:

Fossils are the remains of the organism that once existed on earth. They represent the ancestors of the plants and animals that are alive today. They provide evidences of evolution by revealing the characteristics of the past organisms and the changes that have occurred in these organisms to give rise to the present organisms. Let us explain the importance of fossils in deciding evolutionary history with the help of the following example. Around 100 million years ago, some invertebrates died and were buried in the soil in that area. More sediment accumulated on top of it turning it into sedimentary rock. At the same place, millions of years later, some dinosaurs died and their bodies were buried on top of the sedimentary rock. The mud containing dinosaurs also turned into a rock. Then, millions of years later, some horse-like creatures died in that area and got fossilized in rocks above the dinosaur fossils. Some time later, due to soil erosion or floods in that area, the rocks containing horselike fossils are exposed. If that area is excavated deeper, then the dinosaur and invertebrates fossils can also be found. Thus, by digging that area, scientists can easily predict that horse-like animals evolved later than the dinosaurs and the invertebrates. Thus, the above example suggests that the fossils found closer to the surface of the earth are more recent ones than the fossils

present in deeper layers.



Layers of fossils

HomeDelivery

WE ARE WITH YOU.....

Question 9:

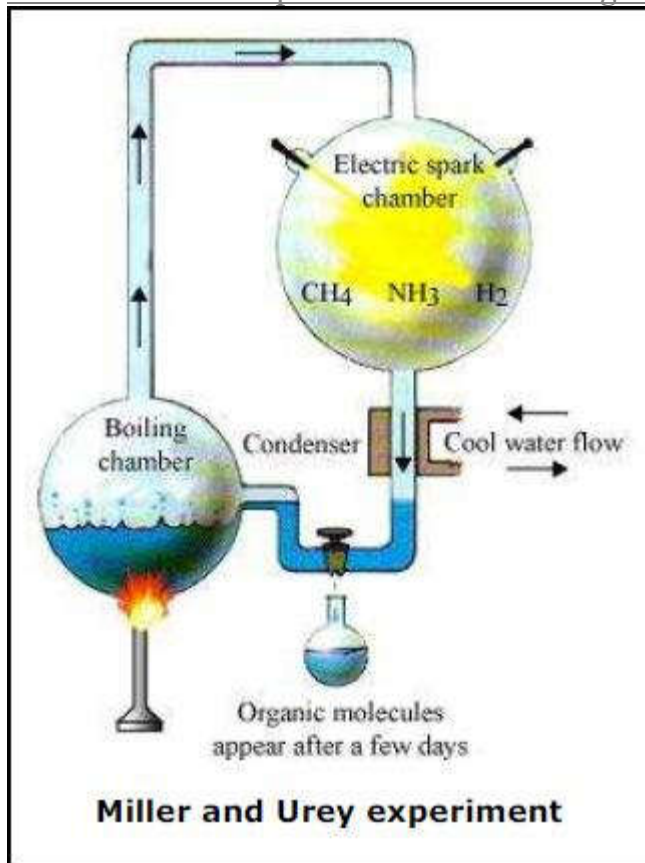
What evidence do we have for the origin of life from inanimate matter?

Answer:

A British scientist, J.B.S. Haldane, suggested that life originated from simple inorganic molecules. He believed that when the earth was formed, it was a hot gaseous mass containing elements such as nitrogen, oxygen, carbon, hydrogen, etc. These elements combined to form molecules like water (H_2O), carbon dioxide (CO_2), methane (CH_4), ammonia (NH_3), etc.

After the formation of water, slowly the earth surface cooled and the inorganic molecules interacted with one another in water to form simple organic molecules such as sugars, fatty acids, amino acids, etc. The energy for these reactions was provided by solar radiations, lightning, volcanic eruptions, etc. This was proved by the experiment of Stanley L. Miller and Harold C. Urey in 1953. They took a mixture of water (H_2O), methane (CH_4), ammonia (NH_3), and hydrogen gas (H_2) in a chamber and sparks were passed through this mixture using two electrodes. After one week, 15% of the carbon from methane was converted into amino acids, sugars, etc. These organic molecules are polymerized and

assembled to form protein molecules that gave rise to life on earth.



Question 10:

Explain how sexual reproduction gives rise to more viable variations than asexual reproduction. How does this affect the evolution of those organisms that reproduce sexually?

Answer:

In sexual reproduction, two individuals having different variations combine their DNA to give rise to a new individual. Therefore, sexual reproduction allows more variations, whereas in asexual reproduction, chance variations can only occur when the copying of DNA is not accurate. Additionally, asexual reproduction allows very less variations because if there are more variations, then the resultant DNA will not be able to survive inside the inherited cellular apparatus. However, in sexual reproduction, more variations are allowed and the resultant DNA is also able to survive, thus making the variations viable.

Variation and Evolution: Variants help the species to survive in all the conditions. Environmental conditions such as heat, light, pests, and food availability can change suddenly at only one place. At that time, only those variants resistant to these conditions would be able to survive. This will slowly lead to the evolution of a better adapted species. Thus, variation helps in the evolution of sexually reproducing organisms.

Question 11:

How is the equal genetic contribution of male and female parents ensured in the progeny?

Answer:

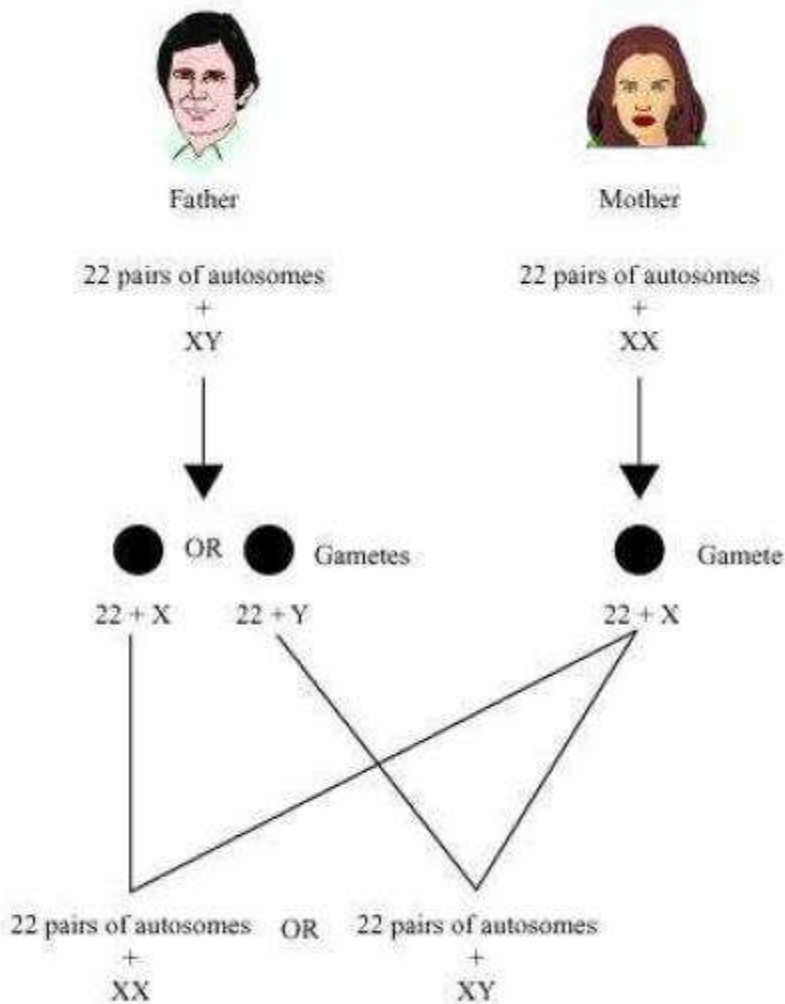
In human beings, every somatic cell of the body contains 23 pairs of chromosomes. Out of these 23 pairs, the first 22 pairs are known as autosomes and the remaining one pair is known as sex chromosomes represented as X and Y.

Females have two X chromosomes and males have one X and one Y chromosome.

The gamete receives half of the chromosomes. Therefore, the male gametes have 22 autosomes and either X or Y chromosome.

The female gamete, on the other hand, has 22 autosomes and X chromosome.

During reproduction, the male and female gametes fuse and thus the progeny receives 22 autosomes and one X or Y chromosome from male parent and 22 autosomes and one X chromosome from the female parent.



Delivery
We are with you.....

Question 12:

Only variations that confer an advantage to an individual organism will survive in a population. Do you agree with this statement? Why or why not?

Answer:

In species, variations that offer survival advantages are naturally selected. Individuals adjust to their environments with the help of these selected variations and consequently these variations are passed on to their progeny. Evolution of organisms occurs as a result of this natural selection.

However, there can be some other variations, which do not offer any survival advantage and arise only accidentally. Such variations in small populations can change the frequency of some genes even if they are not important for survival.

This accidental change in the frequency of genes in small populations is referred to as genetic drift. Thus, genetic drift provides diversity (variations) without any survival advantage.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 10 Light Reflection and Refraction Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 10 Light Reflection and Refraction Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 10 Light Reflection and Refraction NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

Define the principal focus of a concave mirror.

Answer:

Light rays that are parallel to the principal axis of a concave mirror converge at a specific point on its principal axis after reflecting from the mirror. This point is known as the principal focus of the concave mirror.

Question 2:

The radius of curvature of a spherical mirror is 20 cm. What is its focal length?

Answer:

Radius of curvature, $R = 20$ cm

Radius of curvature of a spherical mirror = $2 \times$ Focal length (f)

$$f = \frac{R}{2} = \frac{20}{2} = 10 \text{ cm}$$

Hence, the focal length of the given spherical mirror is 10 cm.

Question 3:

Name the mirror that can give an erect and enlarged image of an object.

Answer:

When an object is placed between the pole and the principal focus of a concave mirror, the image formed is virtual, erect, and enlarged.

Question 4:

Why do we prefer a convex mirror as a rear-view mirror in vehicles?

Answer:

Convex mirrors give a virtual, erect, and diminished image of the objects placed in front of them. They are preferred as a rear-view mirror in vehicles because they give a wider field of view, which allows the driver to see most of the traffic behind him.

—

Question 1:

Find the focal length of a convex mirror whose radius of curvature is 32 cm.

Answer:

Radius of curvature, $R = 32$ cm

Radius of curvature = $2 \times$ Focal length (f)

$$R = 2f$$

$$f = \frac{R}{2} = \frac{32}{2} = 16 \text{ cm}$$

Hence, the focal length of the given convex mirror is 16 cm.

Question 2:

A concave mirror produces three times magnified (enlarged) real image of object placed at 10 cm in front of it. Where is the image located?

Answer:

Magnification produced by a spherical mirror is given by the relation,

$$m = \frac{\text{Height of the image}}{\text{Height of the object}} = -\frac{\text{Image distance}}{\text{Object distance}}$$

$$m = \frac{h_1}{h_o} = -\frac{v}{u}$$

Let the height of the object, $h_o = h$

Then, height of the image, $h_1 = -3h$ (Image formed is real)

$$\frac{-3h}{h} = \frac{-v}{u}$$

$$\frac{v}{u} = 3$$

Object distance, $u = -10 \text{ cm}$

$$v = 3 \times (-10) = -30 \text{ cm}$$

Let the height of the object, $h_o = h$

Then, height of the image, $h_1 = -3h$ (Image formed is real)

$$-3h/h = -u/v$$

$$v/u = 3$$

Object distance, $u = -10 \text{ cm}$

$$v = 3 \times (-10) = -30 \text{ cm}$$

Here, the negative sign indicates that an inverted image is formed at a distance of 30 cm in front of the given concave mirror.

=

Question 1:

A ray of light travelling in air enters obliquely into water. Does the light ray bend towards the normal or away from the normal? Why?

Answer:

The light ray bends towards the normal. When a ray of light travels from an optically rarer medium to an optically denser medium, it gets bent towards the normal. Since water is optically denser than air, a ray of light travelling from air into the water will bend towards the normal.

elivery

Be WITH you.....

Question 2:

Light enters from air to glass having refractive index 1.50. What is the speed of light in the glass? The speed of light in vacuum is $3 \times 10^8 \text{ m s}^{-1}$.

Answer:

Refractive index of a medium n_m is given by,

Refractive index of a medium n_m is given by,

$$n_m = \frac{\text{Speed of light in vacuum}}{\text{Speed of light in the medium}} = \frac{c}{v}$$

Speed of light in vacuum, $c = 3 \times 10^8 \text{ m s}^{-1}$

Refractive index of glass, $n_g = 1.50$

$$v = \frac{c}{n_g} = \frac{3 \times 10^8}{1.50} = 2 \times 10^8 \text{ m s}^{-1}$$

Speed of light in the glass,

Question 3:

Find out, from Table, the medium having highest optical density. Also find the medium with lowest optical density



Material medium	Refractive index	Material medium	Refractive index
Air	1.0003	Canada Balsam	1.53
Ice	1.31	-	-
Water	1.33	Rock salt	1.54
Alcohol	1.36	-	-
Kerosene	1.44	Carbon disulphide	1.63
Fused quartz	1.46	Dense flint glass	1.65
Turpentine oil	1.47	Ruby	1.71
Benzene	1.50	Sapphire	1.77
Crown glass	1.52	Diamond	2.42

Answer:

Highest optical density = Diamond

Lowest optical density = Air

Optical density of a medium is directly related with the refractive index of that medium. A medium which has the highest refractive index will have the highest optical density and vice-versa.

It can be observed from table 10.3 that diamond and air respectively have the highest and lowest refractive index. Therefore, diamond has the highest optical density and air has the lowest optical density.

Question 4:

You are given kerosene, turpentine and water. In which of these does the light travel fastest? Use the information given in Table.

Material medium	Refractive index	Material medium	Refractive index
Air	1.0003	Canada Balsam	1.53
Ice	1.31	-	-
Water	1.33	Rock salt	1.54
Alcohol	1.36	-	-
Kerosene	1.44	Carbon disulphide	1.63
Fused quartz	1.46	Dense flint glass	1.65
Turpentine oil	1.47	Ruby	1.71
Benzene	1.50	Sapphire	1.77
Crown glass	1.52	Diamond	2.42

Answer:

Speed of light in a medium is given by the relation for refractive index (n_m). The relation is given as

$$n_m = \frac{\text{Speed of light in air}}{\text{Speed of light in the medium}} = \frac{c}{v}$$

$$v = \frac{c}{n_m}$$

$$v \propto \frac{1}{n_m}$$

Answer:

Speed of light in a medium is given by the relation for refractive index (n_m). The relation is given as $n(\text{medium}) = \text{speed of light in air} / \text{speed of light in medium} = c/v$
 $v = c/n(\text{medium})$

v is proportional to $1/n(\text{medium})$

It can be inferred from the relation that light will travel the slowest in the material which has the highest refractive index and travel the fastest in the material which has the lowest refractive index. It can be observed from table 10.3 that the refractive indices of kerosene, turpentine, and water are 1.44, 1.47, and 1.33 respectively. Therefore, light travels the fastest in water.

Question 5:

The refractive index of diamond is 2.42. What is the meaning of this statement?

Answer:

Refractive index of a medium n_m is related to the speed of light in that medium v by the relation:

$$n_m = \frac{\text{Speed of light in air}}{\text{Speed of light in the medium}} = \frac{c}{v}$$

Where, c is the speed of light in vacuum/air

The refractive index of diamond is 2.42. This suggests that the speed of light in diamond will reduce by a factor 2.42 compared to its speed in air.

=

Question 1:

Define 1 dioptre of power of a lens.

Answer:

Power of lens is defined as the reciprocal of its focal length. If P is the power of a lens of focal length F in metres, then

$$P = \frac{1}{f(\text{in metres})}$$

The S.I. unit of power of a lens is Dioptre. It is denoted by D.

1 dioptre is defined as the power of a lens of focal length 1 metre.

$$\therefore 1 \text{ D} = 1 \text{ m}^{-1}$$

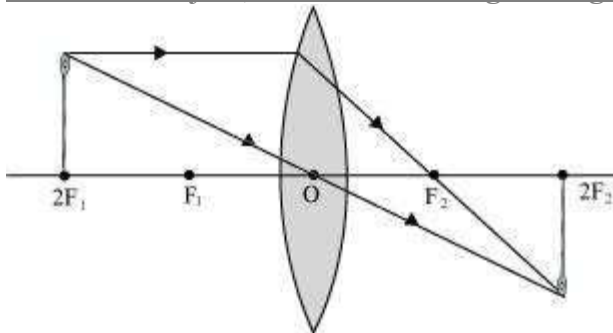
WE ARE WITH YOU.....

Question 2:

A convex lens forms a real and inverted image of a needle at a distance of 50 cm from it. Where is the needle placed in front of the convex lens if the image is equal to the size of the object? Also, find the power of the lens.

Answer:

When an object is placed at the centre of curvature, $2F_1$, of a convex lens, its image is formed at the centre of curvature, $2F_2$, on the other side of the lens. The image formed is inverted and of the same size as the object, as shown in the given figure.



It is given that the image of the needle is formed at a distance of 50 cm from the convex lens. Hence, the needle is placed in front of the lens at a distance of 50 cm.

Object distance, $u = -50 \text{ cm}$

Image distance, $v = 50$ cm

Focal length = f

According to the lens formula,

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{f} = \frac{1}{50} - \frac{1}{(-50)} = \frac{1}{50} + \frac{1}{50} = \frac{1}{25}$$

$$f = 25 \text{ cm} = 0.25 \text{ m}$$

$$\text{Power of the lens, } P = \frac{1}{f(\text{in meters})} = \frac{1}{0.25} = +4 \text{ D}$$

Hence, the power of the given lens is +4 D.

Question 3:

Find the power of a concave lens of focal length 2 m.

Answer:

Focal length of concave lens, $f = 2$ m

$$\text{Power of a lens, } P = \frac{1}{f(\text{in meters})} = \frac{1}{(-2)} = -0.5 \text{ D}$$

Here, negative sign arises due to the divergent nature of concave lens.

Hence, the power of the given concave lens is -0.5 D.

=

Question 1:

Which one of the following materials cannot be used to make a lens?

- (a) Water
- (b) Glass
- (c) Plastic
- (d) Clay

Answer:

(d) A lens allows light to pass through it. Since clay does not show such property, it cannot be used to make a lens.

Question 2:

The image formed by a concave mirror is observed to be virtual, erect and larger than the object.

Where should be the position of the object?

- (a) Between the principal focus and the centre of curvature
- (b) At the centre of curvature
- (c) Beyond the centre of curvature
- (d) Between the pole of the mirror and its principal focus.

Answer:

(d) When an object is placed between the pole and principal focus of a concave mirror, the image formed is virtual, erect, and larger than the object.

Question 3:

Where should an object be placed in front of a convex lens to get a real image of the size of the object?

- (a) At the principal focus of the lens
- (b) At twice the focal length
- (c) At infinity
- (d) Between the optical centre of the lens and its principal focus.

Answer:

(b) When an object is placed at the centre of curvature in front of a convex lens, its image is formed at the centre of curvature on the other side of the lens. The image formed is real, inverted, and of the same size as the object.

Question 4:

A spherical mirror and a thin spherical lens have each a focal length of -15 cm. The mirror and the lens are likely to be

- (a) both concave
- (b) both convex
- (c) the mirror is concave and the lens is convex
- (d) the mirror is convex, but the lens is concave

Answer:

(a) By convention, the focal length of a concave mirror and a concave lens are taken as negative. Hence, both the spherical mirror and the thin spherical lens are concave in nature.

Question 5:

No matter how far you stand from a mirror, your image appears erect. The mirror is likely to be

- (a) plane
- (b) concave
- (c) convex
- (d) either plane or convex

Answer:

(d) A convex mirror always gives a virtual and erect image of smaller size of the object placed in front of it. Similarly, a plane mirror will always give a virtual and erect image of same size as that of the object placed in front of it. Therefore, the given mirror could be either plane or convex.

Question 6:

Which of the following lenses would you prefer to use while reading small letters found in a dictionary?

- (a) A convex lens of focal length 50 cm
- (b) A concave lens of focal length 50 cm
- (c) A convex lens of focal length 5 cm
- (d) A concave lens of focal length 5 cm

Answer:

(c) A convex lens gives a magnified image of an object when it is placed between the radius of curvature and focal length. Also, magnification is more for convex lenses having shorter focal length. Therefore, for reading small letters, a convex lens of focal length 5 cm should be used.

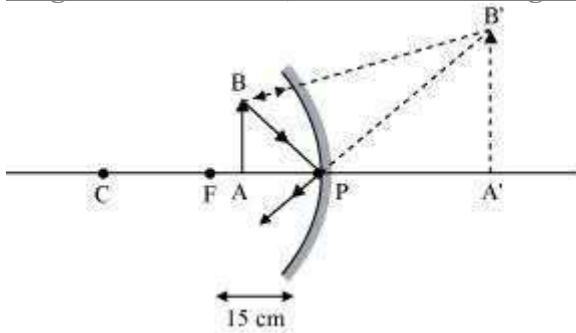
Question 7:

We wish to obtain an erect image of an object, using a concave mirror of focal length 15 cm. What should be the range of distance of the object from the mirror? What is the nature of the image? Is the image larger or smaller than the object? Draw a ray diagram to show the image formation in this case.

Answer:

Range of object distance = 0 cm to 15 cm A concave mirror gives an erect image when an object is placed between its pole (P) and the principal focus (F).

Hence, to obtain an erect image of an object from a concave mirror of focal length 15 cm, the object must be placed anywhere between the pole and the focus. The image formed will be virtual, erect, and magnified in nature, as shown in the given figure.



Question 8:

Name the type of mirror used in the following situations.

(a) Headlights of a car

(b) Side/rear-view mirror of a vehicle

(c) Solar furnace

Support your answer with reason.

Answer:

(a) Concave (b) Convex (c) Concave

Explanation

(a) Concave mirror is used in the headlights of a car. This is because concave mirrors can produce powerful parallel beam of light when the light source is placed at their principal focus.

(b) Convex mirror is used in side/rear view mirror of a vehicle. Convex mirrors give a virtual, erect, and diminished image of the objects placed in front of it. Because of this, they have a wide field of view. It enables the driver to see most of the traffic behind him/her.

(c) Concave mirrors are convergent mirrors. That is why they are used to construct solar furnaces. Concave mirrors converge the light incident on them at a single point known as principal focus. Hence, they can be used to produce a large amount of heat at that point.

Question 9:

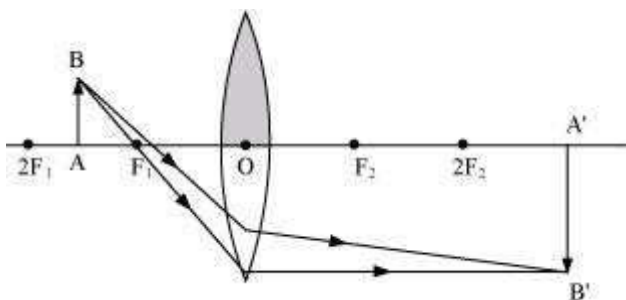
One-half of a convex lens is covered with a black paper. Will this lens produce a complete image of the object? Verify your answer experimentally. Explain your observations.

Answer:

The convex lens will form complete image of an object, even if its one half is covered with black paper. It can be understood by the following two cases.

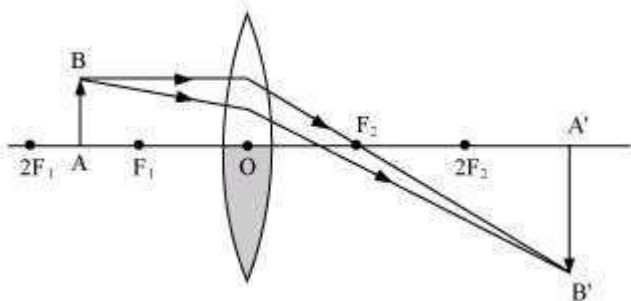
Case I

When the upper half of the lens is covered In this case, a ray of light coming from the object will be refracted by the lower half of the lens. These rays meet at the other side of the lens to form the image of the given object, as shown in the following figure.



Case II

When the lower half of the lens is covered In this case, a ray of light coming from the object is refracted by the upper half of the lens. These rays meet at the other side of the lens to form the image of the given object, as shown in the following figure.



Question 10:

An object 5 cm in length is held 25 cm away from a converging lens of focal length 10 cm. Draw the ray diagram and find the position, size and the nature of the image formed.

Answer:

Object distance, $u = -25$ cm

Object height, $h_o = 5$ cm

Focal length, $f = +10$ cm

According to the lens formula,

WE ARE WITH YOU.....

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{v} = \frac{1}{f} + \frac{1}{u} = \frac{1}{10} - \frac{1}{25} = \frac{15}{250}$$

$$v = \frac{250}{15} = 16.66 \text{ cm}$$

The positive value of v shows that the image is formed at the other side of the lens.

$$\text{Magnification, } m = -\frac{\text{Image distance}}{\text{Object distance}} = -\frac{v}{u} = \frac{-16.66}{25} = -0.66$$

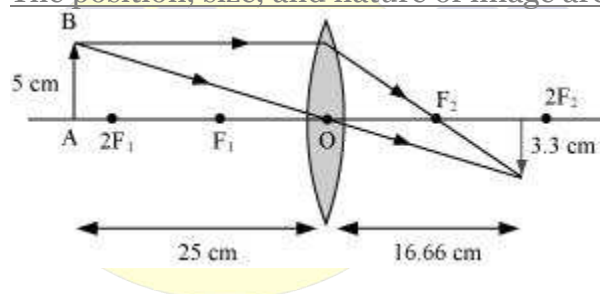
The negative sign shows that the image is real and formed behind the lens.

$$\text{Magnification, } m = \frac{\text{Image height}}{\text{Object height}} = \frac{H_i}{H_o} = \frac{H_i}{5}$$

$$H_i = m \times H_o = -0.66 \times 5 = -3.3 \text{ cm}$$

The negative value of image height indicates that the image formed is inverted.

The position, size, and nature of image are shown in the following ray diagram.



WE ARE WITH YOU.....

Question 11:

A concave lens of focal length 15 cm forms an image 10 cm from the lens. How far is the object placed from the lens? Draw the ray diagram.

Answer:

Focal length of concave lens (OF₁), $f = -15 \text{ cm}$

Image distance, $v = -10 \text{ cm}$

According to the lens formula,

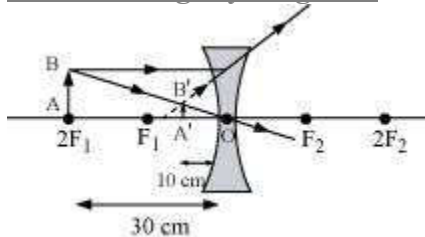
$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{u} = \frac{1}{v} - \frac{1}{f} = \frac{-1}{10} - \frac{1}{(-15)} = \frac{-1}{10} + \frac{1}{15} = \frac{-5}{150}$$

$$u = -30 \text{ cm}$$

The negative value of u indicates that the object is placed 30 cm in front of the lens. This is shown in

the following ray diagram.



Question 12:

An object is placed at a distance of 10 cm from a convex mirror of focal length 15 cm. Find the position and nature of the image.

Answer:

Focal length of convex mirror, $f = +15$ cm

Object distance, $u = -10$ cm

According to the mirror formula,

$$\frac{1}{v} + \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{v} = \frac{1}{f} - \frac{1}{u} = \frac{1}{15} + \frac{1}{10} = \frac{25}{150}$$

$$v = 6 \text{ cm}$$

The positive value of v indicates that the image is formed behind the mirror.

$$\text{Magnification, } m = -\frac{\text{Image distance}}{\text{Object distance}} = -\frac{v}{u} = \frac{-6}{-10} = +0.6$$

Question 13:

The magnification produced by a plane mirror is +1. What does this mean?

Answer:

Magnification produced by a mirror is given by the relation

$$\text{Magnification, } m = \frac{\text{Image height } (H_i)}{\text{Object height } (H_o)}$$

The magnification produced by a plane mirror is +1. It shows that the image formed by the plane mirror is of the same size as that of the object. The positive sign shows that the image formed is virtual and erect.

Question 14:

An object 5.0 cm in length is placed at a distance of 20 cm in front of a convex mirror of radius of curvature 30 cm. Find the position of the image, its nature and size.

Answer:

Object distance, $u = -20$ cm

Object height, $h = 5$ cm

Radius of curvature, $R = 30$ cm

Radius of curvature = $2 \times$ Focal length

$$R = 2f$$

$$f = 15 \text{ cm}$$

According to the mirror formula,

$$\frac{1}{v} + \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{v} = \frac{1}{f} - \frac{1}{u} = \frac{1}{15} + \frac{1}{20} = \frac{4+3}{60} = \frac{7}{60}$$

$$v = 8.57 \text{ cm}$$

The positive value of v indicates that the image is formed behind the mirror.

$$\text{Magnification, } m = -\frac{\text{Image distance}}{\text{Object distance}} = \frac{-8.57}{-20} = 0.428$$

The positive value of magnification indicates that the image formed is virtual.

$$\text{Magnification, } m = \frac{\text{Height of the image}}{\text{Height of the object}} = \frac{h'}{h}$$

$$h' = m \times h = 0.428 \times 5 = 2.14 \text{ cm}$$

The positive value of image height indicates that the image formed is erect. Therefore, the image formed is virtual, erect, and smaller in size.

Question 15:

An object of size 7.0 cm is placed at 27 cm in front of a concave mirror of focal length 18 cm. At what distance from the mirror should a screen be placed, so that a sharp focused image can be obtained? Find the size and the nature of the image.

Answer:

Object distance, $u = -27 \text{ cm}$

Object height, $h = 7 \text{ cm}$

Focal length, $f = -18 \text{ cm}$

According to the mirror formula,

$$\frac{1}{u} + \frac{1}{v} = \frac{1}{f}$$

$$\frac{1}{v} = \frac{1}{f} - \frac{1}{u} = \frac{-1}{18} + \frac{1}{27} = \frac{-1}{54}$$

$$v = -54 \text{ cm}$$

The screen should be placed at a distance of 54 cm in front of the given mirror.

$$\text{Magnification, } m = -\frac{\text{Image distance}}{\text{Object distance}} = \frac{-54}{27} = -2$$

The negative value of magnification indicates that the image formed is real.

$$\text{Magnification, } m = \frac{\text{Height of the image}}{\text{Height of the object}} = \frac{h'}{h}$$

$$h' = 7 \times (-2) = -14 \text{ cm}$$

The negative value of image height indicates that the image formed is inverted.

Question 16:

Find the focal length of a lens of power -2.0 D. What type of lens is this?

Answer:

$$\text{Power of a lens, } P = \frac{1}{f(\text{in metres})}$$

$$P = -2 \text{ D}$$

$$f = \frac{-1}{2} = -0.5 \text{ m}$$

Question 17:

A doctor has prescribed a corrective lens of power $+1.5$ D. Find the focal length of the lens. Is the prescribed lens diverging or converging?

Answer:

$$\text{Power of a lens, } P = \frac{1}{f(\text{in metres})}$$

$$\text{Power, } P = 1.5 \text{ D}$$

$$f = \frac{1}{1.5} = \frac{10}{15} = 0.66 \text{ m}$$

meDelivery

WE ARE WITH YOU.....

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 11 Human Eye and Colourful World Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 11 Human Eye and Colourful World Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 11 Human Eye and Colourful World NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

What is meant by power of accommodation of the eye?

Answer:

When the ciliary muscles are relaxed, the eye lens becomes thin, the focal length increases, and the distant objects are clearly visible to the eyes. To see the nearby objects clearly, the ciliary muscles contract making the eye lens thicker. Thus, the focal length of the eye lens decreases and the nearby objects become visible to the eyes. Hence, the human eye lens is able to adjust its focal length to view both distant and nearby objects on the retina. This ability is called the power of accommodation of the eyes.

Question 2:

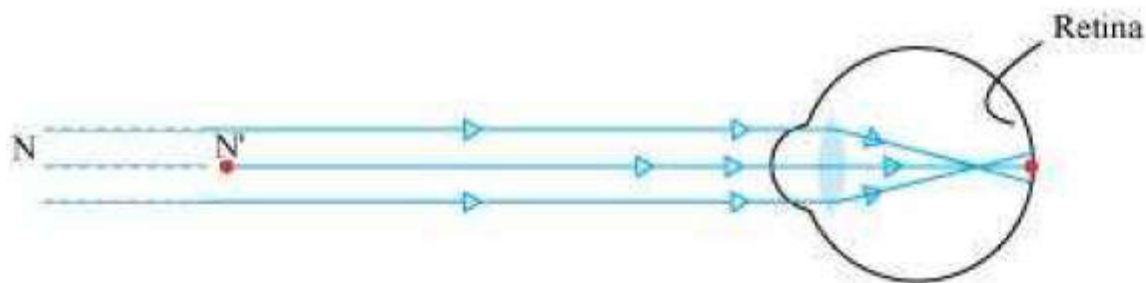
A person with a myopic eye cannot see objects beyond 1.2 m distinctly. What should be the type of the corrective lens used to restore proper vision?

Answer:

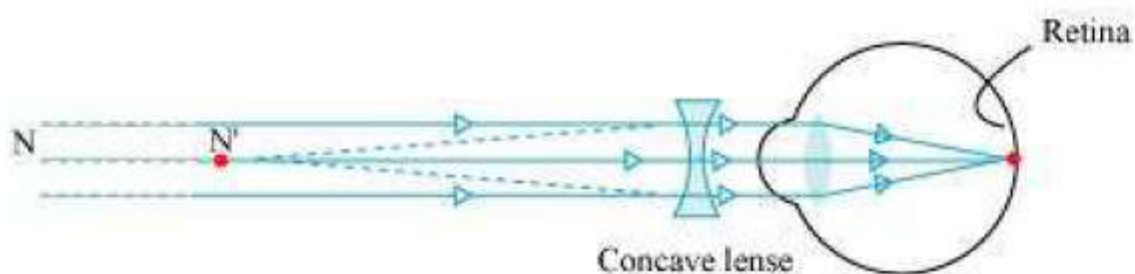
The person is able to see nearby objects clearly, but he is unable to see objects beyond 1.2 m. This happens because the image of an object beyond 1.2 m is formed in front of the retina and not at the

WE ARE WITH YOU.....

retina, as shown in the given figure.



To correct this defect of vision, he must use a concave lens. The concave lens the image back to the retina as shown in the given figure.



Question 3:

What is the far point and near point of the human eye with normal vision?

Answer:

The near point of the eye is the minimum distance of the object from the eye, which can be seen distinctly without strain. For a normal human eye, this distance is 25 cm. The far point of the eye is the maximum distance to which the eye can see the objects clearly. The far point of the normal human eye is infinity.

Question 4:

A student has difficulty reading the blackboard while sitting in the last row. What could be the defect the child is suffering from? How can it be corrected?

Answer:

A student has difficulty in reading the blackboard while sitting in the last row. It shows that he is unable to see distant objects clearly. He is suffering from myopia. This defect can be corrected by using a concave lens.

Question 1:

The human eye can focus objects at different distances by adjusting the focal length of the eye lens.

This is due to

- (a) presbyopia
- (b) accommodation
- (c) near-sightedness
- (d) far-sightedness

Answer:

(b) Human eye can change the focal length of the eye lens to see the objects situated at various distances from the eye. This is possible due to the power of accommodation of the eye lens.

Question 2:

The human eye forms the image of an object at its

(a) cornea (b) iris (c) pupil (d) retina

Answer:

(d) The human eye forms the image of an object at its retina.

Question 3:

The least distance of distinct vision for a young adult with normal vision is about

(a) 25 m

(b) 2.5 cm

(c) 25 cm

(d) 2.5 m

Answer:

(c) The least distance of distinct vision is the minimum distance of an object to see clear and distinct image. It is 25 cm for a young adult with normal visions.

Question 4:

The change in focal length of an eye lens is caused by the action of the

(a) pupil

(b) retina

(c) ciliary muscles

(d) iris

Answer:

(c) The relaxation or contraction of ciliary muscles changes the curvature of the eye lens. The change in curvature of the eye lens changes the focal length of the eyes. Hence, the change in focal length of an eye lens is caused by the action of ciliary muscles.

Question 5:

A person needs a lens of power -5.5 dioptres for correcting his distant vision. For correcting his near vision he needs a lens of power $+1.5$ dioptre. What is the focal length of the lens required for correcting (i) distant vision, and (ii) near vision?

Answer:

For distant vision = -0.181 m, for near vision = 0.667 m

WE ARE WITH YOU.....

The power P of a lens of focal length f is given by the relation

$$P = \frac{1}{f \text{ (in metres)}}$$

(i) Power of the lens used for correcting distant vision = -5.5 D

Focal length of the required lens, $f = \frac{1}{P}$

$$f = \frac{1}{-5.5} = -0.181 \text{ m}$$

The focal length of the lens for correcting distant vision is -0.181 m .

(ii) Power of the lens used for correcting near vision = $+1.5 \text{ D}$

Focal length of the required lens, $f = \frac{1}{P}$

$$f = \frac{1}{1.5} = +0.667 \text{ m}$$

The focal length of the lens for correcting near vision is 0.667 m .

Question 6:

The far point of a myopic person is 80 cm in front of the eye. What is the nature and power of the lens required to correct the problem?

Answer:

The person is suffering from an eye defect called myopia. In this defect, the image is formed in front of the retina. Hence, a concave lens is used to correct this defect of vision.

Object distance, $u = \text{infinity} =$

Image distance, $v = -80 \text{ cm}$

Focal length = f

WE ARE WITH YOU.....

According to the lens formula,

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

$$-\frac{1}{80} - \frac{1}{\infty} = \frac{1}{f}$$

$$\frac{1}{f} = -\frac{1}{80}$$

$$f = -80 \text{ cm} = -0.8 \text{ m}$$

We know,

$$\text{Power, } P = \frac{1}{f (\text{in metres})}$$

$$P = \frac{1}{-0.8} = -1.25 \text{ D}$$

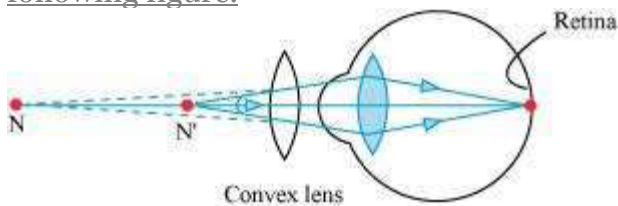
A concave lens of power -1.25 D is required by the person to correct his defect.

Question 7:

Make a diagram to show how hypermetropia is corrected. The near point of a hypermetropic eye is 1 m . What is the power of the lens required to correct this defect? Assume that the near point of the normal eye is 25 cm .

Answer:

A person suffering from hypermetropia can see distinct objects clearly but faces difficulty in seeing nearby objects clearly. It happens because the eye lens focuses the incoming divergent rays beyond the retina. This defect of vision is corrected by using a convex lens. A convex lens of suitable power converges the incoming light in such a way that the image is formed on the retina, as shown in the following figure.



Correction for hypermetropic eye

The convex lens actually creates a virtual image of a nearby object (N' in the figure) at the near point of vision (N) of the person suffering from hypermetropia. The given person will be able to clearly see the object kept at 25 cm (near point of the normal eye), if the image of the object is formed at his near point, which is given as 1 m .

Object distance, $u = -25 \text{ cm}$

Image distance, $v = -1 \text{ m} = -100 \text{ cm}$

Focal length, f

Using the lens formula,

$$\frac{1}{v} - \frac{1}{u} = \frac{1}{f}$$

$$\frac{1}{-100} - \frac{1}{-25} = \frac{1}{f}$$

$$\frac{1}{f} = \frac{1}{25} - \frac{1}{100}$$

$$\frac{1}{f} = \frac{4-1}{100}$$

$$f = \frac{100}{3} = 33.3 \text{ cm} = 0.33 \text{ m}$$

$$\text{Power, } P = \frac{1}{f \text{ (in metres)}} = \frac{1}{0.33 \text{ m}} = +3.0 \text{ D}$$

A convex lens of power +3.0 D is required to correct the defect.

Question 9:

What happens to the image distance in the eye when we increase the distance of an object from the eye?

Answer:

Since the size of eyes cannot increase or decrease, the image distance remains constant. When we increase the distance of an object from the eye, the image distance in the eye does not change. The increase in the object distance is compensated by the change in the focal length of the eye lens. The focal length of the eyes changes in such a way that the image is always formed at the retina of the eye.

Question 10:

Why do stars twinkle?

Answer:

Stars emit their own light and they twinkle due to the atmospheric refraction of light. Stars are very far away from the earth. Hence, they are considered as point sources of light. When the light coming from stars enters the earth's atmosphere, it gets refracted at different levels because of the variation in the air density at different levels of the atmosphere. When the star light refracted by the atmosphere comes more towards us, it appears brighter than when it comes less towards us. Therefore, it appears as if the stars are twinkling at night.

Question 11:

Explain why the planets do not twinkle?

Answer:

Planets do not twinkle because they appear larger in size than the stars as they are relatively closer to earth. Planets can be considered as a collection of a large number of point-size sources of light. The different parts of these planets produce either brighter or dimmer effect in such a way that the average of brighter and dimmer effect is zero. Hence, the twinkling effects of the planets are nullified and they do not twinkle.

Question 12:

Why does the Sun appear reddish early in the morning?

Answer:

During sunrise, the light rays coming from the Sun have to travel a greater distance in the earth's atmosphere before reaching our eyes. In this journey, the shorter wavelengths of lights are scattered out and only longer wavelengths are able to reach our eyes. Since blue colour has a shorter wavelength and red colour has a longer wavelength, the red colour is able to reach our eyes after the atmospheric scattering of light. Therefore, the Sun appears reddish early in the morning.

Question 13:

Why does the sky appear dark instead of blue to an astronaut?

Answer:

The sky appears dark instead of blue to an astronaut because there is no atmosphere in the outer space that can scatter the sunlight. As the sunlight is not scattered, no scattered light reach the eyes of the astronauts and the sky appears black to them.



NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 12 Electricity>

Download 2017 2018 New Edition PDF

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 12 Electricity>
Download 2017 2018 New Edition PDF

Class 10 Science Chapter 12 Electricity NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

What does an electric circuit mean?

Answer:

An electric circuit consists of electric devices, switching devices, source of electricity, etc. that are connected by conducting wires.

Question 2:

Define the unit of current.

Answer:

The unit of electric current is ampere (A). 1 A is defined as the flow of 1 C of charge through a wire in 1 s.

Question 3:

Calculate the number of electrons constituting one coulomb of charge.

Answer:

One electron possesses a charge of 1.6×10^{-19} C, i.e., 1.6×10^{-19} contained in 1 electron.

$\therefore 1 \text{ C of charge is contained in } \frac{1}{1.6 \times 10^{-19}} = 6.25 \times 10^{18} = 6 \times 10^{18} \text{ electrons}$

Therefore, 6×10^{18} electrons constitute one coulomb of charge.

Question 1:

Name a device that helps to maintain a potential difference across a conductor.

Answer:

A source of electricity such as cell, battery, power supply, etc. helps to maintain a potential difference across a conductor.

Question 2:

What is meant by saying that the potential difference between two points is 1 V?

Answer:

If 1 J of work is required to move a charge of amount 1 C from one point to another, then it is said that the potential difference between the two points is 1 V.

Question 3:

How much energy is given to each coulomb of charge passing through a 6 V battery?

Answer:

The energy given to each coulomb of charge is equal to the amount of work done to move it. The amount of work is given by the expression,

$$\text{Potential difference} = \frac{\text{Work done}}{\text{Charge}}$$

$$\text{Work Done} = \text{Potential Difference} \times \text{Charge}$$

Where,

$$\text{Charge} = 1 \text{ C}$$

$$\text{Potential difference} = 6 \text{ V}$$

$$\text{Work Done} = 6 \times 1 = 6 \text{ J}$$

Therefore, 6 J of energy is given to each coulomb of charge passing through a 6 V battery.

Question 1:

On what factors does the resistance of a conductor depend?

Answer:

The resistance of a conductor depends upon the following factors:

- (a) Length of the conductor
- (b) Cross-sectional area of the conductor
- (c) Material of the conductor
- (d) Temperature of the conductor

Question 2:

Will current flow more easily through a thick wire or a thin wire of the same material when connected to the same source? Why?

Answer:

Resistance of a wire,
$$R = \rho \frac{l}{A}$$

Where,

ρ = Resistivity of the material of the wire

l = Length of the wire

A = Area of cross-section of the wire

Resistance is inversely proportional to the area of cross-section of the wire.

Thicker the wire, lower is the resistance of the wire and vice-versa. Therefore, current can flow more easily through a thick wire than a thin wire.

Question 3:

Let the resistance of an electrical component remains constant while the potential difference across the two ends of the component decreases to half of its initial value.

What change will occur in the current through it?

Answer:

The change in the current flowing through the component is given by Ohm's law.

$$V = IR$$

$$I = \frac{V}{R}$$

Where,

Resistance of the electrical component = R

Potential difference = V

Current = I

The potential difference is reduced to half, keeping resistance constant.

Let the new resistance be R' and the new amount of current be I'.

Therefore, from Ohm's law, we obtain the amount of new current.

$$I' = \frac{V'}{R'} = \frac{\frac{V}{2}}{R} = \frac{1}{2} \left(\frac{V}{R} \right) = \frac{I}{2}$$

Therefore, the amount of current flowing through the electrical component is reduced by half.

Question 4:

Why are coils of electric toasters and electric irons made of an alloy rather than a pure metal?

Answer:

The resistivity of an alloy is higher than the pure metal. Moreover, at high temperatures, the alloys do not melt readily. Hence, the coils of heating appliances such as electric toasters and electric irons are made of an alloy rather than a pure metal.

Question 5:

Use the data in Table 12.2 to answer the following –

Table 12.2 Electrical resistivity of some substances at 20°C

–	Material	Resistivity ($\Omega \text{ m}$)
Conductors	Silver	1.60×10^{-8}
	Copper	1.62×10^{-8}
	Aluminium	2.63×10^{-8}
	Tungsten	5.20×10^{-8}
	Nickel	6.84×10^{-8}
	Iron	10.0×10^{-8}

	Chromium	12.9×10^{-8}
	Mercury	94.0×10^{-8}
	Manganese	1.84×10^{-6}
	Constantan (alloy of Cu and Ni)	49×10^{-6}
Alloys	Manganin (alloy of Cu, Mn and Ni)	44×10^{-6}
	Nichrome (alloy of Ni, Cr, Mn and Fe)	100×10^{-6}
	Glass	$10^{10} - 10^{14}$
Insulators	Hard rubber	$10^{13} - 10^{16}$
	Ebonite	$10^{15} - 10^{17}$
	Diamond	$10^{12} - 10^{13}$
	Paper (dry)	10^{12}

(a) Which among iron and mercury is a better conductor?

(b) Which material is the best conductor?

Answer:

(a) Resistivity of iron = $10.0 \times 10^{-8} \Omega \text{ m}$

Resistivity of mercury = $94.0 \times 10^{-8} \Omega \text{ m}$

Resistivity of mercury is more than that of iron. This implies that iron is a better conductor than mercury.

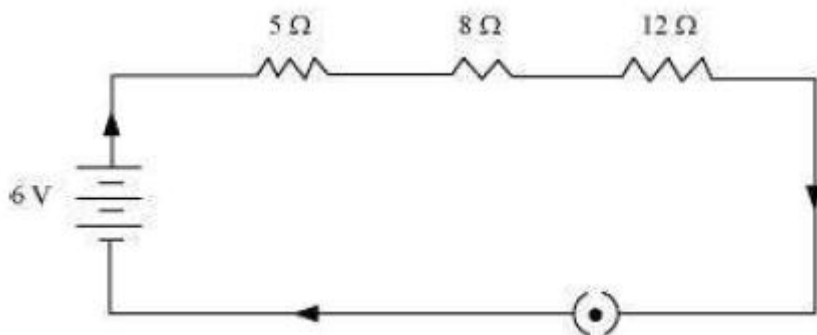
(b) It can be observed from Table 12.2 that the resistivity of silver is the lowest among the listed materials. Hence, it is the best conductor.

Question 1:

Draw a schematic diagram of a circuit consisting of a battery of three cells of 2 V each, a 5 G resistor, an 8 G resistor, and a 12 G resistor, and a plug key, all connected in series.

Answer:

Three cells of potential 2 V, each connected in series, is equivalent to a battery of potential $2 \text{ V} + 2 \text{ V} + 2 \text{ V} = 6 \text{ V}$. The following circuit diagram shows three resistors of resistances 5 G, 8 G and 12 G respectively connected in series and a battery of potential 6 V.



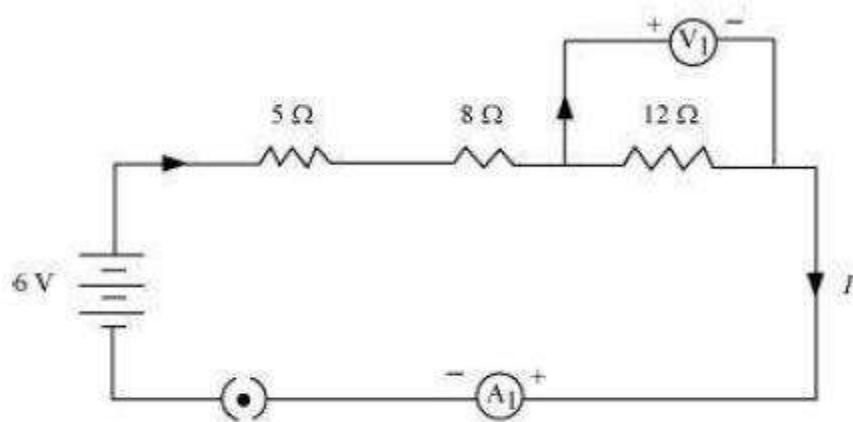
Question 2:

Redraw the circuit of question 1, putting in an ammeter to measure the current through the resistors and a voltmeter to measure potential difference across the 12 G resistor.

What would be the readings in the ammeter and the voltmeter?

Answer:

To measure the current flowing through the resistors, an ammeter should be connected in the circuit in series with the resistors. To measure the potential difference across the 12 G resistor, a voltmeter should be connected parallel to this resistor, as shown in the following figure.



The resistances are connected in series.

Ohm's law can be used to obtain the readings of ammeter and voltmeter. According to

Ohm's law,

$$V = IR,$$

Where,

Potential difference, $V = 6 \text{ V}$

Current flowing through the circuit/resistors = I

Resistance of the circuit, $R = 5 + 8 + 12 = 25 \Omega$

$$I = \frac{V}{R} = \frac{6}{25} = 0.24 \text{ A}$$

Potential difference across 12Ω resistor = V_1

Current flowing through the 12Ω resistor, $I = 0.24 \text{ A}$

Therefore, using Ohm's law, we obtain

$$V_1 = IR = 0.24 \times 12 = 2.88 \text{ V}$$

Therefore, the reading of the ammeter will be 0.24 A .

The reading of the voltmeter will be 2.88 V .

Question 1:

Judge the equivalent resistance when the following are connected in parallel – (a) 1Ω and $10^6 \Omega$, (b) 1Ω and $10^3 \Omega$ and $10^6 \Omega$.

Answer:

(a) When 1Ω and $10^6 \Omega$ are connected in parallel:

Let R be the equivalent resistance.

$$\therefore \frac{1}{R} = \frac{1}{1} + \frac{1}{10^6}$$

$$R = \frac{10^6}{10^6 + 1} \approx \frac{10^6}{10^6} = 1 \Omega$$

Therefore, equivalent resistance $\approx 1 \Omega$

(b) When 1Ω , $10^3 \Omega$, and $10^6 \Omega$ are connected in parallel:

Let R be the equivalent resistance.

ivery

ITH YOU.....

$$\frac{1}{R} = \frac{1}{1} + \frac{1}{10^3} + \frac{1}{10^6} = \frac{10^6 + 10^3 + 1}{10^6}$$

$$R = \frac{1000000}{1001001} = 0.999 \, \Omega$$

Therefore, equivalent resistance = $0.999 \, \Omega$

Question 2:

An electric lamp of $100 \, \Omega$, a toaster of resistance $50 \, \Omega$, and a water filter of resistance $500 \, \Omega$ are connected in parallel to a $220 \, \text{V}$ source. What is the resistance of an electric iron connected to the same source that takes as much current as all three appliances, and what is the current through it?

Answer:

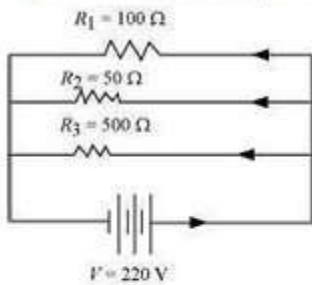
Resistance of electric lamp, $R_1 = 100 \, \Omega$

Resistance of toaster, $R_2 = 50 \, \Omega$

Resistance of water filter, $R_3 = 500 \, \Omega$

Voltage of the source, $V = 220 \, \text{V}$

These are connected in parallel, as shown in the following figure.



Let R be the equivalent resistance of the circuit.

$$\frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_2} + \frac{1}{R_3} = \frac{1}{100} + \frac{1}{50} + \frac{1}{500}$$

$$= \frac{5 + 10 + 1}{500} = \frac{16}{500}$$

$$R = \frac{500}{16} \, \Omega$$

According to Ohm's law,

delivery
WITH YOU.....

$$V = IR$$

$$I = \frac{V}{R}$$

Where,

Current flowing through the circuit = I

$$I = \frac{220}{\frac{500}{16}} = \frac{220 \times 16}{500} = 7.04 \text{ A}$$

7.04 A of current is drawn by all the three given appliances.

Therefore, current drawn by an electric iron connected to the same source of potential 220 V = 7.04 A

Let R' be the resistance of the electric iron. According to Ohm's law,

$$V = IR'$$

$$R' = \frac{V}{I} = \frac{220}{7.04} = 31.25 \Omega$$

Therefore, the resistance of the electric iron is 31.25Ω and the current flowing through it is 7.04 A.

Question 3:

What are the advantages of connecting electrical devices in parallel with the battery instead of connecting them in series?

Answer:

There is no division of voltage among the appliances when connected in parallel. The potential difference across each appliance is equal to the supplied voltage.

The total effective resistance of the circuit can be reduced by connecting electrical appliances in parallel.

Question 4:

How can three resistors of resistances 2 Ω , 3 Ω and 6 Ω be connected to give a total resistance of (a) 4 Ω , (b) 1 Ω ?

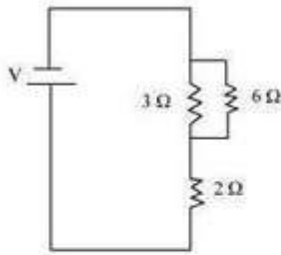
Answer:

There are three resistors of resistances 2 Ω , 3 Ω , and 6 Ω respectively.

(a) The following circuit diagram shows the connection of the three resistors.

delivery

WITH YOU.....



Here, 6 Ω and 3 Ω resistors are connected in parallel.

Therefore, their equivalent resistance will be given by

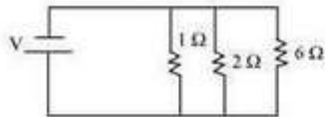
$$\frac{1}{\frac{1}{6} + \frac{1}{3}} = \frac{6 \times 3}{6 + 3} = 2 \Omega$$

This equivalent resistor of resistance 2 Ω is connected to a 2 Ω resistor in series.

Therefore, equivalent resistance of the circuit = 2 Ω + 2 Ω = 4 Ω

Hence, the total resistance of the circuit is 4 Ω.

2. The following circuit diagram shows the connection of the three resistors.



All the resistors are connected in series. Therefore, their equivalent resistance will be given as

$$\frac{1}{\frac{1}{2} + \frac{1}{3} + \frac{1}{6}} = \frac{1}{\frac{3+2+1}{6}} = \frac{6}{6} = 1 \Omega$$

Therefore, the total resistance of the circuit is 1 Ω.

Question 5:

What is (a) the highest, (b) the lowest total resistance that can be secured by combinations of four coils of resistance 4 Ω, 8 Ω, 12 Ω, 24 Ω?

Answer:

There are four coils of resistances 4 Ω, 8 Ω, 12 Ω, and 24 Ω respectively.

(a) If these coils are connected in series, then the equivalent resistance will be the highest, given by the sum 4 + 8 + 12 + 24 = 48 Ω

(b) If these coils are connected in parallel, then the equivalent resistance will be the lowest, given by

$$\frac{1}{\frac{1}{4} + \frac{1}{8} + \frac{1}{12} + \frac{1}{24}} = \frac{1}{\frac{6+3+2+1}{24}} = \frac{24}{12} = 2 \Omega$$

Therefore, 2 Ω is the lowest total resistance.

Question 1:

elivery

e WITH you.....

Why does the cord of an electric heater not glow while the heating element does?

Answer:

The heating element of an electric heater is a resistor. The amount of heat produced by it is proportional to its resistance. The resistance of the element of an electric heater is very high. As current flows through the heating element, it becomes too hot and glows red. On the other hand, the resistance of the cord is low. It does not become red when current flows through it.

Question 2:

Compute the heat generated while transferring 96000 coulomb of charge in one hour through a potential difference of 50 V.

Answer:

The amount of heat (H) produced is given by the Joule's law of heating as

$$H = VIt$$

Where,

Voltage, $V = 50 \text{ V}$

Time, $t = 1 \text{ h} = 1 \times 60 \times 60 \text{ s}$

$$\text{Amount of current, } I = \frac{\text{Amount of charge}}{\text{Time of flow of charge}} = \frac{96000}{1 \times 60 \times 60} = \frac{80}{3} \text{ A}$$

$$H = 50 \times \frac{80}{3} \times 60 \times 60 = 4.8 \times 10^6 \text{ J}$$

Therefore, the heat generated is $4.8 \times 10^6 \text{ J}$.

Question 3:

An electric iron of resistance 20Ω takes a current of 5 A. Calculate the heat developed in 30 s.

Answer:

The amount of heat (H) produced is given by the joule's law of heating as

$$H = VIt$$

Where,

Current, $I = 5 \text{ A}$

Time, $t = 30 \text{ s}$

Voltage, $V = \text{Current} \times \text{Resistance} = 5 \times 20 = 100 \text{ V}$

ery
ou.....

$$H = 100 \times 5 \times 30 = 1.5 \times 10^4 \text{ J}$$

Therefore, the amount of heat developed in the electric iron is $1.5 \times 10^4 \text{ J}$.

Question 1:

What determines the rate at which energy is delivered by a current?

Answer:

The rate of consumption of electric energy in an electric appliance is called electric power. Hence, the rate at which energy is delivered by a current is the power of the appliance.

Question 2:

An electric motor takes 5 A from a 220 V line. Determine the power of the motor and the energy consumed in 2 h.

Answer:

Power (P) is given by the expression,

$$P = VI$$

Where,

Voltage, $V = 220 \text{ V}$

Current, $I = 5 \text{ A}$

$$P = 220 \times 5 = 1100 \text{ W}$$

Energy consumed by the motor = Pt

Where,

Time, $t = 2 \text{ h} = 2 \times 60 \times 60 = 7200 \text{ s}$

$$\therefore P = 1100 \times 7200 = 7.92 \times 10^6 \text{ J}$$

Therefore, power of the motor = 1100 W

Energy consumed by the motor = $7.92 \times 10^6 \text{ J}$

ery
ou.....

Question 1:

A piece of wire of resistance R is cut into five equal parts. These parts are then connected in parallel. If the equivalent resistance of this combination is R' , then the ratio R/R' is –

(a) $\frac{1}{25}$

(b) $\frac{1}{5}$

(c) 5

(d) 25

Answer:

(d) Resistance of a piece of wire is proportional to its length. A piece of wire has a resistance R . The wire is cut into five equal parts.

Therefore, resistance of each part = $\frac{R}{5}$

All the five parts are connected in parallel. Hence, equivalent resistance (R') is given as

$$\frac{1}{R'} = \frac{5}{R} + \frac{5}{R} + \frac{5}{R} + \frac{5}{R} + \frac{5}{R} = \frac{5+5+5+5+5}{R}$$

$$\frac{1}{R'} = \frac{25}{R}$$

$$\frac{R}{R'} = 25$$

Therefore, the ratio $\frac{R}{R'}$ is 25.

Question 2:

Which of the following terms does not represent electrical power in a circuit?

(a) I^2R

(b) IR^2

(c) VI

(d) $\frac{V^2}{R}$

Answer:

elivery

Be WITH you.....

(b) Electrical power is given by the expression, $P = VI$... (i)

According to Ohm's law, $V = IR$... (ii)

Where,

V = Potential difference

I = Current

R = Resistance

$$\therefore P = VI$$

From equation (i), it can be written

$$P = (IR) \times I$$

$$\therefore P = I^2 R$$

From equation (ii), it can be written

$$I = \frac{V}{R}$$

$$\therefore P = V \times \frac{V}{R}$$

$$P = \frac{V^2}{R}$$

$$\therefore P = VI = I^2 R = \frac{V^2}{R}$$

Power P cannot be expressed as IR^2 .

Question 3:

An electric bulb is rated 220 V and 100 W. When it is operated on 110 V, the power consumed will be –

- (a) 100 W
- (b) 75 W
- (c) 50 W
- (d) 25 W

Answer:

(d) Energy consumed by an appliance is given by the expression,

$$P = VI = \frac{V^2}{R}$$

elivery
E WITH YOU.....

$$R = \frac{V^2}{P}$$

Where,

Power rating, $P = 100 \text{ W}$

Voltage, $V = 220 \text{ V}$

$$\text{Resistance, } R = \frac{(220)^2}{100} = 484 \, \Omega$$

The resistance of the bulb remains constant if the supply voltage is reduced to 110 V. If the bulb is operated on 110 V, then the energy consumed by it is given by the expression for power as

$$\therefore P' = \frac{(V')^2}{R} = \frac{(110)^2}{484} = 25 \text{ W}$$

Therefore, the power consumed will be 25 W.

Question 4:

Two conducting wires of the same material and of equal lengths and equal diameters are first connected in series and then parallel in a circuit across the same potential difference. The ratio of heat produced in series and parallel combinations would be –

- (a) 1:2
- (b) 2:1
- (c) 1:4
- (d) 4:1

Answer:

(c) Heat produced in the circuit is inversely proportional to the resistance R .

Let R_s and R_p be the equivalent resistances of the wires if connected in series and parallel respectively. Hence, for same potential difference V , the ratio of heat produced in the circuit is given by

$$\frac{H_s}{H_p} = \frac{\frac{V^2}{R_s} t}{\frac{V^2}{R_p} t} = \frac{R_p}{R_s}$$

Where,

Heat produced in the series circuit = H_s

elivery

!E WITH YOU.....

Heat produced in the parallel circuit = H_p

Equivalent resistance, $R_s = R + R = 2R$

$$= \frac{1}{\frac{1}{R} + \frac{1}{R}} = \frac{R}{2}$$

Equivalent resistance, R_p

$$\text{Hence, ratio } \frac{H_s}{H_p} = \frac{\frac{R}{2}}{2R} = \frac{1}{4}$$

Therefore, the ratio of heat produced in series and parallel combinations is 1:4.

Question 5:

How is a voltmeter connected in the circuit to measure the potential difference between two points?

Answer:

To measure the potential difference between two points, a voltmeter should be connected in parallel to the points.

Question 6:

A copper wire has diameter 0.5 mm and resistivity of $1.6 \times 10^{-8} \Omega \text{ m}$. What will be the length of this wire to make its resistance 10 Ω ? How much does the resistance change if the diameter is doubled?

Answer:

Resistance (R) of a copper wire of length l and cross-section A is given by the expression,

$$R = \rho \frac{l}{A}$$

Where,

Resistivity of copper, $\rho = 1.6 \times 10^{-8} \Omega \text{ m}$

Area of cross-section of the wire, $A = \pi \left(\frac{\text{Diameter}}{2} \right)^2$

Diameter = 0.5 mm = 0.0005 m

Resistance, $R = 10 \Omega$

delivery
WITH YOU.....

$$l = \frac{RA}{\rho} = \frac{10 \times 3.14 \times \left(\frac{0.0005}{2}\right)^2}{1.6 \times 10^{-8}} = \frac{10 \times 3.14 \times 25}{4 \times 1.6} = 122.72 \text{ m}$$

Hence, length of the wire,

If the diameter of the wire is doubled, new diameter $= 2 \times 0.5 = 1 \text{ mm} = 0.001 \text{ m}$

Therefore, resistance R'

$$R' = \rho \frac{l}{A} = \frac{1.6 \times 10^{-8} \times 122.72}{\pi \left(\frac{1}{2} \times 10^{-3}\right)^2}$$

$$= \frac{1.6 \times 10^{-8} \times 122.72 \times 4}{3.14 \times 10^{-6}} = 250.2 \times 10^{-2} = 2.5 \Omega$$

Therefore, the length of the wire is 122.7 m and the new resistance is 2.5Ω .

Question 7:

The values of current I flowing in a given resistor for the corresponding values of potential difference V across the resistor are given below –

I (amperes) 0.5 1.0 2.0 3.0 4.0

V (volts) 1.6 3.4 6.7 10.2 13.2

Plot a graph between V and I and calculate the resistance of that resistor.

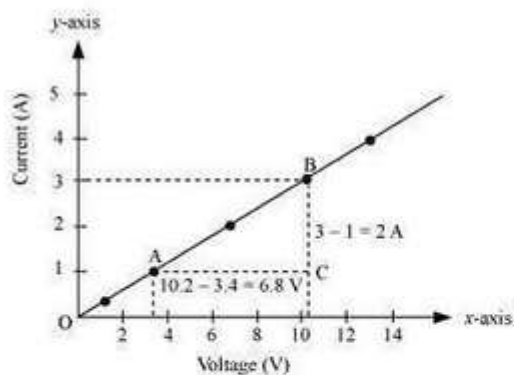
Answer:

The plot between voltage and current is called IV characteristic. The voltage is plotted on x -axis and current is plotted on y -axis. The values of the current for different values of the voltage are shown in the given table.

V (volts)	1.6	3.4	6.7	10.2	13.2
I (amperes)	0.5	1.0	2.0	3.0	4.0

The IV characteristic of the given resistor is plotted in the following figure.

very
TH YOU.....



The slope of the line gives the value of resistance (R) as,

$$\text{Slope} = \frac{1}{R} = \frac{BC}{AC} = \frac{2}{6.8}$$

$$R = \frac{6.8}{2} = 3.4 \, \Omega$$

Therefore, the resistance of the resistor is $3.4 \, \Omega$.

Question 8:

When a 12 V battery is connected across an unknown resistor, there is a current of 2.5 mA in the circuit. Find the value of the resistance of the resistor.

Answer:

Resistance (R) of a resistor is given by Ohm's law as,

$$V = IR$$

$$R = \frac{V}{I}$$

Where,

Potential difference, $V = 12 \, \text{V}$

Current in the circuit, $I = 2.5 \, \text{mA} = 2.5 \times 10^{-3} \, \text{A}$

$$R = \frac{12}{2.5 \times 10^{-3}} = 4.8 \times 10^3 \, \Omega = 4.8 \, \text{k}\Omega$$

Therefore, the resistance of the resistor is $4.8 \, \text{k}\Omega$.

delivery

? WITH YOU.....

Question 9:

A battery of 9 V is connected in series with resistors of 0.2 Ω , 0.3 Ω , 0.4 Ω , 0.5 Ω and 12 Ω , respectively. How much current would flow through the 12 Ω resistor?

Answer:

There is no current division occurring in a series circuit. Current flow through the component is the same, given by Ohm's law as

$$V = IR$$

$$I = \frac{V}{R}$$

Where,

R is the equivalent resistance of resistances 0.2 Ω , 0.3 Ω , 0.4 Ω , 0.5 Ω , and 12 Ω . These are connected in series. Hence, the sum of the resistances will give the value of R .

$$R = 0.2 + 0.3 + 0.4 + 0.5 + 12 = 13.4 \Omega$$

Potential difference, $V = 9 \text{ V}$

$$I = \frac{9}{13.4} = 0.671 \text{ A}$$

Therefore, the current that would flow through the 12 Ω resistor is 0.671 A.

Question 10:

How many 176 Ω resistors (in parallel) are required to carry 5 A on a 220 V line?

Answer:

For x number of resistors of resistance 176 Ω , the equivalent resistance of the resistors connected in parallel is given by Ohm's law as

$$V = IR$$

$$R = \frac{V}{I}$$

Where,

Supply voltage, $V = 220 \text{ V}$

Current, $I = 5 \text{ A}$

Equivalent resistance of the combination = R , given as

$$\frac{1}{R} = x \times \left(\frac{1}{176} \right)$$

$$R = \frac{176}{x}$$

elivery

Be WITH you.....

From Ohm's law,

$$\frac{V}{I} = \frac{176}{x}$$
$$x = \frac{176 \times I}{V} = \frac{176 \times 5}{220} = 4$$

Therefore, four resistors of $176 \, \Omega$ are required to draw the given amount of current.

Question 11:

Show how you would connect three resistors, each of resistance $6 \, \Omega$, so that the combination has a resistance of (i) $9 \, \Omega$, (ii) $4 \, \Omega$.

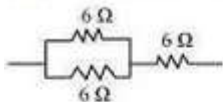
Answer:

If we connect the resistors in series, then the equivalent resistance will be the sum of the resistors, i.e., $6 \, \Omega + 6 \, \Omega + 6 \, \Omega = 18 \, \Omega$, which is not desired. If we connect the resistors in parallel, then the equivalent resistance will be

$$\frac{6}{2} = 3 \, \Omega, \text{ which is also not desired.}$$

Hence, we should either connect the two resistors in series or parallel.

(i) Two resistors in parallel

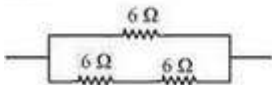


Two $6 \, \Omega$ resistors are connected in parallel. Their equivalent resistance will be

$$\frac{1}{\frac{1}{6} + \frac{1}{6}} = \frac{6 \times 6}{6 + 6} = 3 \, \Omega$$

The third $6 \, \Omega$ resistor is in series with $3 \, \Omega$. Hence, the equivalent resistance of the circuit is $6 \, \Omega + 3 \, \Omega = 9 \, \Omega$.

(ii) Two resistors in series



Two $6 \, \Omega$ resistors are in series. Their equivalent resistance will be the sum $6 + 6 = 12 \, \Omega$

delivery
WITH YOU.....

The third $6\ \Omega$ resistor is in parallel with $12\ \Omega$. Hence, equivalent resistance will be

$$\frac{1}{\frac{1}{12} + \frac{1}{6}} = \frac{12 \times 6}{12 + 6} = 4\ \Omega$$

Therefore, the total resistance is $4\ \Omega$.

Question 12:

Several electric bulbs designed to be used on a 220 V electric supply line, are rated 10 W . How many lamps can be connected in parallel with each other across the two wires of 220 V line if the maximum allowable current is 5 A ?

Answer:

Resistance R_1 of the bulb is given by the expression,

$$P_1 = \frac{V^2}{R_1}$$

$$R_1 = \frac{V^2}{P_1}$$

Where,

Supply voltage, $V = 220\text{ V}$

Maximum allowable current, $I = 5\text{ A}$

Rating of an electric bulb $P_1 = 10\text{ W}$

$$R_1 = \frac{(220)^2}{10} = 4840\ \Omega$$

According to Ohm's law,

$$V = IR$$

Where,

R is the total resistance of the circuit for x number of electric bulbs

$$R = \frac{V}{I} = \frac{220}{5} = 44\ \Omega$$

Resistance of each electric bulb, $R_1 = 4840\ \Omega$

ivery

TH YOU.....

$$\therefore \frac{1}{R} = \frac{1}{R_1} + \frac{1}{R_1} + \dots \text{up to } x \text{ times}$$

$$\frac{1}{R} = \frac{1}{R_1} \times x$$

$$x = \frac{R_1}{R} = \frac{4840}{44} = 110$$

Therefore, 110 electric bulbs are connected in parallel.

Question 13:

A hot plate of an electric oven connected to a 220 V line has two resistance coils A and B, each of 24Ω resistances, which may be used separately, in series, or in parallel. What are the currents in the three cases?

Answer:

Supply voltage, $V = 220 \text{ V}$

Resistance of one coil, $R = 24 \Omega$

(i) Coils are used separately

According to Ohm's law,

$$V = I_1 R_1$$

Where,

I_1 is the current flowing through the coil

$$I_1 = \frac{V}{R_1} = \frac{220}{24} = 9.166 \text{ A}$$

Therefore, 9.16 A current will flow through the coil when used separately.

(ii) Coils are connected in series

Total resistance, $R_2 = 24 \Omega + 24 \Omega = 48 \Omega$

According to Ohm's law,

$$V = I_2 R_2$$

Where,

I_2 is the current flowing through the series circuit

$$I_2 = \frac{V}{R_2} = \frac{220}{48} = 4.58 \text{ A}$$

delivery
? WITH YOU.....

Therefore, 4.58 A current will flow through the circuit when the coils are connected in series.

(iii) Coils are connected in parallel

$$\frac{1}{\frac{1}{24} + \frac{1}{24}} = \frac{24}{2} = 12 \Omega$$

Total resistance, R_3 is given as

According to Ohm's law,

$$V = I_3 R_3$$

Where,

I_3 is the current flowing through the circuit

$$I_3 = \frac{V}{R_3} = \frac{220}{12} = 18.33 \text{ A}$$

Therefore, 18.33 A current will flow through the circuit when coils are connected in parallel.

Question 14:

Compare the power used in the 2Ω resistor in each of the following circuits: (i) a 6 V battery in series with 1Ω and 2Ω resistors, and (ii) a 4 V battery in parallel with 12Ω and 2Ω resistors.

Answer:

(i) Potential difference, $V = 6 \text{ V}$

1Ω and 2Ω resistors are connected in series. Therefore, equivalent resistance of the circuit, $R = 1 + 2 = 3 \Omega$

According to Ohm's law,

$$V = IR$$

Where,

I is the current through the circuit

$$I = \frac{6}{3} = 2 \text{ A}$$

This current will flow through each component of the circuit because there is no division of current in series circuits. Hence, current flowing through the 2Ω resistor is 2 A . Power is given by the expression,

ivery
JTH YOU.....

$$P = (I)^2 R = (2)^2 \times 2 = 8 \text{ W}$$

(ii) Potential difference, $V = 4 \text{ V}$

12Ω and 2Ω resistors are connected in parallel. The voltage across each component of a parallel circuit remains the same. Hence, the voltage across 2Ω resistor will be 4 V .

Power consumed by 2Ω resistor is given by

$$P = \frac{V^2}{R} = \frac{4^2}{2} = 8 \text{ W}$$

Therefore, the power used by 2Ω resistor is 8 W .

Question 15:

Two lamps, one rated 100 W at 220 V , and the other 60 W at 220 V , are connected in parallel to electric mains supply. What current is drawn from the line if the supply voltage is 220 V ?

Answer:

Both the bulbs are connected in parallel. Therefore, potential difference across each of them will be 220 V , because no division of voltage occurs in a parallel circuit.

Current drawn by the bulb of rating 100 W is given by,

$$\text{Power} = \text{Voltage} \times \text{Current}$$

$$\text{Current} = \frac{\text{Power}}{\text{Voltage}} = \frac{100}{220} \text{ A}$$

Similarly, current drawn by the bulb of rating 60 W is given by,

$$\text{Power} = \text{Voltage} \times \text{Current}$$

$$\text{Current} = \frac{\text{Power}}{\text{Voltage}} = \frac{60}{220} \text{ A}$$

$$\text{Hence, current drawn from the line} = \frac{100}{220} + \frac{60}{220} = 0.727 \text{ A}$$

Question 15:

Two lamps, one rated 100 W at 220 V , and the other 60 W at 220 V , are connected in parallel to electric mains supply. What current is drawn from the line if the supply voltage is 220 V ?

Answer:

Both the bulbs are connected in parallel. Therefore, potential difference across each of them will be 220 V , because no division of voltage occurs in a parallel circuit.

delivery

WITH YOU.....

Current drawn by the bulb of rating 100 W is given by,

$$\text{Power} = \text{Voltage} \times \text{Current}$$

$$\text{Current} = \frac{\text{Power}}{\text{Voltage}} = \frac{100}{220} \text{ A}$$

Similarly, current drawn by the bulb of rating 60 W is given by,

$$\text{Power} = \text{Voltage} \times \text{Current}$$

$$\text{Current} = \frac{\text{Power}}{\text{Voltage}} = \frac{60}{220} \text{ A}$$

$$\text{Hence, current drawn from the line} = \frac{100}{220} + \frac{60}{220} = 0.727 \text{ A}$$

Question 16:

Which uses more energy, a 250 W TV set in 1 hr, or a 1200 W toaster in 10 minutes?

Answer:

Energy consumed by an electrical appliance is given by the expression,

$$H = Pt$$

Where,

Power of the appliance = P

Time = t

Energy consumed by a TV set of power 250 W in 1 h = $250 \times 3600 = 9 \times 10^5 \text{ J}$

Energy consumed by a toaster of power 1200 W in 10 minutes = 1200×600

$$= 7.2 \times 10^5 \text{ J}$$

Therefore, the energy consumed by a 250 W TV set in 1 h is more than the energy consumed by a toaster of power 1200 W in 10 minutes.

Question 17:

An electric heater of resistance 8Ω draws 15 A from the service mains 2 hours. Calculate the rate at which heat is developed in the heater.

Answer:

Rate of heat produced by a device is given by the expression for power as

$$P = I^2 R$$

Where,

Resistance of the electric heater, $R = 8 \Omega$

Current drawn, $I = 15 \text{ A}$

elivery

! WITH YOU.....

$$P = (15)^2 \times 8 = 1800 \text{ J/s}$$

Therefore, heat is produced by the heater at the rate of 1800 J/s.

Question 18:

Explain the following.

- (a) Why is the tungsten used almost exclusively for filament of electric lamps?
- (b) Why are the conductors of electric heating devices, such as bread-toasters and electric irons, made of an alloy rather than a pure metal?
- (c) Why is the series arrangement not used for domestic circuits?
- (d) How does the resistance of a wire vary with its area of cross-section?
- (e) Why are copper and aluminium wires usually employed for electricity transmission?

Answer:

(a) The melting point and resistivity of tungsten are very high. It does not burn readily at a high temperature. The electric lamps glow at very high temperatures. Hence, tungsten is mainly used as heating element of electric bulbs.

(b) The conductors of electric heating devices such as bread toasters and electric irons are made of alloy because resistivity of an alloy is more than that of metals. It produces large amount of heat.

(c) There is voltage division in series circuits. Each component of a series circuit receives a small voltage for a large supply voltage. As a result, the amount of current decreases and the device becomes hot. Hence, series arrangement is not used in domestic circuits.

(d) Resistance (R) of a wire is inversely proportional to its area of cross-section (A), i.e.,

$$R \propto \frac{1}{A}$$

(e) Copper and aluminium wires have low resistivity. They are good conductors of electricity. Hence, they are usually employed for electricity transmission.

Try
on.....

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 13 Magnetic Effects of Electric Current Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 13 Magnetic Effects of Electric Current Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 13 Magnetic Effects of Electric Current NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

Why does a compass needle get deflected when brought near a bar magnet?

Answer:

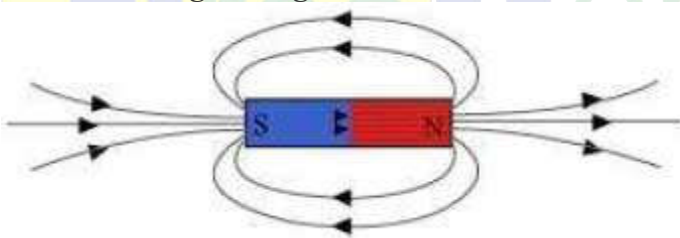
A compass needle is a small bar magnet. When it is brought near a bar magnet, its magnetic field lines interact with that of the bar magnet. Hence, a compass needle shows a deflection when brought near the bar magnet.

Question 1:

Draw magnetic field lines around a bar magnet.

Answer:

Magnetic field lines of a bar magnet emerge from the north pole and terminate at the south pole. Inside the magnet, the field lines emerge from the south pole and terminate at the north pole, as shown in the given figure.



Question 2:

List the properties of magnetic lines of force.

Answer:

The properties of magnetic lines of force are as follows.

- (a) Magnetic field lines emerge from the north pole.
- (b) They merge at the south pole.
- (c) The direction of field lines inside the magnet is from the south pole to the north pole.
- (d) Magnetic lines do not intersect with each other.

Question 3:

Why don't two magnetic lines of force intersect each other?

Answer:

If two field lines of a magnet intersect, then at the point of intersection, the compass needle points in two different directions. This is not possible. Hence, two field lines do not intersect each other.

Question 1:

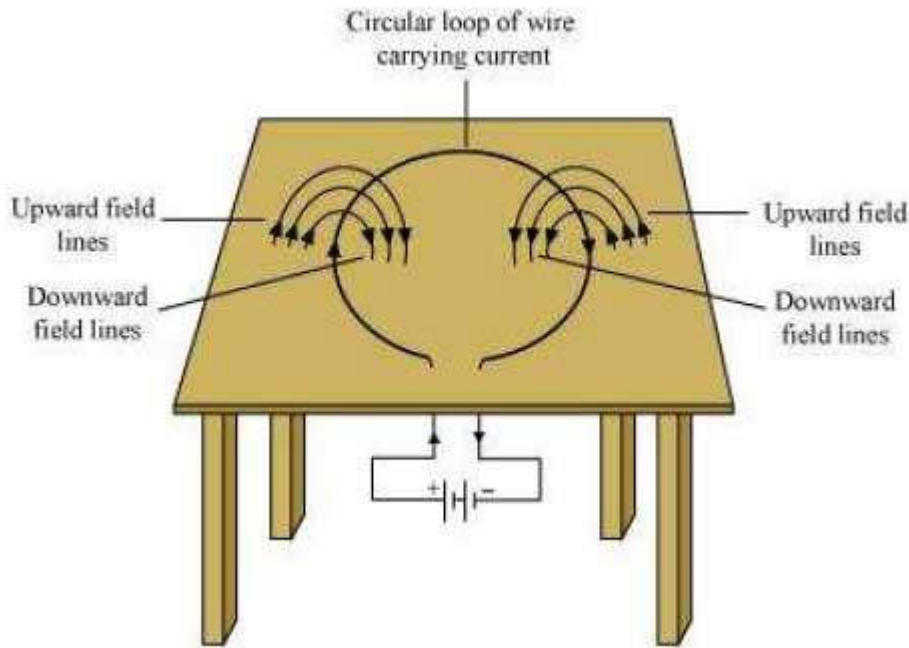
Consider a circular loop of wire lying in the plane of the table. Let the current pass through the loop clockwise. Apply the right-hand rule to find out the direction of the magnetic field inside and outside the loop.

Answer:

Inside the loop = Pierce inside the table

Outside the loop = Appear to emerge out from the table

For downward direction of current flowing in the circular loop, the direction of magnetic field lines will be as if they are emerging from the table outside the loop and merging in the table inside the loop. Similarly, for upward direction of current flowing in the circular loop, the direction of magnetic field lines will be as if they are emerging from the table outside the loop and merging in the table inside the loop, as shown in the given figure.

**Question 3:**

How much energy is given to each coulomb of charge passing through a 6 V battery?

Answer:

The energy given to each coulomb of charge is equal to the amount of work required to move it. The amount of work is given by the expression,

$$\text{Potential difference} = \frac{\text{Work done}}{\text{Charge}}$$

Potential difference =

$$\text{Work Done} = \text{Potential Difference} \times \text{Charge}$$

Where,

Charge = 1 C

Potential difference = 6 V

Work Done = $6 \times 1 \text{ J} = 6 \text{ J}$

Therefore, 6 J of energy is given to each coulomb of charge passing through a battery of 6 V.

Question 3:

Choose the correct option.

The magnetic field inside a long straight solenoid-carrying current

- (a) is zero
- (b) decreases as we move towards its end
- (c) increases as we move towards its end

(d) is the same at all points

Answer:

(d) The magnetic field inside a long, straight, current-carrying solenoid is uniform. It is the same at all points inside the solenoid.

Question 1:

Which of the following property of a proton can change while it moves freely in a magnetic field?

(There may be more than one correct answer.)

(a) mass

(b) speed

(c) velocity

(d) momentum

Answer:

(c) and (d)

When a proton enters in a region of magnetic field, it experiences a magnetic force. As a result of the force, the path of the proton becomes circular. Hence, its velocity and momentum change.

Question 2:

In Activity 13.7 (page: 230), how do we think the displacement of rod AB will be affected if (i) current in rod AB is increased: (ii) a stronger horse-shoe magnet is used: and (iii) length of the rod AB is increased?

Answer:

A current-carrying conductor placed in a magnetic field experiences a force. The magnitude of force increases with the amount of current, strength of the magnetic field, and the length of the conductor. Hence, the magnetic force exerted on rod AB and its deflection will increase if

(i) current in rod AB is increased

(ii) a stronger horse-shoe magnet is used

(iii) length of rod AB is increased

Question 3:

A positively-charged particle (alpha-particle) projected towards west is deflected towards north by a magnetic field. The direction of magnetic field is

(a) towards south (b) towards east

(c) downward (d) upward

Answer:

(d) The direction of the magnetic field can be determined by the Fleming's left hand rule. According to this rule, if we arrange the thumb, the centre finger, and the forefinger of the left hand at right angles to each other, then the thumb points towards the direction of the magnetic force, the centre finger gives the direction of current, and the forefinger points in the direction of magnetic field. Since the direction of positively charged alpha particle is towards west, the direction of current will be the same i.e., towards west. Again, the direction of magnetic force is towards north. Hence, according to Fleming's left hand rule, the direction of magnetic field will be upwards.

Question 1:

State Fleming's left-hand rule.

Answer:

Fleming's left hand rule states that if we arrange the thumb, the centre finger, and the forefinger of the left hand at right angles to each other, then the thumb points towards the direction of the magnetic force, the centre finger gives the direction of current, and the forefinger points in the direction of magnetic field.

Question 2:

What is the principle of an electric motor?

Answer:

The working principle of an electric motor is based on the magnetic effect of current. A current-carrying loop experiences a force and rotates when placed in a magnetic field.

The direction of rotation of the loop is given by the Fleming's left-hand rule.

Question 3:

What is the role of the split ring in an electric motor?

Answer:

The split ring in the electric motor acts as a commutator. The commutator reverses the direction of current flowing through the coil after each half rotation of the coil. Due to this reversal of the current, the coil continues to rotate in the same direction.

Question 1:

Explain different ways to induce current in a coil.

Answer:

The different ways to induce current in a coil are as follows:

(a) If a coil is moved rapidly between the two poles of a horse-shoe magnet, then an electric current is induced in the coil.

(b) If a magnet is moved relative to a coil, then an electric current is induced in the coil.

Question 1:

State the principle of an electric generator.

Answer:

An electric generator works on the principle of electromagnetic induction. It generates electricity by rotating a coil in a magnetic field.

Question 2:

Name some sources of direct current.

Answer:

Some sources of direct current are cell, DC generator, etc.

Question 3:

Which sources produce alternating current?

Answer:

AC generators, power plants, etc., produce alternating current.

Question 4:

Choose the correct option.

A rectangular coil of copper wires is rotated in a magnetic field. The direction of the induced current changes once in each

(a) two revolutions (b) one revolution

(c) half revolution (d) one-fourth revolution

Answer:

(c) When a rectangular coil of copper is rotated in a magnetic field, the direction of the induced current in the coil changes once in each half revolution. As a result, the direction of current in the coil remains the same.

Question 1:

Name two safety measures commonly used in electric circuits and appliances.

Answer:

Two safety measures commonly used in electric circuits and appliances are as follows:

- (i) Each circuit must be connected with an electric fuse. This prevents the flow of excessive current through the circuit. When the current passing through the wire exceeds the maximum limit of the fuse element, the fuse melts to stop the flow of current through that circuit, hence protecting the appliances connected to the circuit.
- (ii) Earthing is a must to prevent electric shocks. Any leakage of current in an electric appliance is transferred to the ground and people using the appliance do not get the shock.

Question 2:

An electric oven of 2 kW is operated in a domestic electric circuit (220 V) that has a current rating of 5

A. What result do you expect? Explain.

Answer:

Current drawn by the electric oven can be obtained by the expression,

$$P = VI$$

$$I = \frac{P}{V}$$

Where,

Current = I

Power of the oven, $P = 2 \text{ kW} = 2000 \text{ W}$

Voltage supplied, $V = 220 \text{ V}$

$$I = \frac{2000}{220} = 9.09 \text{ A}$$

Hence, the current drawn by the electric oven is 9.09 A, which exceeds the safe limit of the circuit. Fuse element of the electric fuse will melt and break the circuit.

Question 3:

What precaution should be taken to avoid the overloading of domestic electric circuits?

Answer:

The precautions that should be taken to avoid the overloading of domestic circuits are as follows:

- (a) Too many appliances should not be connected to a single socket.
- (b) Too many appliances should not be used at the same time.
- (c) Faulty appliances should not be connected in the circuit.
- (d) Fuse should be connected in the circuit.

Question 1:

Which of the following correctly describes the magnetic field near a long straight wire?

- (a) The field consists of straight lines perpendicular to the wire
- (b) The field consists of straight lines parallel to the wire
- (c) The field consists of radial lines originating from the wire
- (d) The field consists of concentric circles centred on the wire

Answer:

(d) The magnetic field lines, produced around a straight current-carrying conductor, are concentric circles. Their centres lie on the wire.

Question 2:

The phenomenon of electromagnetic induction is

- (a) the process of charging a body
- (b) the process of generating magnetic field due to a current passing through a coil
- (c) producing induced current in a coil due to relative motion between a magnet and the coil
- (d) the process of rotating a coil of an electric motor

Answer:

(c) When a straight coil and a magnet are moved relative to each other, a current is induced in the coil. This phenomenon is known as electromagnetic induction.

Question 3:

The device used for producing electric current is called a

- (a) generator
- (b) galvanometer
- (c) ammeter
- (d) motor

Answer:

(a) An electric generator produces electric current. It converts mechanical energy into electricity.

Question 4:

The essential difference between an AC generator and a DC generator is that

- (a) AC generator has an electromagnet while a DC generator has permanent magnet.
- (b) DC generator will generate a higher voltage.
- (c) AC generator will generate a higher voltage.
- (d) AC generator has slip rings while the DC generator has a commutator.

Answer:

(d) An AC generator has two rings called slip rings. A DC generator has two half rings called commutator. This is the main difference between both the types of generators.

Question 5:

At the time of short circuit, the current in the circuit

- (a) reduces substantially
- (b) does not change
- (c) increases heavily
- (d) vary continuously

Answer:

(c) When two naked wires of an electric circuit touch each other, the amount of current that is flowing in the circuit increases abruptly. This causes short-circuit.

Question 6:

State whether the following statements are true or false.

- (a) An electric motor converts mechanical energy into electrical energy.
- (b) An electric generator works on the principle of electromagnetic induction.
- (c) The field at the centre of a long circular coil carrying current will be parallel straight lines.
- (d) A wire with a green insulation is usually the live wire of an electric supply.

Answer:

- (a) False

An electric motor converts electrical energy into mechanical energy.

- (b) True

A generator is an electric device that generates electricity by rotating a coil in a magnetic field. It

works on the principle of electromagnetic induction.

(c) True

A long circular coil is a long solenoid. The magnetic field lines inside the solenoid are parallel lines.

(d) False

Live wire has red insulation cover, whereas earth wire has green insulation colour in the domestic circuits.

Question 7:

List three sources of magnetic fields.

Answer:

Three sources of magnetic fields are as follows:

(a) Current-carrying conductors

(b) Permanent magnets

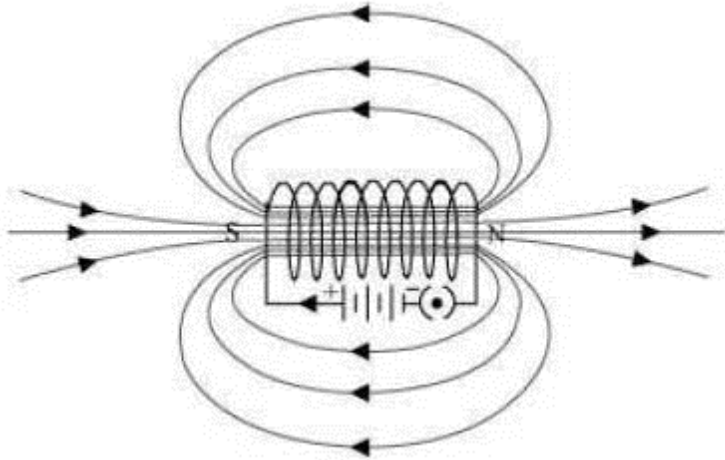
(c) Electromagnets

Question 8:

How does a solenoid behave like a magnet? Can you determine the north and south poles of a current-carrying solenoid with the help of a bar magnet? Explain.

Answer:

A solenoid is a long coil of circular loops of insulated copper wire. Magnetic field lines are produced around the solenoid when a current is allowed to flow through it. The magnetic field produced by it is similar to the magnetic field of a bar magnet. The field lines produced in a current-carrying solenoid is shown in the following figure.



In the above figure, when the north pole of a bar magnet is brought near the end connected to the negative terminal of the battery, the solenoid repels the bar magnet. Since like poles repel each other, the end connected to the negative terminal of the battery behaves as the north pole of the solenoid and the other end behaves as a south pole. Hence, one end of the solenoid behaves as a north pole and the other end behaves as a south pole.

Question 9:

When is the force experienced by a current-carrying conductor placed in a magnetic field largest?

Answer:

The force experienced by a current-carrying conductor is the maximum when the direction of current is perpendicular to the direction of the magnetic field.

Question 10:

Imagine that you are sitting in a chamber with your back to one wall. An electron beam, moving

horizontally from back wall towards the front wall, is deflected by a strong magnetic field to your right side. What is the direction of magnetic field?

Answer:

The direction of magnetic field is given by Fleming's left hand rule. Magnetic field inside the chamber will be perpendicular to the direction of current (opposite to the direction of electron) and direction of deflection/force i.e., either upward or downward. The direction of current is from the front wall to the back wall because negatively charged electrons are moving from back wall to the front wall. The direction of magnetic force is rightward.

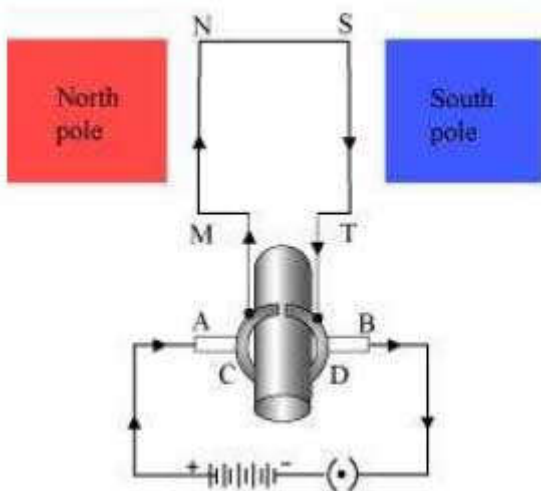
Hence, using Fleming's left hand rule, it can be concluded that the direction of magnetic field inside the chamber is downward.

Question 11:

Draw a labelled diagram of an electric motor. Explain its principle and working. What is the function of a split ring in an electric motor?

Answer:

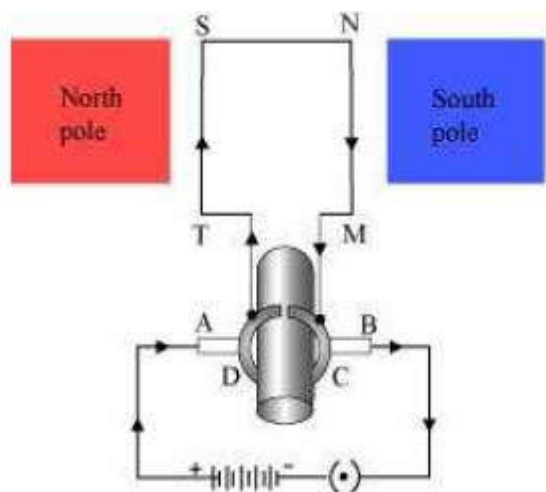
An electric motor converts electrical energy into mechanical energy. It works on the principle of the magnetic effect of current. A current-carrying coil rotates in a magnetic field. The following figure shows a simple electric motor.



When a current is allowed to flow through the coil MNST by closing the switch, the coil starts rotating anti-clockwise. This happens because a downward force acts on length MN and at the same time, an upward force acts on length ST. As a result, the coil rotates anti-clockwise.

Current in the length MN flows from M to N and the magnetic field acts from left to right, normal to length MN. Therefore, according to Fleming's left hand rule, a downward force acts on the length MN. Similarly, current in the length ST flows from S to T and the magnetic field acts from left to right, normal to the flow of current. Therefore, an upward force acts on the length ST. These two forces cause the coil to rotate anti-clockwise.

After half a rotation, the position of MN and ST interchange. The half-ring D comes in contact with brush A and half-ring C comes in contact with brush B. Hence, the direction of current in the coil MNST gets reversed.



The current flows through the coil in the direction TSNM. The reversal of current through the coil MNST repeats after each half rotation. As a result, the coil rotates unidirectional. The split rings help to reverse the direction of current in the circuit. These are called the commutator.

Question 12:

Name some devices in which electric motors are used?

Answer:

Some devices in which electric motors are used are as follows:

- (a) Water pumps
- (b) Electric fans
- (c) Electric mixers
- (d) Washing machines

Question 13:

A coil of insulated copper wire is connected to a galvanometer. What will happen if a bar magnet is (i) pushed into the coil, (ii) withdrawn from inside the coil, (iii) held stationary inside the coil?

Answer:

A current induces in a solenoid if a bar magnet is moved relative to it. This is the principle of electromagnetic induction.

(i) When a bar magnet is pushed into a coil of insulated copper wire, a current is induced momentarily in the coil. As a result, the needle of the galvanometer deflects momentarily in a particular direction.

(ii) When the bar magnet is withdrawn from inside the coil of the insulated copper wire, a current is again induced momentarily in the coil in the opposite direction. As a result, the needle of the galvanometer deflects momentarily in the opposite direction.

(iii) When a bar magnet is held stationary inside the coil, no current will be induced in the coil. Hence, galvanometer will show no deflection.

Question 14:

Two circular coils A and B are placed closed to each other. If the current in the coil A is changed, will some current be induced in the coil B? Give reason.

Answer:

Two circular coils A and B are placed close to each other. When the current in coil A is changed, the magnetic field associated with it also changes. As a result, the magnetic field around coil B also changes. This change in magnetic field lines around coil B induces an electric current in it. This is called electromagnetic induction.

Question 15:

State the rule to determine the direction of a (i) magnetic field produced around a straight conductor-carrying current, (ii) force experienced by a current-carrying straight conductor placed in a magnetic field which is perpendicular to it, and (iii) current induced in a coil due to its rotation in a magnetic field.

Answer:

(i) Maxwell's right hand thumb rule

(ii) Fleming's left hand rule

(iii) Fleming's right hand rule

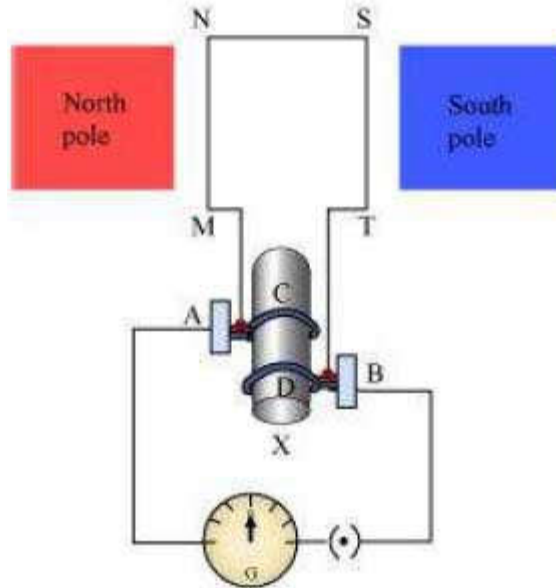
Question 16:

Explain the underlying principle and working of an electric generator by drawing a labelled diagram. What is the function of brushes?

Answer:

An electric generator converts mechanical energy into electrical energy.

The principle of working of an electric generator is that when a loop is moved in a magnetic field, an electric current is induced in the coil. It generates electricity by rotating a coil in a magnetic field. The following figure shows a simple AC generator.



MNST → Rectangular coil

A and B → Brushes

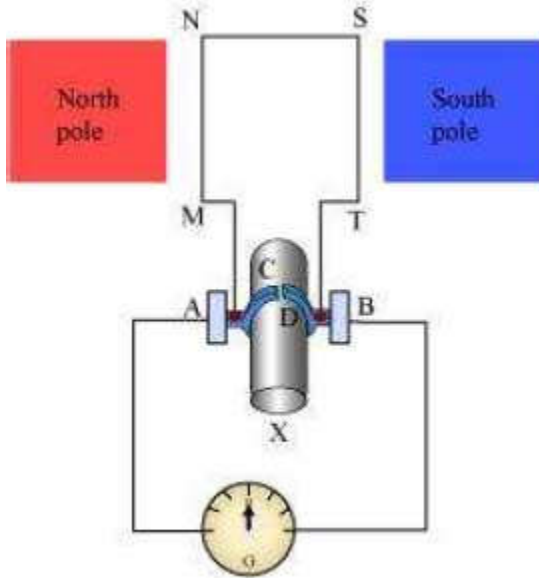
C and D → Two slip rings

X → Axle, G → Galvanometer

If axle X is rotated clockwise, then the length MN moves upwards while length ST moves downwards. Since the lengths MN and ST are moving in a magnetic field, a current will be induced in both of them due to electromagnetic induction. Length MN is moving upwards and the magnetic field acts from left to right. Hence, according to Fleming's right hand rule, the direction of induced current will be from M to N. Similarly, the direction of induced current in the length ST will be from S to T.

The direction of current in the coil is MNST. Hence, the galvanometer shows a deflection in a particular direction. After half a rotation, length MN starts moving down whereas length ST starts moving upward. The direction of the induced current in the coil gets reversed as TSNM. As the direction of current gets reversed after each half rotation, the produced current is called an alternating current (AC).

To get a unidirectional current, instead of two slip rings, two split rings are used, as shown in the following figure.



In this arrangement, brush A always remains in contact with the length of the coil that is moving up whereas brush B always remains in contact with the length that is moving down. The split rings C and D act as a commutator.

The direction of current induced in the coil will be MNST for the first rotation and TSNM in the second half of the rotation. Hence, a unidirectional current is produced from the generator called DC generator. The current is called AC current.

Question 17:

When does an electric short circuit occur?

Answer:

If the resistance of an electric circuit becomes very low, then the current flowing through the circuit becomes very high. This is caused by connecting too many appliances to a single socket or connecting high power rating appliances to the light circuits. This results in a short circuit.

When the insulation of live and neutral wires undergoes wear and tear and then touches each other, the current flowing in the circuit increases abruptly. Hence, a short circuit occurs.

Question 18:

What is the function of an earth wire? Why is it necessary to earth metallic appliances?

Answer:

The metallic body of electric appliances is connected to the earth by means of earth wire so that any leakage of electric current is transferred to the ground. This prevents any electric shock to the user. That is why earthing of the electrical appliances is necessary.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 14 Sources of Energy Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 14 Sources of Energy Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 14 Sources of Energy NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

What is a good source of energy?

Answer:

A good source of energy fulfils the following criteria:

- (I) It produces a lot of heat per unit mass.
- (II) It does a huge amount of work per unit mass.
- (III) It is easily accessible.
- (IV) It is easy to store and transport.
- (V) It is economical.
- (VI) It produces less amount of smoke.

Question 2:

What is a good fuel?

Answer:

A good fuel produces a huge amount of heat on burning, does not produce a lot of smoke, and is easily available.

Question 3:

If you could use any source of energy for heating your food, which one would you use and why?

Answer:

Natural gas can be used for heating and cooking food because it is a clean source of energy. It does not produce huge amount of smoke on burning. Although it is highly inflammable, it is easy to use, transport, and it produces a huge amount of heat on burning.

Question 1:

What are the disadvantages of fossil fuels?

Answer:

The disadvantages of fossil fuels are as follows:

- (a) Burning of coal and petroleum produces a lot of pollutants causing air pollution.
- (b) Fossil fuels release oxides of carbon, nitrogen, sulphur, etc. that cause acid rain, which affects the soil fertility and potable water.
- (c) Burning of fossil fuels produce gases such as carbon dioxide that causes global warming.

Question 2:

Why are we looking at alternate sources of energy?

Answer:

Fossil fuels, which have been traditionally used by human beings as an energy sources, are non-renewable sources of energy. These sources of energy are limited and cannot replenish on their own. They are being consumed at a large rate. If this rate of consumption continues, then the fossil fuels would be exhausted from the Earth.

Therefore, we have to conserve the energy sources. Hence, we should look for alternate sources of energy.

Question 3:

How has the traditional use of wind and water energy been modified for our convenience?

Answer:

Traditionally, waterfalls were used as a source of potential energy which was converted to electricity with the help of turbines. Since waterfalls are few in number, water dams have been constructed in large numbers. Nowadays, hydro-dams are used in order to harness potential energy of stored water. In water dams, water falls from a height on the turbine, which produces electricity.

Earlier, the windmills were used to harness wind energy to do mechanical work such as lifting/drawing water from a well. Today, windmills are used to generate electricity. In windmills, the kinetic energy of wind is harnessed and converted into electricity. The rotatory motion of the blades turns the turbine of the electric generator to generate electricity.

Question 1:

What kind of mirror – concave, convex or plain – would be best suited for use in a solar cooker? Why?

Answer:

A solar cooker uses heat of the sunlight to cook and heat food. A mirror is used in order to reflect and focus sunlight at a point. A concave mirror is used in a solar cooker for this purpose. The mirror focuses all the incident sunlight at a point. The temperature at that point increases, thereby cooking and heating the food placed at that point.

Question 2:

What are the limitations of the energy that can be obtained from the oceans?

Answer:

The forms of energy that can be obtained from the ocean are tidal energy, wave energy, and ocean thermal energy. There are several limitations in order to harness these energies.

(i) Tidal energy depends on the relative positioning of the Earth, moon, and the Sun.

(ii) High dams are required to be built to convert tidal energy into electricity.

(iii) Very strong waves are required to obtain electricity from wave energy.

(iv) To harness ocean thermal energy efficiently, the difference in the temperature of surface water (hot) and the water at depth (cold) must be 20°C or more.

Question 3:

What is geothermal energy?

Answer:

Geothermal power plants use heat of the Earth to generate electricity. This heat energy of the Earth is known as geothermal energy.

When there are geological changes, the molten rocks present in the core of the earth are pushed to the earth's crust. This forms regions of hot spot. Steam is generated when the underground water comes in contact with these hot spots forming hot springs. This trapped steam is used to generate electricity in the geothermal power plants.

Question 4:

What are the advantages of nuclear energy?

Answer:

The advantages of nuclear energy are as follows:

(a) Large amount of energy is produced per unit mass.

(b) It does not produce smoke. It is a clean energy.

(c) Fission of one atom of uranium produces 10 million times the energy released by burning of one atom of carbon.

(d) Fusion of four hydrogen atoms produces huge amount of energy approximately equal to 27 MeV.

Question 1:

Can any source of energy be pollution-free? Why or why not?

Answer:

No source of energy can be pollution-free. It is considered that solar cells are pollution-free.

However, even their making causes environmental damage indirectly.

Also, in the case of nuclear energy, there is no waste produced after the fusion reactions. However, it is not totally pollution-free. To start the fusion reactions, approximately 10⁷ K temperature is required, which is provided by fission reactions. The wastes released from fission reactions are very hazardous. Hence, no source of energy is pollution-free.

Question 2:

Hydrogen has been used as a rocket fuel. Would you consider it a cleaner fuel than CNG? Why or why not?

Answer:

Hydrogen gas is cleaner than CNG. CNG contains hydrocarbons. Therefore, it has carbon contents. Carbon is a form of pollutant present in CNG. On the other hand, hydrogen is waste-free. The fusion of hydrogen does not produce any waste. Hence, hydrogen is cleaner than CNG.

Question 1:

Name two energy sources that you would consider to be renewable. Give reasons for your choices.

Answer:

Two renewable sources of energy are as follows:

(a) Sun: The energy derived from the Sun is known as solar energy. Solar energy is produced by the fusion of hydrogen into helium, fusion of helium into other heavy elements, and so on. A large amount of hydrogen and helium is present in the Sun. Therefore, solar energy can replenish on its own. The Sun has 5 billion years more to burn. Hence, solar energy is a renewable source of energy.

(b) Wind: Wind energy is derived from air blowing with high speed. Wind energy is harnessed by windmills in order to generate electricity. Air blows because of uneven heating of the Earth. Since the heating of the Earth will continue forever, wind energy will also be available forever.

Question 2:

Give the names of two energy sources that you would consider to be exhaustible. Give reasons for your choices.

Answer:

Two exhaustible energy sources are as follows:

(a) Coal: It is produced from dead remains of plants and animals that remain buried under the earth's crust for millions of years. It takes millions of years to produce coal. Industrialization has increased the demand of coal. However, coal cannot replenish within a short period of time. Hence, it is a non-renewable or exhaustible source of energy.

(b) Wood: It is obtained from forests. Deforestation at a faster rate has caused a reduction in the number of forests on the Earth. It takes hundreds of years to grow a forest. If deforestation is continued at this rate, then there would be no wood left on the Earth. Hence, wood is an exhaustible source of energy.

Question 1:

A solar water heater cannot be used to get hot water on

- (a) a sunny day (b) a cloudy day
(c) a hot day (d) a windy day

Answer:

(b) A solar water heater uses solar energy to heat water. It requires bright and intense sunlight to function properly. On a cloudy day, the sunlight reflects back in the sky from the clouds and is unable to reach the ground. Therefore, solar energy is not available for the solar heater to work properly. Hence, solar water heater does not function on a cloudy day.

Question 2:

Which of the following is not an example of a bio-mass energy source?

- (a) wood (b) gobar gas
(c) nuclear energy (d) coal

Answer:

(c) Bio-mass is a source of energy that is obtained from plant materials and animal wastes. Nuclear energy is released during nuclear fission and fusion. In nuclear fission, uranium atom is bombarded with low-energy neutrons. Hence, uranium atom splits into two relatively lighter nuclei. This reaction produces huge amount of energy. In nuclear fusion reaction, lighter nuclei are fused together to form a relatively heavier nuclei. This reaction produces tremendous amount of energy. Hence, nuclear energy is not an example of bio-mass energy source.

Wood is a plant material, gobar gas is formed from animal dung, and coal is a fossil fuel obtained from the buried remains of plants and animals. Hence, these are bio-mass products.

Question 3:

Most of the sources of energy we use represent stored solar energy. Which of the following is not ultimately derived from the Sun's energy?

- (a) Geothermal energy
(b) Wind energy
(c) Nuclear energy
(d) Bio-mass

Answer:

(c) Nuclear energy is released during nuclear fission and fusion. In nuclear fission, uranium atom is bombarded with low-energy neutrons. Hence, uranium atom splits into two relatively lighter nuclei. This reaction produces huge amount of energy. In nuclear fusion reaction, lighter nuclei are fused together to form a relatively heavier nuclei. The energy required to fuse the lighter nuclei is provided by fission reactions. This reaction produces tremendous amount of energy. These reactions can be carried out in the absence or presence of sunlight. There is no effect of sunlight on these reactions. Hence, nuclear energy is not ultimately derived from Sun's energy.

Geothermal energy, wind energy, and bio-mass are all ultimately derived from solar energy.

Geothermal energy is stored deep inside the earth's crust in the form of heat energy.

The heating is caused by the absorption of atmospheric and oceanic heat. It is the sunlight that heats the atmosphere and oceans. Wind energy is harnessed from the blowing of winds. The uneven heating of the earth's surface by the Sun causes wind.

Bio-mass is derived from dead plants and animal wastes. Chemical changes occur in these dead plants

and animal wastes in the presence of water and sunlight. Hence, bio- mass is indirectly related to sunlight.

Question 4:

Compare and contrast fossil fuels and the Sun as direct sources of energy.

Answer:

Fossil fuels are energy sources, such as coal and petroleum, obtained from underneath the Earth's crust. They are directly available to human beings for use. Hence, fossil fuels are the direct source of energy. These are limited in amount. These are non-renewable sources of energy because these cannot be replenished in nature. Fossil fuels take millions of years for their formation. If the present fossil fuel of the Earth gets exhausted, its formation will take several years. Fossil fuels are also very costly. On the other hand, solar energy is a renewable and direct source of energy. The Sun has been shining for several years and will do so for the next five billion years. Solar energy is available free of cost to all in unlimited amount. It replenishes in the Sun itself.

Question 5:

Compare and contrast bio-mass and hydro electricity as sources of energy.

Answer:

Bio-mass and hydro-electricity both are renewable sources of energy. Bio-mass is derived from dead plants and animal wastes. Hence, it is naturally replenished. It is the result of natural processes. Wood, gobar gas, etc. are some of the examples of bio- mass. Hydro-electricity, on the other hand, is obtained from the potential energy stored in water at a height. Energy from it can be produced again and again. It is harnessed from water and obtained from mechanical processes.

Question 6:

What are the limitations of extracting energy from –
(a) the wind? (b) waves? (c) tides?

Answer:

(a) Wind energy is harnessed by windmills. One of the limitations of extracting energy from wind is that a windmill requires wind of speed more than 15 km/h to generate electricity. Also, a large number of windmills are required, which covers a huge area.
(b) Very strong ocean waves are required in order to extract energy from waves.
(c) Very high tides are required in order to extract energy from tides. Also, occurrence of tides depends on the relative positions of the Sun, moon, and the Earth.

Question 7:

On what basis would you classify energy sources as

(a) renewable and non-renewable?

(b) exhaustible and inexhaustible?

Are the options given in (a) and (b) the same?

Answer:

(a) The source of energy that replenishes in nature is known as renewable source of energy. Sun, wind, moving water, bio-mass, etc. are some of the examples of renewable sources of energy. The source of energy that does not replenish in nature is known as non-renewable source of energy. Coal, petroleum, natural gas, etc. are some of the examples of non- renewable sources of energy.
(b) Exhaustible sources are those sources of energy, which will deplete and exhaust after a few hundred years. Coal, petroleum, etc. are the exhaustible sources of energy.
Inexhaustible resources of energy are those sources, which will not exhaust in future.

These are unlimited. Bio-mass is one of the inexhaustible sources of energy. Yes. The options given in (a) and (b) are the same.

Question 8:

What are the qualities of an ideal source of energy?

Answer:

An ideal source of energy must be:

I. Economical

II. Easily accessible

III. Smoke/pollution free

IV. Easy to store and transport

V. Able to produce huge amount of heat and energy on burning

Question 9:

What are the advantages and disadvantages of using a solar cooker? Are there places where solar cookers would have limited utility?

Answer:

Solar cooker uses Sun's energy to heat and cook food. It is inexhaustible and clean renewable source of energy. It is free for all and available in unlimited amount. Hence, operating a solar cooker is not expensive.

Disadvantage of a solar cooker is that it is very expensive. It does not work without sunlight. Hence, on cloudy day, it becomes useless.

The places where the days are too short or places with cloud covers round the year, have limited utility for solar cooker.

Question 10:

What are the environmental consequences of the increasing demand for energy? What steps would you suggest to reduce energy consumption?

Answer:

Industrialization increases the demand for energy. Fossil fuels are easily accessible sources of energy that fulfil this demand. The increased use of fossil fuels has a harsh effect on the environment. Too much exploitation of fossil fuels increases the level of green house gas content in the atmosphere, resulting in global warming and a rise in the sea level.

It is not possible to completely reduce the consumption of fossil fuels. However, some measures can be taken such as using electrical appliances wisely and not wasting electricity. Unnecessary usage of water should be avoided. Public transport system with mass transit must be adopted on a large scale. These small steps may help in reducing the consumption of natural resources and conserving them.

NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 15 Our Environment>

Download 2017 2018 New Edition PDF

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 15 Our Environment>
Download 2017 2018 New Edition PDF

Class 10 Science Chapter 15 Our Environment NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1. Why are some substances biodegradable and some non-biodegradable ?

Answer: Substances that are broken down by biological processes are said to be biodegradable. In our environment, many of the substances are broken easily by decomposers (bacteria and fungi) as they possess specific enzymes for such activity. However, there are other substances also which are not broken down in this manner and are known as non-biodegradable substances. Since these substances are not degraded by bacteria and fungi, so they persist for a long time. These non-biodegradable substances will be acted upon by physical processes like heat and pressure.

Question 2. Give any two ways in which biodegradable substances would affect the environment.

Answer:

1. They may produce foul smell during decomposition process.
2. They may produce some harmful gases such as ammonia, methane, carbon dioxide, etc., which can further-cause global warming.

Question 3. Give any two ways in which non-biodegradable substances would affect the environment.

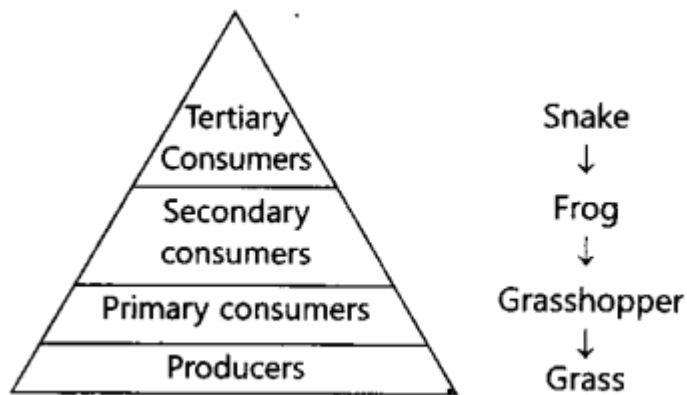
Answer:

1. These inert substances simply persist in the environment. This means that these substances require land area for dumping.
2. Excess of fertilizers, pesticides and other chemicals changes soil chemistry and also affects aquatic life.
3. Most of these chemicals and heavy metal are easily absorbed by the organisms. This causes biological magnification.

Page : 261

Question 1. What are trophic levels? Give an example of a food chain and state the different trophic levels in it.

Answer: Each step or level of the food chain forms a trophic level. Consider the following food chain:



Question 2. What is the role of decomposers in the ecosystem?

Answer: Role of decomposers in the ecosystem :

1. They help in breaking down the complex organic into simple inorganic that go into the soil and are used up by the plants.
2. They the nutrient pool of the putting In this way, ad as cleansing agents of nature.
3. They help in maintaining the fertility of by adding humus content to it.

Page : 264

Question 1. What is ozone and how does it affect any ecosystem?

Answer: Ozone (O_3) is a molecule formed by three atoms of oxygen. At the higher of the atmosphere, it shields the surface of the earth from ultraviolet (UV) radiation from the Sun. It may affect any ecosystem in the following ways :

1. At the surface of the earth, it is a deadly poison for all lower forms of life.
2. If this layer gets depleted, then it may cause cancer in human beings including other plants and animals.

Question 2. How can you help in reducing the problem of waste disposal? Give any two methods?

Answer:

1. By changing our lifestyle and change in attitude will reduce disposable waste.
2. Reducing packaging.
3. Recycling of waste.
4. Preparing compost of biodegradable waste.

Excercise:

Question 1. Which of the following groups contain only biodegradable items?

- (a) Grass, flowers and leather
- (b) Grass, wood and plastic
- (c) Fruit-peels, cake and lime-juice
- (d) Cake, wood and grass

Answer: (c) Fruit-peels, cake and lime-juice and (d) Cake, wood and grass

Question 2. Which of the following constitute a food-chain?

- (a) Grass, wheat and mango
- (b) Grass, goat and human
- (c) Goat, cow and elephant
- (d) Grass, fish and goat

Answer: (b) Grass, goat and human

Question 3. Which of the following are environment-friendly practices?

- (a) Carrying cloth-bags to put purchases in while shopping
- (b) Switching off unnecessary lights and fans
- (c) Walking to school instead of getting your mother to drop you on her scooter
- (d) All of the above

Answer: (d) All of the above

Question 4. What will happen if we kill all the organisms in one trophic level?

Answer: If we kill all the organisms in one trophic level, then transfer of energy as well as matter to next higher level will stop. It will lead to over- population at one particular level causing amongst the individuals. This would seriously disturb the food chain and can cause the collapse of an ecosystem even.

Question 5. Will the impact of removing all the organisms in a trophic level be different for different trophic levels? Can the organisms of any trophic level be removed without causing any damage to the ecosystem?

Answer: Yes, the impact Of removing all the Organisms in a trophic level will be different for different trophic levels. It will not be possible to remove any organism in any trophic level without causing damage to the ecosystem.

Question 6. What is biological magnification? Will the levels of this magnification be different at different levels of the ecosystem?

Answer: The accumulation of harmful chemicals in the body of living organisms at different trophic levels in a food chain is called biological magnification. Yes, the concentration of these harmful chemicals will be different at different trophic levels. It will be maximum at the last trophic levels which is mostly of the top carnivores (quaternary consumers).

Question 7. What are the problems caused by the non-biodegradable wastes that we generate?

Answer: (i) Non-biodegradable wastes persist in the environment for a long time and cause greater harm to the various members of the ecosystem by causing biological magnification.

(ii) Non-biodegradable waste such as fertilizers, pesticides, weedicides, etc., changes the soil chemistry. in turn affects the fertility of soil and subsequently reduces the crop yield.

Question 8. If all the waste we generate is biodegradable, will this have no impact on the environment?

Answer: Biodegradable waste will be recycled easily by the decomposers such as bacteria and fungi. It will have only this bad impact on our environment that, many Of the gases released during decomposition process may result in global warming.

Question 9. Why is damage to the ozone layer a cause for concern? What steps are being taken to limit this damage?

Answer: The ozone shields the surface of the earth from ultraviolet (UV) radiation from the sun. These radiations are highly damaging as they can cause cancer in both plants and animals, damage to eyes and immune system. They can also lead to variations in global rainfall, ecological disturbances and dwindling of global food supplies. Due to these reasons, damage to the ozone layer is a major cause for concern.

Steps which are taken to limit this damage :

1. To decrease the use of synthetic chemicals like chlorofluorocarbons (CFCs) which are used as refrigerants and in fire extinguishers.
2. In 1987, the United Nations Environment Programme (UNEP) succeeded in reaching an agreement to freeze CFC production at 1986



NCERT SOLUTIONS For Class 10 Science

<http://freehomedelivery.net/Chapter 16 Management of Natural Resources Download 2017 2018 New Edition PDF>

NCERT SOLUTIONS For Class 10 Science <http://freehomedelivery.net/Chapter 16 Management of Natural Resources Download 2017 2018 New Edition PDF>

Class 10 Science Chapter 16 Management of Natural Resources NCERT SOLUTIONS For Download 2017 2018 New Edition PDF

Question 1:

What changes can you make in your habits to become more environment-friendly?

Answer:

We should switch off the electrical appliances when not in use. Water and food should not be wasted. Close the tap when not in use. Dump the objects made of plastic and glass in designated recycling boxes. Plastic, paper, or glass must be recycled or reused and not dumped with other wastes. This is because objects made of plastic do not get decomposed easily. Besides soil fertility, they badly affect our environment. We should dispose the wastes safely and not disperse in public places. These are a few things that can be done to become more environment- friendly.

Question 2:

What would be the advantages of exploiting resources with short-term aims?

Answer:

There should be a judicious use of natural resources as they are limited in nature. We should not exploit resources for our short term gains as this would only lead to depletion of natural resources for the present generation as well as generations to come. Hence, we can say that there are hardly any advantages of exploiting natural resources for short term gains.

Question 3:

How would these advantages differ from the advantages of using a long-term perspective in managing our resources?

Answer:

In the case of a long-time perspective in managing our resources, these resources will last for the generations to come. This management ensures uniform distribution among the people. It conserves the natural resources for many years and not just for a few years, as in the case of a short-term perspective in conserving natural resources.

Question 4:

Why do you think there should be equitable distribution of resources? What forces would be working against an equitable distribution of our resources?

Answer:

Natural resources of the Earth must be distributed among the people uniformly so that each and every one gets his share of the resource.

Human greed, corruption, and the lobby of the rich and powerful are the forces working against an equitable distribution of resources.

Question 1:

Why should we conserve forests and wildlife?

Answer:

We should conserve forests and wildlife to preserve the biodiversity (range of different life-forms) so as to avoid the loss of ecological stability. A large number of tribes are the habitants in and around the forests. If the forests are not conserved, then it may affect these habitants. Without proper management of forest and wildlife, the quality of soil, the water sources, and even the amount of rainfall may be affected. Without forest and wildlife, life would become impossible for human beings.

Question 2:

Suggest some approaches towards the conservation of forests.

Answer:

Some approaches towards the conservation of forests are as follows:

- (a) People should show their participation in saving the forest by protesting against the cutting of trees. For example, Chipko Andolan
- (b) Planting of trees should be increased. Rate of afforestation must be more than that of deforestation.
- (c) Some people cut precious trees such as Chandan to earn money. Government should take legal steps to catch these wood smugglers.
- (d) Habitants of forests must not be bothered by the forest officials. Otherwise, this would result in the clash between tribal people and the government officials, thereby enhancing the naxal activities in forests.

Question 1:

Find out about the traditional systems of water harvesting/management in your region.

Answer:

One of the traditional systems of water harvesting used in our region is tanks.

Question 2:

Compare the above system with the probable systems in hilly/mountainous areas or plains or plateau regions.

Answer:

In plains, the water harvesting structures are crescent-shaped earthen embankments.

These are low, straight, and concrete.

In hilly regions, the system of canal irrigation called Kulhs is used for water harvesting.

This involves a collection of rain water in a stream, which is then diverted into manmade channels down the hill sides.

Question 3:

Find out the source of water in your region/locality. Is water from this source available to all people living in that area?

Answer:

The source of water in our region is ground water. Water from the source is available to all the people living in that area.

Question 1:

What changes would you suggest in your home in order to be environment-friendly?

Answer:

Changes that can be undertaken in our homes to be environment-friendly are listed below:

- (i) Switch off the electrical appliances when not in use.
- (ii) Turn the taps off while brushing or bathing and repair the leaking taps.
- (iii) Throw biodegradable and non-biodegradable waste into separate bins.

(iv) Construct composting pits.

(v) Food items such as jam, pickles, etc., come packed in plastic bottles. These bottles can later be used for storing things in the kitchen.

Question 2:

Can you suggest some changes in your school which would make it environmentfriendly?

Answer:

Changes that can be undertaken in our schools to make it environment friendly are listed below:

(i) Electricity can be saved by switching off lights and fans when not required.

(ii) Turn the taps off when not in use.

(iii) Biodegradable and non-biodegradable wastes should be thrown into separate bins.

Question 3:

We saw in this chapter that there are four main stakeholders when it comes to forests and wildlife. Which among these should have the authority to decide the management of forest produce? Why do you think so?

Answer:

The forest department of the government should have the authority to decide the management of forest produces. This is because the forest department is the care taker of the forest land and is responsible for any damage to the forest.

Question 4:

How can you as an individual contribute or make a difference to the management of (a) forests and wildlife, (b) water resources and (c) coal and petroleum?

Answer:

(a) Forest and wildlife:

(i) We should protest against the cutting of trees (deforestation).

(ii) We should protest against the poaching of wild animals.

(iii) We should stop the annexation of forest land for our use.

(b) Water resources:

(i) Turn the taps off while brushing or bathing and repair leaking taps.

(ii) We should practice rainwater harvesting.

(iii) We should avoid the discharge of sewage and other wastes into rivers and other water resources.

(c) Coal and petroleum:

(i) We should take a bus or practice car pooling to avoid excessive use of petroleum.

(ii) We should stop using coal as a fuel (angithis).

(iii) We should use alternative sources of energy such as hydro-energy and solar energy instead of depending largely on coal and petroleum.

Question 5:

What can you as an individual do to reduce your consumption of the various natural resources?

Answer:

Natural resources such as water, forests, coal and petroleum, etc. are important for the survival of human beings. The ways in which we can reduce the consumption of various natural resources are as follows:

(i) We should stop the cutting of trees (deforestation).

(ii) We should use recycled paper to reduce the cutting down of trees.

(iii) We should not waste water.

(iv) We should practice rainwater harvesting.

- (v) We should practice car pooling to avoid the excessive use of petroleum.
- (vi) We should use alternative sources of energy such as hydro-energy and solar energy.

Question 6:

List five things you have done over the last one week to –

- (a) conserve our natural resources
- (b) increase the pressure on our natural resources

Answer:

(a) To conserve our natural resources:

- (i) Travel by a CNG bus for long distances and walk for short distances.
- (ii) Use recycled paper
- (iii) Throw biodegradable and non-biodegradable waste into separate bins
- (iv) Plant trees
- (v) Harvest rainwater

(b) To increase the pressure on our natural resources:

- (i) Use non-renewable resources of energy
- (ii) Waste water
- (iii) Waste electricity
- (iv) Use plastics and polythene bags for carrying goods
- (v) Use escalators

Question 7:

On the basis of the issues raised in this chapter, what changes would you incorporate in your life-style in a move towards a sustainable use of our resources?

Answer:

One should incorporate the following changes in life-style in a move towards a sustainable use of our resources:

- (i) Stop cutting trees and practice plantation of trees.
- (ii) Stop using plastic and polythene bags for carrying goods.
- (iii) Use recycled paper.
- (iv) Throw biodegradable and non-biodegradable waste into separate bins.
- (v) Waste minimum amount of water while using and repair leaking taps.
- (vi) Practice rainwater harvesting.
- (vi) Avoid using vehicles for short distances. Instead, one can walk or cycle to cover short distances. To cover long distances, one should take a bus instead of using personal vehicles.
- (vii) Switch off electrical appliances when not in use.
- (viii) Use fluorescent tubes in place of bulbs to save electricity.
- (ix) Take stairs and avoid using lifts.
- (x) During winters, wear an extra sweater to avoid using heaters.